

NHSM Determination from Empire 5/7/2024

Wednesday, May 22, 2024 1:03 PM



Andrews, Edward S <edward.s.andrews@wv.gov>

Fwd: Empire Green Generation NHSM Self Determination

1 message

Andrews, Edward S <edward.s.andrews@wv.gov>

Tue, May 7, 2024 at 6:16 AM

To: Beverly D Mckeone <beverly.d.mckeone@wv.gov>, Marycate Opila <opila.marycate@epa.gov>, "Supplee, Gwendolyn" <supplee.gwendolyn@epa.gov>, ott.steven@epa.gov

Steven,

I am forwarding Empire's determination as your office requested. See forwarded email with attachment.

Ed

Edward Andrews, P.E.
Engineer
WVDEP/Division of Air Quality
304-926-0499 Ext 41244
601 57th Street, SE
Charleston, WV 25304

----- Forwarded message -----

From: **Wood, Katie** <katie.wood@tetrattech.com>
Date: Fri, May 3, 2024 at 3:09 PM
Subject: Empire Green Generation NHSM Self Determination
To: Edward Andrews <edward.s.andrews@wv.gov>
CC: Farley R Wood <fwood@empirede.com>

Ed,

Please find attached the Empire Green Generation NHSM self determination as we discussed.

Thanks,

Katie Wood* | Environmental Scientist
Direct +1 (740) 298-9062 | Mobile +1 (304) 559-9980 | katie.wood@tetrattech.com
Formerly Katie Pugh, please note name change
Tetra Tech | *Leading with Science®* | OGA
47443 National Rd Suite 3 | St. Clairesville, OH 43950 | tetrattech.com

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Empire Green Generation NHSM 5-3-24.pdf
394K

Follow-up From EPA 5/6/2024

Wednesday, May 22, 2024 12:58 PM



Andrews, Edward S <edward.s.andrews@wv.gov>

RE: Re: Regulatory Interpretation Request Empire Green Generation LLC

1 message

Ott, Steven <Ott.Steven@epa.gov>
To: "edward.s.andrews@wv.gov" <edward.s.andrews@wv.gov>

Mon, May 6, 2024 at 3:16 PM

Good afternoon,

I wanted to reach out and make sure this was received. Again, you have any questions please reach out at any time.

Thanks,

Steve

From: Ott, Steven
Sent: Tuesday, April 30, 2024 12:26 PM
To: edward.s.andrews@wv.gov
Cc: Hall, Kristen <hall.kristen@epa.gov>; Opila, MaryCate <Opila.MaryCate@epa.gov>; Supplee, Gwendolyn <Supplee.Gwendolyn@epa.gov>; Stankunas, Krystal <Stankunas.Krystal@epa.gov>; Adkins, Jesse D <jesse.d.adkins@wv.gov>; Beverly.d.mckeone@wv.gov
Subject: Re: Regulatory Interpretation Request Empire Green Generation LLC

Dear Mr. Andrews:

This is in response to your letter dated March 6th, 2023, requesting a regulatory interpretation (request) from the Administrator (EPA), regarding Empire Green Generation's (EGG) proposed modification to their Follansbee, WV facility. EGG proposed to cease processing medical waste and transition to processing plastics. To support the change of feedstock EGG proposed constructing a hydrochloric acid truck loading facility in conjunction with an appropriate scrubbing system. The proposed feedstock change would allow the facility to produce hydrochloric acid as an additional product of pyrolysis.

In the request the West Virginia Department of Environmental Protection, Division of Air Quality (WVDEP) posits multiple questions. EPA Region 3 has determined that a waste determination needs to be performed prior to further analysis (40 CFR 60.2175(v)).

40 CFR 258.2 defines solid waste to mean:

"...any garbage, or refuse, sludge from a wastewater treatment plant, water supply treatment plant, or air pollution control facility and other discarded material, including solid, liquid, semi-solid, or contained gaseous material resulting from industrial, commercial, mining, and agricultural operations, and from community activities, but does not include solid or dissolved materials in domestic sewage, or solid or dissolved materials in irrigation return flows or industrial discharges that are point sources subject to permit under 33 U.S.C. 1342..."

The permit modification application submitted to WVDEP, by EGG, on page 29 states that the plastic will be received and processed prior to being delivered to Empire Green Generation's Follansbee, WV facility. This may be covered under "... other discarded material..." of 40 CFR 258.2.

EGG needs to perform a waste analysis for their plastic feedstock, in accordance with the requirements of 40 CFR 241 to be fully evaluated under 40 CFR Part 60 Subpart CCCC. The material must be "Sufficiently processed" (40 CFR 241.3(b) (4)) and meet the following legitimacy criteria as described in 40 CFR 241.3(d)(1): The material is managed as a valuable commodity, the material has a meaningful heating value, and the material has levels of contamination comparable or less than a similar traditional fuel, which the [pyrolysis] unit is designed to burn.

A guide and flowchart for 40 CFR 241 is available at www.epa.gov/rcra/non-hazardous-secondary-material-nhsm-guide-wastenon-waste-determinations and is also attached.

EPA has additional questions needed to determine the applicability of 40 CFR Part 60 Subparts AAAA, CCCC, and EEEE however, the Waste Determination is the first step needed and once completed we request that WVDEP share the results of the determination with Region 3 so we can continue our review or your request.

Please feel free to reach out if you have any questions.

Best regards,

Steve Ott

	<p>Steve Ott Air Inspector ECAD US EPA Mid-Atlantic Region Phone 215-814-2267 Email ott.steven@epa.gov</p> <p> </p>
---	--

NHSM Determination 5/3/2024

Wednesday, May 22, 2024 12:59 PM



Andrews, Edward S <edward.s.andrews@wv.gov>

Empire Green Generation NHSM Self Determination

1 message

Wood, Katie <katie.wood@tetrattech.com>
To: Edward Andrews <edward.s.andrews@wv.gov>
Cc: Farley R Wood <fwood@empirede.com>

Fri, May 3, 2024 at 3:09 PM

Ed,


Please find attached the Empire Green Generation NHSM self determination as we discussed.

Thanks,

Katie Wood* | Environmental Scientist
Direct **+1 (740) 298-9062** | Mobile **+1 (304) 559-9980** | katie.wood@tetrattech.com
Formerly Katie Pugh, please note name change
Tetra Tech | *Leading with Science®* | OGA
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 **Empire Green Generation NHSM 5-3-24.pdf**
394K



Empire
Green Ge...

Empire Green Generation Non-Hazardous Secondary Material Self-Determination

Empire Green Generation is self-determining that the feedstock material for their proposed pyrolysis plastic processing facility would be exempt from treatment as a solid waste. This is based on the Non-Hazardous Secondary Material Rule (NHSM) and the feedstock is not a solid waste, in accordance with 40 CFR 241.3(b). The site-specific self-determination requirements under 40 CFR 241.3(b) are:

- The facility generating or combusting an NHSM determines if they will make a waste or nonwaste determination for an NHSM (1) used as fuel managed within the control the generator, (2) used as an ingredient, or (3) used as a fuel or an ingredient produced from processed discarded NHSM.
- The NHSM must meet the legitimacy criteria of 40 CFR 241.3(d).
- See the flow chart and additional information below (Figure 1).



Figure 1

Empire Green Generation Non-Hazardous Secondary Material Self-Determination

Question 1 – Is the material a traditional fuel or clean cellululosic biomass?

Answer 1 – No, the feedstock is not considered a traditional fuel or clean cellululosic biomass as defined in 40 CFR Part 241.2.

Question 2: Is the material a categorical non-waste?

Answer 2: No, the feedstock is not considered a categorical non-waste as defined in 40 CFR Part 241.4.

Question 3: Is the material managed within the control of the generator?

Answer 3: Yes, 40 CFR Part 241.2 defines "within control of the generator" as meaning that "the non-hazardous secondary material is generated and burned in combustion units at the generating facility; or that such material is generated and burned in combustion units at different facilities, provided the facility combusting the non-hazardous secondary material is controlled by the generator; or both the generating facility and the facility combusting the non-hazardous secondary material are under the control of the same person as defined in this section." While Empire Green Generation's NHSM is not burning or combusting the plastics, the generating facility and the facility (pyrolysis) the non-hazardous secondary material are under the control of the same person as defined in this section.

Question 4: Does the material satisfy all three legitimacy criteria?

Answer 4: Yes, the feedstock does meet the legitimacy criteria, as specified in 40 CFR Part 241.3(d)(2), as follows:

1. The NHSM must be managed as a valuable commodity based on the following factors:
 - a. The storage of the non-hazardous secondary material prior to use must not exceed reasonable time frames: This is met at the facility call's for a maximum storage time of 7 days.
 - b. Where there is an analogous fuel, the non-hazardous secondary material must be managed in a manner consistent with the analogous fuel or otherwise be adequately contained to prevent releases to the environment: The plastic feedstock is comprised of synthetic polymers made up of repeating hydrocarbon molecules. The pyrolysis process breaks down the complex molecules into its component hydrocarbon molecules to create syngas. Polyvinyl chloride (PVC) plastic can be processed which will liberate chlorine molecules that are collected and processed into saleable hydrochloric acid.
2. The non-hazardous secondary material must provide a useful contribution to the production or manufacturing process. The non-hazardous secondary material provides a useful contribution if it contributes a valuable ingredient to the product or intermediate or is an effective substitute for a commercial product. The feedstock material is the basis for the synthesis gas (syngas) that is created in the pyrolysis unit. The feedstock material is also the basis for making Hydrochloric Acid which will be sold.
3. The non-hazardous secondary material must be used to produce a valuable product or intermediate. The product or intermediate is valuable if:
 - a. The non-hazardous secondary material is sold to a third party. The NHSM is used by the Generator as an ingredient to produce syngas and hydrochloric acid that is sold to a 3rd party, and generates electricity for self-consumption.
 - b. Or the non-hazardous secondary material is used as an effective substitute for a commercial product or as an ingredient or intermediate in an industrial

Empire Green Generation Non-Hazardous Secondary Material Self-Determination

process. The NHSM is used as a substitute for crude oil in the production of syngas which is used to run the unit.

4. The non-hazardous secondary material must result in products that contain contaminant at levels that are comparable in concentration to or lower than those found in traditional products that are manufactured without the non-hazardous secondary material. Empire Green Generation has an agreement with the end buyer to produce the hydrochloric acid to the industry standards with a threshold of percent weight of hydrochloric at a minimum of 30% and a maximum of 32%, Iron ppm maximum of 7, organics at a non-detect, Mercury at non-detect, Specific Gravity of minimum of 1.1525 maximum 1.1628 and a pH less than 7. The syn-gas produced, which will be used to run the plant is derived from the same base material as the traditional fuel. This product comparable or lower than traditional fuel for contaminants.

The Green Generation process does not combust the feedstock material, but rather uses pyrolysis to process the material in a non-combustible environment, it meets the legitimacy criteria specified in 40 CFR 241 to be classified as a NHSM based on the answers listed above.

Request for NHSM Determination 4/30/2024

Wednesday, May 22, 2024 1:02 PM



Andrews, Edward S <edward.s.andrews@wv.gov>

Fwd: Regulatory Interpretation Request Empire Green Generation LLC

1 message

Andrews, Edward S <edward.s.andrews@wv.gov>

Tue, Apr 30, 2024 at 12:46 PM

To: "Wood, Katie" <katie.wood@tetrattech.com>, Farley Wood <fwood@empirede.com>, Bernard Brown <bbrown@empirede.com>

In our RIR to R3, EPA found that EGG needs to conduct a waste determination needs to be completed before the Administrator can respond to the DAQ's RIR.

Please provide waste/non-waste determination of the proposed plastic feedstock for EGG's pyrolysis units at the Folloasabee Facility by no later than May 15, 2024.

Should you have any questions about this request, please contact me.

Thanks,

Ed

--

Edward Andrews, P.E.
Engineer
WVDEP/Division of Air Quality
304-926-0499 Ext 41244
601 57th Street, SE
Charleston, WV 25304

----- Forwarded message -----

From: **Ott, Steven** <Ott.Steven@epa.gov>

Date: Tue, Apr 30, 2024 at 12:25 PM

Subject: Re: Regulatory Interpretation Request Empire Green Generation LLC

To: edward.s.andrews@wv.gov <edward.s.andrews@wv.gov>

Cc: Hall, Kristen <hall.kristen@epa.gov>, Opila, MaryCate <Opila.MaryCate@epa.gov>, Supplee, Gwendolyn <Supplee.Gwendolyn@epa.gov>, Stankunas, Krystal <Stankunas.Krystal@epa.gov>, Adkins, Jesse D <jesse.d.adkins@wv.gov>, beverly.d.mckeone <beverly.d.mckeone@wv.gov>

Dear Mr. Andrews:

This is in response to your letter dated March 6th, 2023, requesting a regulatory interpretation (request) from the Administrator (EPA), regarding Empire Green Generation's (EGG) proposed modification to their Follansbee, WV facility. EGG proposed to cease processing medical waste and transition to processing plastics. To support the change of feedstock EGG proposed constructing a hydrochloric acid truck loading facility in conjunction with an appropriate scrubbing system. The proposed feedstock change would allow the facility to produce hydrochloric acid as an additional product of pyrolysis.

In the request the West Virginia Department of Environmental Protection, Division of Air Quality (WVDEP) posits multiple questions. EPA Region 3 has determined that a waste determination needs to be performed prior to further analysis (40 CFR 60.2175(v)).

40 CFR 258.2 defines solid waste to mean:

"...any garbage, or refuse, sludge from a wastewater treatment plant, water supply treatment plant, or air pollution control facility and other discarded material, including solid, liquid, semi-solid, or contained gaseous material resulting from industrial, commercial, mining, and agricultural operations, and from community activities, but does not include solid or dissolved materials in domestic sewage, or solid or dissolved materials in irrigation return flows or industrial discharges that are point sources subject to permit under 33 U.S.C. 1342..."

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Please feel free to reach out if you have any questions.

Best regards,

Steve Ott

	<p>Steve Ott Air Inspector ECAD US EPA Mid-Atlantic Region Phone 215-814-2267 Email ott.steven@epa.gov</p> <p> </p>
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 **NHSM Guide for Waste Non-Waste Determinations.pdf**
686K

EPA Request for waste/non-determination 4/30/2024

Wednesday, May 22, 2024 12:57 PM



Andrews, Edward S <edward.s.andrews@wv.gov>

Re: Regulatory Interpretation Request Empire Green Generation LLC

1 message

Ott, Steven <Ott.Steven@epa.gov>

Tue, Apr 30, 2024 at 12:25 PM

To: "edward.s.andrews@wv.gov" <edward.s.andrews@wv.gov>

Cc: "Hall, Kristen" <hall.kristen@epa.gov>, "Opila, MaryCate" <Opila.MaryCate@epa.gov>, "Supplee, Gwendolyn" <Supplee.Gwendolyn@epa.gov>, "Stankunas, Krystal" <Stankunas.Krystal@epa.gov>, "Adkins, Jesse D" <jesse.d.adkins@wv.gov>, "beverly.d.mckeone" <beverly.d.mckeone@wv.gov>

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Please feel free to reach out if you have any questions.

Best regards,

Steve Ott

	<p>Steve Ott Air Inspector ECAD US EPA Mid-Atlantic Region Phone 215-814-2267 Email ott.steven@epa.gov</p> <p> </p>
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 **NHSM Guide for Waste Non-Waste Determinations.pdf**
686K

Checking In 3/21/2024

Wednesday, March 27, 2024 2:09 PM



Andrews, Edward S <edward.s.andrews@wv.gov>

Checking In

1 message

Wood, Katie <katie.wood@tetrattech.com>
To: Edward Andrews <edward.s.andrews@wv.gov>

Thu, Mar 21, 2024 at 1:23 PM

Hey Ed,

Just wanted to check in and see if you needed anything further for Empire.

Thanks,

Katie

Katie Wood* | Environmental Scientist
Direct **+1 (740) 298-9062** | Mobile **+1 (304) 559-9980** | katie.wood@tetrattech.com
Formerly Katie Pugh, please note name change
Tetra Tech | *Leading with Science®* | OGA
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Further Discussion 3/11/2024

Wednesday, March 27, 2024 2:08 PM



Andrews, Edward S <edward.s.andrews@wv.gov>

Re: Empire Discussion

1 message

Andrews, Edward S <edward.s.andrews@wv.gov>

Mon, Mar 11, 2024 at 1:26 PM

To: "Farley R. Wood, P.E." <fwood@empirede.com>

Cc: "Wood, Katie" <katie.wood@tetrattech.com>, Malcolm Kingston <design@technotherm.co.za>

Thanks for the follow up with my concerns about the process.

I will provide either a my understanding of the decay of PVC and formation of HCl with recovery or an illustration/flow diagram with which we can have a discussion about.

Thanks
Ed

On Mon, Mar 11, 2024 at 1:15 PM Farley R. Wood, P.E. <fwood@empirede.com> wrote:

Ed,

See my comments below:



Farley R. Wood, P.E.
Vice President of Engineering


Main Office (304) 935-5851
Mobile: (304) 650-2023
Teams: [Click Here](#)

fwood@empirede.com
www.empirediversifiedenergy.com

From: Wood, Katie <katie.wood@tetrattech.com>
Sent: Friday, March 8, 2024 3:15 PM
To: Farley R. Wood, P.E. <fwood@empirede.com>
Subject: FW: Empire Discussion

You don't often get email from katie.wood@tetrattech.com. [Learn why this is important](#)

From: Andrews, Edward S <edward.s.andrews@wv.gov>
Sent: Friday, March 8, 2024 3:10 PM
To: Wood, Katie <katie.wood@tetrattech.com>
Subject: Re: Empire Discussion

 **CAUTION:** This email originated from an external sender. Verify the source before opening links or attachments.



Katie,

I have looked over the latest request to my questions.

Some things are making sense(e.g, formation of HCl from the pyrolysis unit).

However, I am still a little confused on how the gaseous HCl in the synthetic gas is going to be stripped out with the gas cleaning trains without a significant amount of water being entrained into the cleaned synthetic gas stream.

The Cl gas will be liberated in the pre-pyrolysis unit and removed before the feed enters the pyrolysis unit, where the syngas is produced.

I saw that there was some HCl estimated from the scrubber in the revised application but I did not receive any attachments in your February 21 email (tanks/HCl calculations).

Katie will forward to you.

Here are some of my concerns/questions that I still have:

A)1) Make up the PVC in the plastic feedstock. At 100% PVC, about 15.5 gallons per minute of 20% HCl could potentially be produced. At this rate, the storage capacity would be exceeded in one day.

We cannot run 100% PVC feed to the plant. We are limited to 15%, and should produce a maximum of 5,500 gallons per day of HCl.

We also have the option of running less than 15% to reduce HCl production, or not run any PVC to effectively stop HCl production.

A)2) Operating temperature range of the pyrolysis unit.

The pre-pyrolysis units operating temperature is 320° C (608° F). The operating temperature of the pyrolysis unit is 830° C (1,526° F).

A)3) Will oils/tars still be produced? One of the references in the provided paper noted the formation of methane, ethane, ethyne, 1-butane, hydrogen, chlorine, and benzene, which are in the form of C_nH_m. None of these compounds

are close to C14

Oils/tars will be produced. We have an oil/water separator on the HCl circuit, and a larger oil/water separator on each of the three gas cleanup trains.

The collected oil/tars are pumped to a central collection tank where the product is used as fuel in the combustion chamber of the pyrolysis units.

B) I am assuming the gas cleaning trains are going to be used to extract the gaseous HCl out of the synthetic gas. How does EGG plan on doing this and what properties are going to be monitored?

I am envisioning that the scrubber would be blown down once the circulating water reached a HCl conc of 20% or sg of 1.1. This needs to be spelled out.

The Cl gas will be collected and removed from the system prior to the generation of syngas.

C) Venting the HCl scrubber to the RTO might be the simplest approach. Operating pressures and water/HCl carryover might adversely affect the life and performance of the RTO. Please see Note 3 on DWG Poly Scrub Basis 5'x5' scrubber that the vessel is designed for 1.9 SpG Mat'l @ 100 F/Atmos Pressure.

The HCl fumes will pass through the scrubber which is 99% efficient. There will not be any water carryover.

There will be high dilution of the very low concentration HCl fumes returning to the RTO that the designers of the system feel any adverse effects to life cycle or performance would be negligible.

Stipping gaseous HCl with water is going to generate a significant amount of heat energy.

The HCl system has two chiller units included in the circuit. One for the gasses entering the spray tower, and a second for the liquid leaving the tower.

Plus connecting the vent line to the RTO is going to reduce the operating pressure below atmospheric (negative pressure).

That is correct. We want negative pressure from the HCl scrubber to the RTO to prevent leakage.

D) Classification of the plastic feedstock is either a fuel/raw ingredient in accordance with 40 CFR 241.

Good

On Fri, Mar 1, 2024 at 10:23 AM Wood, Katie <katie.wood@tetrattech.com> wrote:

Ed,

We have some answers to your questions below in red. Please let me know if you would like to discuss.

Thanks,

Katie

From: Andrews, Edward S <edward.s.andrews@wv.gov>

Sent: Friday, February 23, 2024 2:24 PM

To: Wood, Katie <katie.wood@tetrattech.com>; Farley Wood <fwood@empirede.com>

Subject: Re: Empire Discussion

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Using Chlorine gas and water reaction to produce HCl also produces a by-product of HOCl (hypochlorous acid). <https://www.bing.com/search?q=chlorine+gas+water+reaction&qs=UT&pq=chlorine+gas+water+reaction&sc=10-27&cvid=1D306673308642F9BCEE5D950B9BFB08&FORM=QBRE&sp=1&ghc=1&lq=0> The process does not produce any Chlorine gas, it produces Chloride gas so no HOCL is produced. I have attached a paper provided by Technotherm for futher information

Question is: Will Empire separate hypochlorous acid from the HCl or send it out as it? As per above no HOCL will be produced

Will the emissions of HCl go through a separate release point than the RTO stack? If it is a separate stack, I need the stack id and stack parameters of this point. No all emissions will go through the RTO

Also, I will need the calculations to support your emission estimate of HCl and concentration of HCl in the effluent release to the atmosphere from the production/storage/loading out of HCl. Calculations attached, these were included in the last submittal

Ed

On Thu, Feb 22, 2024 at 1:00 PM Wood, Katie <katie.wood@tetrattech.com> wrote:

Microsoft Teams meeting

Join on your computer, mobile app or room device

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Passcode: Gbfah4

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+1 213-357-2812,,114016353# United States, Los Angeles

Phone Conference ID: 114 016 353#

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--

Edward Andrews, P.E.

Engineer

WVDEP/Division of Air Quality

304-926-0499 Ext 41244

601 57th Street, SE

Charleston, WV 25304

--

Edward Andrews, P.E.

Engineer

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304-926-0499 Ext 41244

601 57th Street, SE

Charleston, WV 25304

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Edward Andrews, P.E.
Engineer
WVDEP/Division of Air Quality
304-926-0499 Ext 41244
601 57th Street, SE
Charleston, WV 25304



Andrews, Edward S <edward.s.andrews@wv.gov>

RE: Empire Discussion

1 message

Wood, Katie <katie.wood@tetratech.com>
To: "Andrews, Edward S" <edward.s.andrews@wv.gov>
Cc: Farley R Wood <fwood@empirede.com>

Mon, Mar 11, 2024 at 1:09 PM

Ed,

Attached is the emission control sheet for the scrubber. Farley is pulling together answers for you questions below and will be sending over shortly.

Thanks,

Katie

From: Andrews, Edward S <edward.s.andrews@wv.gov>
Sent: Friday, March 8, 2024 3:10 PM
To: Wood, Katie <katie.wood@tetratech.com>
Subject: Re: Empire Discussion

 **CAUTION:** This email originated from an external sender. Verify the source before opening links or attachments. 

Katie,

I have looked over the latest request to my questions.

Some things are making sense(e.g, formation of HCl from the pyrolysis unit).

However, I am still a little confused on how the gaseous HCl in the synthetic gas is going to be stripped out with the gas cleaning trains without a significant amount of water being entrained into the cleaned synthetic gas stream.

I saw that there was some HCl estimated from the scrubber in the revised application but I did not receive any attachments in your February 21 email (tanks/HCl calculations).

Here are some of my concerns/questions that I still have:

A)1) Make up the PVC in the plastic feedstock. At 100% PVC, about 15.5 gallons per minute of 20% HCl could potentially be produced. At this rate, the storage capacity would be exceeded in one day.

A)2) Operating temperature range of the pyrolysis unit.

A)3) Will oils/tars still be produced? One of the references in the provided paper noted the formation of methane, ethane, ethyne, 1-butane, hydrogen, chlorine, and benzene, which are in the form of C_nH_m. None of these compounds are close to C₁₄

B) I am assuming the gas cleaning trains are going to be used to extract the gaseous HCl out of the synthetic gas. How does EGG plan on doing this and what properties are going to be monitored?

I am envisioning that the scrubber would be blown down once the circulating water reached a HCl conc of 20% or sg of 1.1. This needs to be spelled out.

C) Venting the HCl scrubber to the RTO might be the simplest approach. Operating pressures and water/HCl carryover might adversely affect the life and performance of the RTO. Please see Note 3 on DWG Poly Scrub Basis 5'x5' scrubber that the vessel is designed for 1.9 SpG Mat'l @ 100 F/Atmos Pressure.

Stripping gaseous HCl with water is going to generate a significant amount of heat energy.

Plus connecting the vent line to the RTO is going to reduce the operating pressure below atmospheric (negative pressure).

D) Classification of the plastic feedstock is either a fuel/raw ingredient in accordance with 40 CFR 241.

On Fri, Mar 1, 2024 at 10:23 AM Wood, Katie <katie.wood@tetratech.com> wrote:

Ed,

We have some answers to your questions below in red. Please let me know if you would like to discuss.

Thanks,

Katie

From: Andrews, Edward S <edward.s.andrews@wv.gov>
Sent: Friday, February 23, 2024 2:24 PM
To: Wood, Katie <katie.wood@tetratech.com>; Farley Wood <fwood@empirede.com>
Subject: Re: Empire Discussion

 **CAUTION:** This email originated from an external sender. Verify the source before opening links or attachments.



Using Chlorine gas and water reaction to produce HCl also produces a by-product of HOCl (hypochlorous acid). <https://www.bing.com/search?q=chlorine+gas+water+reaction&q=UT&pq=chlorine+gas+water+reaction&sc=10-27&cvid=1D306673308642F9BCEE5D950B9BFB08&FORM=QBRE&sp=1&ghc=1&lq=0> The process does not produce any Chlorine gas, it produces Chloride gas so no HOCL is produced. I have attached a paper provided by Technotherm for further information

Question is: Will Empire separate hypochlorous acid from the HCl or send it out as it? As per above no HOCL will be produced

Will the emissions of HCl go through a separate release point than the RTO stack? If it is a separate stack, I need the stack id and stack parameters of this point. No all emissions will go through the RTO

Also, I will need the calculations to support your emission estimate of HCl and concentration of HCl in the effluent release to the atmosphere from the production/storage/loading out of HCl. Calculations attached, these were included in the last submittal

Ed

On Thu, Feb 22, 2024 at 1:00 PM Wood, Katie <katie.wood@tetratech.com> wrote:

Microsoft Teams meeting

Join on your computer, mobile app or room device

[Click here to join the meeting](#)

Meeting ID: 248 808 531 032

Passcode: Gbfah4

[Download Teams](#) | [Join on the web](#)

Or call in (audio only)

+1 213-357-2812,,114016353# United States, Los Angeles

Phone Conference ID: 114 016 353#

[Find a local number](#) | [Reset PIN](#)

[Learn More](#) | [Meeting options](#)

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Edward Andrews, P.E.

Engineer

WVDEP/Division of Air Quality

304-926-0499 Ext 41244

[601 57th Street, SE](#)

[Charleston, WV 25304](#)

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Edward Andrews, P.E.


Engineer

WVDEP/Division of Air Quality

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[601 57th Street, SE](#)

[Charleston, WV 25304](#)

 **M-3.pdf**
2632K



M-3

Attachment M
Air Pollution Control Device Sheet
(WET COLLECTING SYSTEM-SCRUBBER)

Control Device ID No. (must match Emission Units Table):

Equipment Information

1. Manufacturer: Poly Processing Model No. 3' X 5' Scrubber	2. Method: <input type="checkbox"/> Packed Bed <input type="checkbox"/> Spray Tower <input type="checkbox"/> Mechanical <input checked="" type="checkbox"/> Other, specify	<input type="checkbox"/> Venturi <input type="checkbox"/> Cyclone <input type="checkbox"/> Orifice
3. Provide diagram(s) of unit describing capture system with duct arrangement and size of duct, air volume, capacity, horsepower of motors. If applicable, state hood face velocity and hood collection efficiency.		
4. Provide a scale diagram of the scrubber showing internal construction. Please include packing type and size, spray configurations, baffle plates, and mist eliminators.		
5. What type of liquid entrainment eliminators or system will be used? Submit a schematic diagram showing thickness, mesh, and material of construction.		
6. Describe the scrubber's construction material: Polyethylene (XLPE tank with PVC internals and exterior piping. Water with NaOH is added to the vessel above the set level. Vent gas from the HCl tank passes through the gas diffuser where HCl fumes are neutralized before the scrubbed gas is vented from the top.		
7. What will be the power requirements of the collector? Fan: NO HP Inlet scrubbing liquid pump: NA HP		
8. What type of fan(s) will be used? Type of fan blade: None Number of blades: None Diameter of blade: None in. Also supply a fan curve for each fan to be used.		
9. Estimated gas pressure drop at maximum flow rate: 1 inches H ₂ O		

Scrubbing Liquor Characteristics

10. Scrubbing Liquor	11. Scrubbing liquor losses (evaporation, etc.): 0.5 gal/1000 ACF gas										
<table border="1"> <thead> <tr> <th>Composition</th> <th>Weight %</th> </tr> </thead> <tbody> <tr> <td>1 Water</td> <td>99</td> </tr> <tr> <td>2 NaOH</td> <td>1</td> </tr> <tr> <td>3</td> <td></td> </tr> <tr> <td>4</td> <td></td> </tr> </tbody> </table>	Composition	Weight %	1 Water	99	2 NaOH	1	3		4		12. Liquor pressure to scrubber: 0.25 PSIA
Composition	Weight %										
1 Water	99										
2 NaOH	1										
3											
4											
	13. Pressure drop through scrubber: 6 in. H ₂ O										
14. Source of liquor (explain): Batch liquid added to tank	15. Liquor flow rates to scrubber: Design maximum: 1,000 gal/min Average expected: 500 gal/min										
16. Describe system to be used to supply liquor to collector: Manual drain and re-fill process											
17. Give the expected solids content of the liquor: No suspended solids will be created. The system will convert HCl gas vapors will react with NaOH to form water (H ₂ O) and solid (NaCl). Water will be changed out well before NaCl concentration reaches saturation.											

18. If the liquor is to be recirculated, describe any treatment performed:	
19. Data for Venturi Scrubber: Throat Dimensions: NA (Specify Units) Throat Velocity: NA ft/sec	20. Data for Packed Towers: Type of Packing: NA Superficial Gas Velocity through Bed:

Gas Stream Characteristics

21. Gas flow into the collector: 1000 ACF @ 20 °F and 14.2 PSIA	22. Gas stream temperature: Inlet: ambient °F Outlet: ambient °F
23. Gas flow rate: Design Maximum: 135 ACFM Average Expected: 67 ACFM	24. Particulate Grain Loading in grains/scf: Inlet: NA Outlet: NA

25. Emission rate of each pollutant (specify) into and out of collector:					
Pollutant	IN		OUT		Guaranteed Minimum Collection Efficiency
	lb/hr	grains/scf	lb/hr	grains/scf	
A HCl	0.0015/hr		0.00015/hr		99
B					
C					
D					
E					

26. Type of pollutant(s) controlled: <input type="checkbox"/> SO ₂ <input type="checkbox"/> Odor <input type="checkbox"/> Particulate (type): <input checked="" type="checkbox"/> Other: HCl
27. By what method were the uncontrolled emissions calculated? <input checked="" type="checkbox"/> Material Balance <input type="checkbox"/> Stack Test <input type="checkbox"/> Pilot Test <input type="checkbox"/> Other:
28. Dimensions of stack: Height: 4'-9" ft Diameter: 5'-1" ft
29. Supply an equilibrium curve and/or solubility data (at various temperatures) for the proposed system.
30. Supply a curve showing proposed collection efficiency versus gas volume from 25 to 100 percent of design rating of collector.

Particulate Distribution

31. Complete the table:	Particle Size Distribution at Inlet to Collector		Fraction Efficiency of Collector
	Particulate Size Range (microns)	Weight % for Size Range	Weight % for Size Range
	0 – 2	NA, vapor to liquid solvent conversion.	
	2 – 4		
	4 – 6		
	6 – 8		
	8 – 10		
	10 – 12		
	12 – 16		
	16 – 20		
	20 – 30		
	30 – 40		
	40 – 50		
	50 – 60		
	60 – 70		
	70 – 80		
	80 – 90		
	90 – 100		
	>100		
32. Describe any air pollution control device inlet and outlet gas conditioning processes (e.g., gas cooling, gas reheating, gas humidification):			
33. Describe the collection material disposal system: Neutralized HCl gas will become a salt brine that will be disposed of in accordance with local, state (PADEP), and federal regulations.			
34. Have you included Wet Collecting (Scrubber) Control Device in the Emissions Points Data Summary Sheet? Yes			

35. Proposed Monitoring, Recordkeeping, Reporting, and Testing Please propose monitoring, recordkeeping, and reporting in order to demonstrate compliance with the proposed operating parameters. Please propose testing in order to demonstrate compliance with the proposed emissions limits.	
MONITORING:	RECORDKEEPING:
REPORTING:	TESTING:
MONITORING: Please list and describe the process parameters and ranges that are proposed to be monitored in order to demonstrate compliance with the operation of this process equipment or air control device. RECORDKEEPING: Please describe the proposed recordkeeping that will accompany the monitoring. REPORTING: Please describe any proposed emissions testing for this process equipment or air pollution control device. TESTING: Please describe any proposed emissions testing for this process equipment or air pollution control device.	
36. Manufacturer's Guaranteed Capture Efficiency for each air pollutant.	
37. Manufacturer's Guaranteed Control Efficiency for each air pollutant. 0.997%	
38. Describe all operating ranges and maintenance procedures required by Manufacturer to maintain warranty.	

**Attachment L
EMISSIONS UNIT DATA SHEET
BULK LIQUID TRANSFER OPERATIONS**

Furnish the following information for each new or modified bulk liquid transfer area or loading rack, as shown on the *Equipment List Form* and other parts of this application. This form is to be used for bulk liquid transfer operations such as to and from drums, marine vessels, rail tank cars, and tank trucks.

Identification Number (as assigned on <i>Equipment List Form</i>):				
1. Loading Area Name:				
2. Type of cargo vessels accommodated at this rack or transfer point (check as many as apply): <input type="checkbox"/> Drums <input type="checkbox"/> Marine Vessels <input type="checkbox"/> Rail Tank Cars <input checked="" type="checkbox"/> Tank Trucks				
3. Loading Rack or Transfer Point Data:				
Number of pumps				
Number of liquids loaded	2			
Maximum number of marine vessels, tank trucks, tank cars, and/or drums loading at one time	1			
4. Does ballasting of marine vessels occur at this loading area? <input type="checkbox"/> Yes <input type="checkbox"/> No <input type="checkbox"/> Does not apply				
5. Describe cleaning location, compounds and procedure for cargo vessels using this transfer point:				
6. Are cargo vessels pressure tested for leaks at this or any other location? <input type="checkbox"/> Yes <input type="checkbox"/> No If YES, describe:				
7. Projected Maximum Operating Schedule (for rack or transfer point as a whole):				
Maximum	Jan. - Mar.	Apr. - June	July - Sept.	Oct. - Dec.
hours/day				
days/week				

weeks/quarter				
---------------	--	--	--	--

8. Bulk Liquid Data (add pages as necessary):						
Pump ID No.						
Liquid Name	HCL					
Max. daily throughput (1000 gal/day)	5k					
Max. annual throughput (1000 gal/yr)	1300					
Loading Method ¹	sub					
Max. Fill Rate (gal/min)	1000					
Average Fill Time (min/loading)	500					
Max. Bulk Liquid Temperature (°F)	75					
True Vapor Pressure ²						
Cargo Vessel Condition ³						
Control Equipment or Method ⁴						
Minimum control efficiency (%)						
Maximum Emission Rate	Loading (lb/hr)					
	Annual (lb/yr)					
Estimation Method ⁵						
¹ BF = Bottom Fill SP = Splash Fill SUB = Submerged Fill						
² At maximum bulk liquid temperature						
³ B = Ballasted Vessel, C = Cleaned, U = Uncleaned (dedicated service), O = other (describe)						
⁴ List as many as apply (complete and submit appropriate Air Pollution Control Device Sheets): CA = Carbon Adsorption LOA = Lean Oil Adsorption CO = Condensation SC = Scrubber (Absorption) CRA = Compressor-Refrigeration-Absorption TO = Thermal Oxidation or Incineration CRC = Compressor-Refrigeration-Condensation VB = Dedicated Vapor Balance (closed system) O = other (describe)						
⁵ EPA = EPA Emission Factor as stated in AP-42 MB = Material Balance						

TM = Test Measurement based upon test data submittal
O = other (describe)

9. Proposed Monitoring, Recordkeeping, Reporting, and Testing

Please propose monitoring, recordkeeping, and reporting in order to demonstrate compliance with the proposed operating parameters. Please propose testing in order to demonstrate compliance with the proposed emissions limits.

MONITORING	RECORDKEEPING
REPORTING	TESTING

MONITORING. PLEASE LIST AND DESCRIBE THE PROCESS PARAMETERS AND RANGES THAT ARE PROPOSED TO BE MONITORED IN ORDER TO DEMONSTRATE COMPLIANCE WITH THE OPERATION OF THIS PROCESS EQUIPMENT OPERATION/AIR POLLUTION CONTROL DEVICE.

RECORDKEEPING. PLEASE DESCRIBE THE PROPOSED RECORDKEEPING THAT WILL ACCOMPANY THE MONITORING.

REPORTING. PLEASE DESCRIBE THE PROPOSED FREQUENCY OF REPORTING OF THE RECORDKEEPING.

TESTING. PLEASE DESCRIBE ANY PROPOSED EMISSIONS TESTING FOR THIS PROCESS EQUIPMENT/AIR POLLUTION CONTROL DEVICE.

10. Describe all operating ranges and maintenance procedures required by Manufacturer to maintain warranty

--



101 Fairview Avenue
Pittsburgh, PA 15222

EMAIL: SALES@V-SYST.COM
Web: www.v-syst.com

Telephone: 412-228-8200
Fax: 412-228-2182

Quote #2023-560-H-R

July 26, 2023

Mr. Farley R. Wood, P.E.
Vice President of Engineering
Empire Diversified Energy
1400 Main Street
Follersbee, WV 26037

Subject: Chemical System - Scrubber – Quote #2023-560-H-R

Dear Mr. Wood,

Please find attached our proposal for the above referenced equipment/project. We appreciate the opportunity to provide a quote for this opportunity.

You will also find our most recent line card attached for your reference. I hope you will think of us during your next project. If you would have any questions or require additional information, please give us a call at (412) 826-8200.

Sincerely,

Russell C. Hufftmyer
President & CEO
V-Systems, Inc.
101 Fairview Avenue
Pittsburgh, PA 15238

Enclosure

ansr

TOLL FREE: 1 (888) 826-0225  EMAIL: SALES@V-SYST.COM

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101 Fairview Avenue
Pittsboro, PA 15222

EMAIL: sales@v-system.com
Web: www.v-system.com

TELEPHONE: 212-225-9200
FAX: 212-225-2152

Quotation:

Project Name: Chemical System - Scrubber **Contact Name:** Farley Wood
Company Name: Empire Diversified Energy **Email/ fax:** fwood@empireda.com
Address/Street: 1400 Main Street **Phone:** 304-914-2624
City/State/Zip: Follensbee, WV 26037 **Date:** July 21, 2023 / Revised July 26, 2023
Quote Number: 2023-560-H-R

Thank you for the opportunity to provide you with the following quote:

Quantity	Item Description	Net Price Each	Total Net Price
<u>Lids/ Scrubber Tank</u>			
1	PolyScrub Scrubber Tank, 700-Open Top, Rated: 1.90 Specific Gravity Wall Thickness, Material: Crosslinked Polyethylene (XLPE)4, Color: Natural (yellowish white) <ul style="list-style-type: none"> • (1) Lid/Manway - 81" Cover Assembly Open Top /Stainless Steel/Pe • (1) Vent - 8" U-Vent PVC • (1) Outlet / Overflow - 1" Scrubber Outlet/Overflow PVC/Charge Viton /o-276 • (1) Scrubber - 8" Scrubber Assembly PVC/Charge Viton • (1) Fill - 1" Bulkhead Fitting Assembly Socket x thread PVC/Charge Viton • Warranty - 5 Years, Full Replacement, Non-Prorated Includes Product Engineering / Permitting Support for Permit Application	\$16,500.00	\$16,500.00
<u>Lids/ Tank</u>			
1	10,305-Gallon Vertical Tank, Rated: 1.90 Specific Gravity Wall Thickness, Material: Crosslinked Polyethylene (XLPE)4, Color: Natural (yellowish white) <ul style="list-style-type: none"> • (1) Lid/Manway - 24" Manway Cover 24" Fume Tight /Stainless Steel/pe • (1) Fill - 2" Bulkhead Fitting Assembly Socket x thread PVC/EPDM • (2) Dome Fitting - 2" Bulkhead Fitting Assembly Socket x thread PVC/EPDM • (3) Sidewall Fitting - 2" Bolted Flange Fitting Socket PVC/o-276/EPDM • (1) Vent - 8" U-vent with Bolted Flange PVC/o-276/EPDM • WARRANTY: 5 Years, Full Replacement, Non-Prorated 	\$30,675.00	\$30,675.00

V-SYSTEMS

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101 Fairview Avenue
Pittsburgh, PA 15222

EMAIL: sales@v-syst.com
Web: www.v-syst.com

TELEPHONE: 412-228-9200
FAX: 412-228-2152

<u>Leg: Items</u>			
2	March Model# TE-10K-MD 3PH 10 HP - Inlet: 3" MPT, Outlet: 2" MPT, Wet End: Natural Kynar (Front Housing, Rear Housing), Glass Filled Kynar (Impeller) with 6.625" Impeller Trim, Viton (Gasket), Carbon (Bushing), Ceramic (Shaft, Thrust Washers), and driven by a 10 HP, 3500 RPM, 3/60/230/460, TEFC Motor	\$7,920.00	\$16,840.00
<u>Note:</u>			
<ul style="list-style-type: none"> • Lead Time is currently 1 Week, A.R.O. • FOB-out of March in IL (must ship by LTL due to size and weight) 			
<u>Leg: VFDs</u>			
2	Xylem, Variable Frequency Drives, 10 HP, 460-3-40, NEMA 3R, BACnet	\$7,780.00	\$15,560.00
<u>Note:</u>			
<ul style="list-style-type: none"> • Lead Time is currently at 7 Weeks, A.R.O. 			
<u>Leg: Chem Feed Skid</u>			
1	SS1-C_FLOOR_050_PVCEPDM_PD PPYRE Prominent Skid for solenoid driven pumps, (20"W x 18"D x 40"H) 1/2" PVC/EPDM socket weld pipe and fittings Wye strainer 500ml PVC calibration volume 164ml CPVC/EPDM pulsation dampers Pressure relief valves Pressure gauge with isolator Back pressure valve Plumbing and components rated at 150 PSI regardless of pump pressure.	\$21,505.00	\$21,505.00
<ul style="list-style-type: none"> • (1) GMXAC708PVT2000UDC1300EN Prominent Gamma/X 2 GPH/102PSI, PVDF/PTFE, bleed valve w/spring, 4-20mA output • (1) Prominent CP1 ONE PUMP 120VAC SCADA PANEL 			
Total Quoted Amount:			\$109,000.00

QUOTED BY: RUSS HUFFMYER

If you need further information concerning the products that have been included in the quote, please feel free to contact me at 412-228-9200 and/or rhuffmyer@v-syst.com.

We appreciate the opportunity to provide you with this quote and look forward to working with you on this important project.

Thank you,



Russell C. Huffmyer
President & CEO

arsr

V-SYSTEMS

COMMERCIAL | INDUSTRIAL | MUNICIPAL

101 Fairview Avenue
Pittsburgh, PA 15222

Email: sales@v-system.com
Web: www.v-system.com

Telephone: 412-225-6200
Fax: 412-225-2152

THIS QUOTATION OR SELLER'S ACCEPTANCE OF THIS ORDER IS EXPRESSLY LIMITED TO, AND EXPRESSLY MADE CONDITIONAL ON, BUYER'S ACCEPTANCE OF THE V-SYSTEMS-TEC, INC. STANDARD TERMS AND CONDITIONS OF SALE. A COPY OF THESE TERMS AND CONDITIONS IS AVAILABLE AT <http://www.v-system.com/terms-and-conditions-of-sale-and-service>. SELLER OBJECTS TO ANY DIFFERENT OR ADDITIONAL TERMS.

General Comments

Warranty applies per Sales & Service Terms and Conditions if the following are met:

- Equipment installed per industry standards and manufacturer instruction manual.
- Operation of equipment in accordance with manufacturer instruction manual.
- Maintenance and lubrication per manufacturer instruction manual. Note, maintenance log showing dates required.
- Equipment must be stored per manufacturer instruction manual and protected from the weather.

If warranty items occur, V-Systems needs to be contacted in writing before any repairs are made, whereas a mutual course of action will be performed. Equipment cannot be disassembled without V-Systems being present.

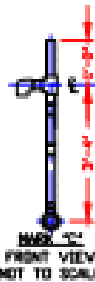
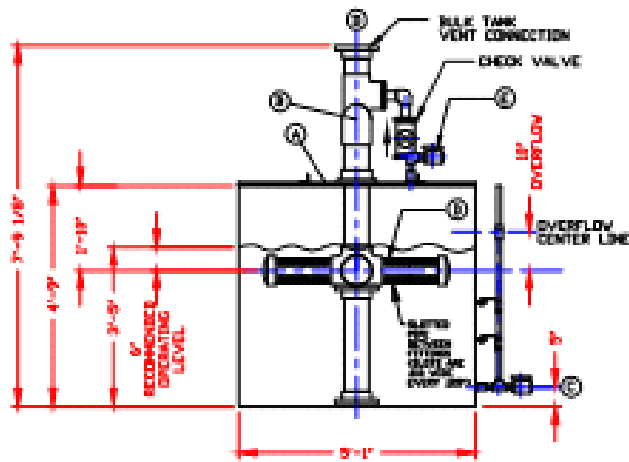
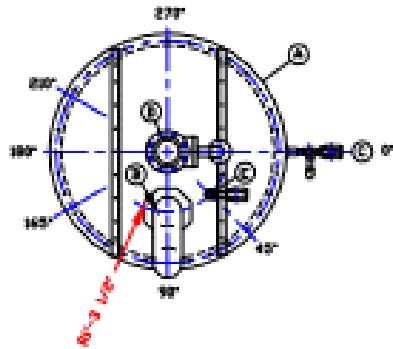
Acknowledged and Accepted by Buyer:

Name: _____ *Tax Exempt? Yes ___ No ___

Signature: _____ PO #: _____

Date: _____ Ship To: _____

*IF APPLICABLE, please send a copy of your company's tax-exempt form. Otherwise, our accounting department will assume that this order is taxable.

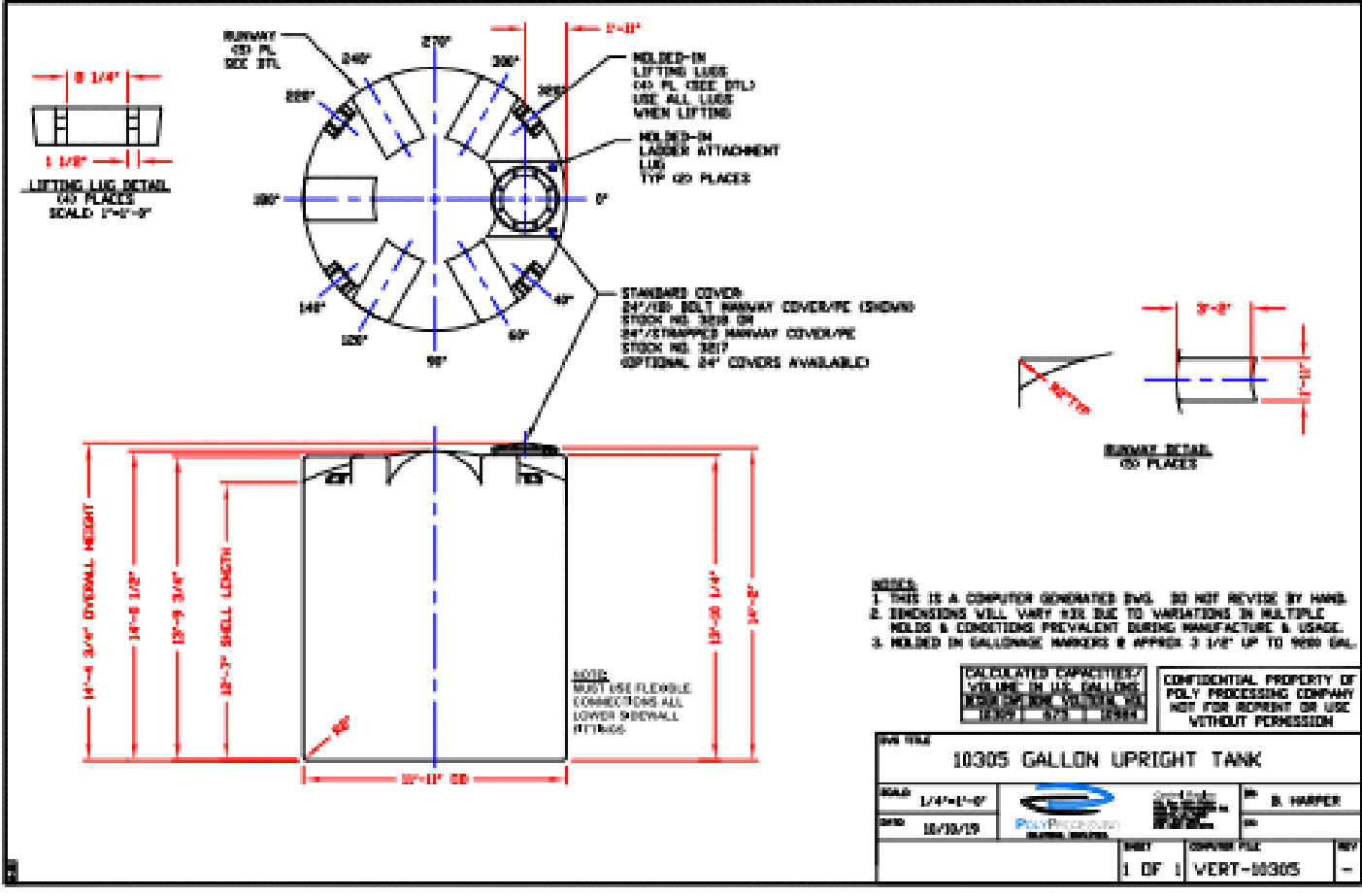


NOTE: INCOMING VENT PIPE MUST BE INDEPENDENTLY SUPPORTED
 ELEVATIONS ONLY---
 HOLE SIZE & ACCESSORIES
 ROTATED INTO VIEW---
 FOR TRUE ORIENTATION
 SEE PLAN VIEW

- NOTES:
 1. THIS IS A COMPUTER GENERATED DWG. DO NOT REVISE BY HAND.
 2. DIMENSIONS WILL VARY +/- DUE TO VARIATIONS IN MULTIPLE HOLE & COMPONENT PREVALENT DURING MANUFACTURE & USAGE.
 3. TANK DESIGNED FOR L3 SpG MAT'L @ 100% ATMOS PRESSURE.

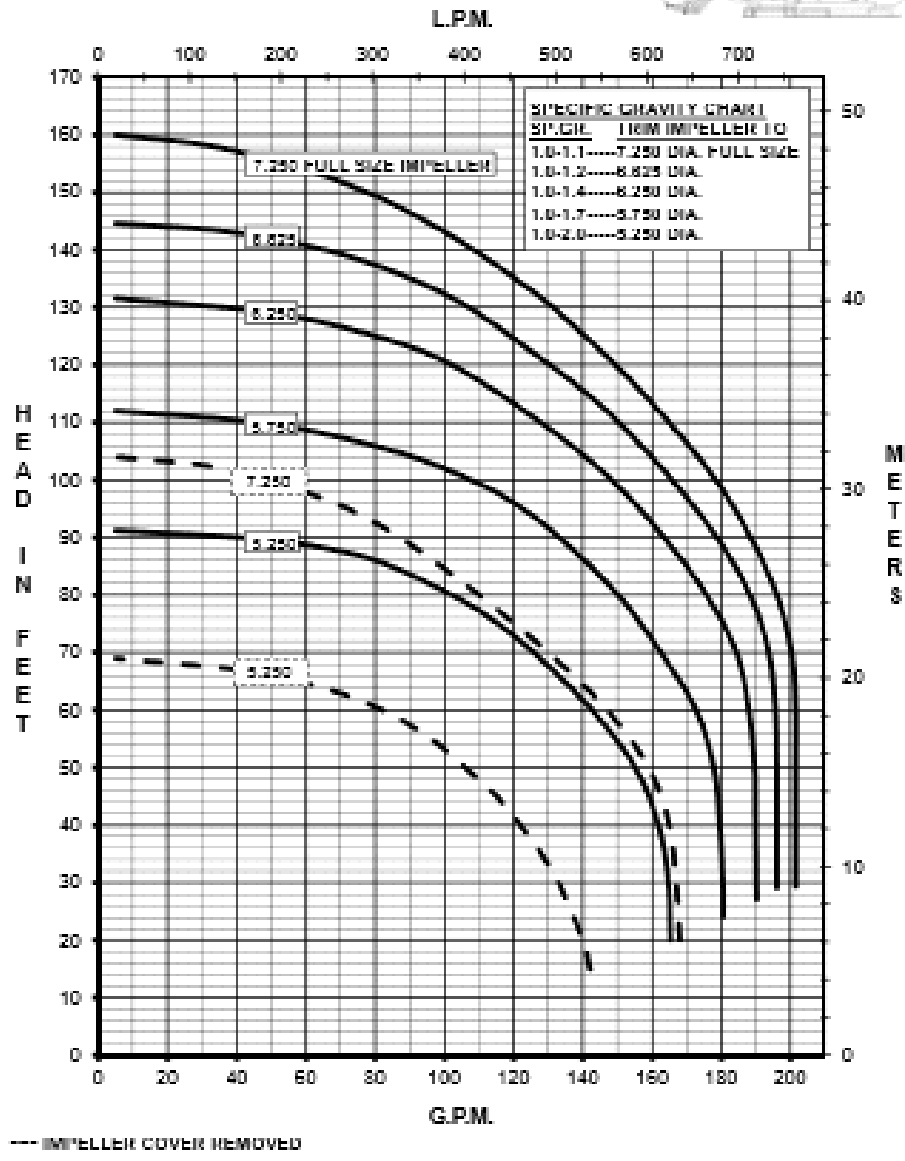
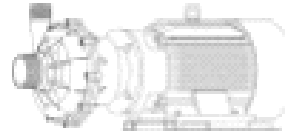
PART NAME		POLYSCRUB BASIC	
		5' X 5' SCRUBBER	
SERVICE/SCRUBBER MEDIA		L3 SpG/ALP/NATURAL	
SCALE	1/2"=1'-0"	BY	B. HANSEN
DATE	4/12/2023	APP'D	
CALCULATED CAPACITY TO OPERATING LEVEL		DRAWING FILE	
-500 GALLONS		1 OF 1 POLYSCRUB BASIC	

NON-CONTROLLED COPY
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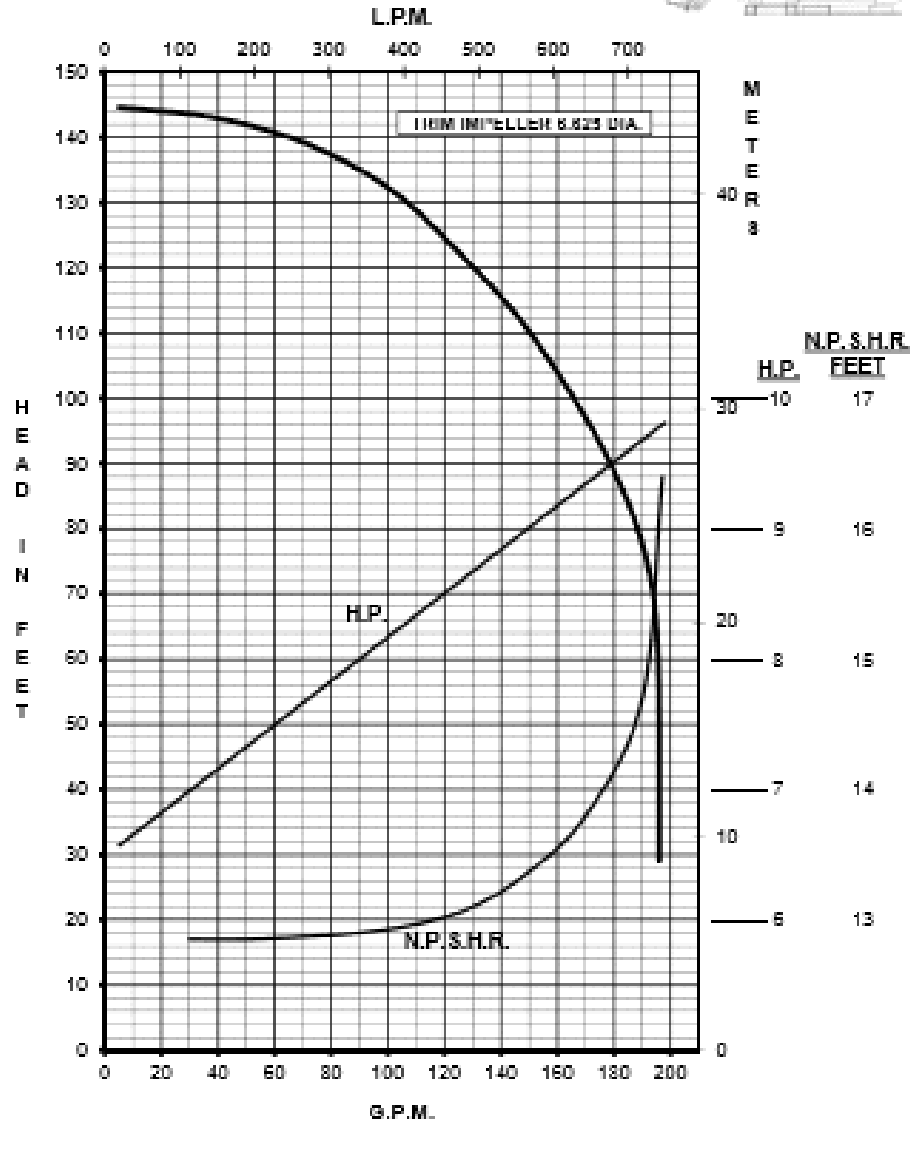
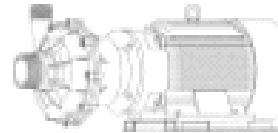
TE-10K-MD, TE-10P-MD 3PH 60HZ TRIM CURVES





MARCH
PUMPS

TE-10K-MD, TE-10P-MD
TRIM IMPELLER 6.625 DIA.
3PH 60HZ



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Steam Boilers

- Efficiencies up to 90%
- Low Pressure & High Pressure Applications
- Finetube Design
- Natural Gas, Propane, #2 Fuel Oil, True Dual Fuel (NG & #2 Oil)
- Size Range: 4 to 2500 HP

Hydronic Boilers

- Efficiencies up to 99.1%
- Condensing, Non-Condensing, & Non-Condensing Boilers
- Finetube & Watertube Designs
- Natural Gas, Propane, #2 Fuel Oil, True Dual Fuel (NG & #2 Oil)
- Size Range from 55,000 to 12,000,000 BTU (y)

Additional Offerings:

- Thermal Fluid Heaters
- Ancillary Equipment
- Custom Engineered Systems



Hydronic Boilers

- Models from 55,000 to 8.0 Million Btu/Hr
- Up to 99.1% Thermal Efficiency
- Up to 15:1 Turndown Ratio
- Featuring CONX-US® Remote Connectivity and SMART TOUCH™ Operating Control

Thermal Fluid Heaters

- Models from 125,000 to 4,000,000 Btu/Hr
- Up to 99% Thermal Efficiency
- Up to 10:1 Turndown Ratio
- Direct Vent Flexibility up to 150 Ft.
- SMART TOUCH™ Operating Control and CONX-US® Remote Connect

Additional Offerings:

- High Efficiency Condensing Type
- Non-Condensing Boilers
- Pool Heaters
- Buffer Tanks, Storage Tanks
- Ancillary Equipment
- Custom Engineered Systems



Pumps (Commercial)

- Base Mounted
- Circulating
- Vertical Split Case
- Dual Arm
- Vertical In-Line Centrifugal

Fire Pumps (Commercial)

- Electric & Diesel Driven Systems
- Horizontal Split Case
- Vertical In-Line
- Vertical Turbine
- Skid & House System Packages
- Design Envelope Fire Pump Unit

Performance Range:

- Up to 18,000 US GPM flow
- Up to 500 Ft. head

Temperature:

- 200 Degree Fahrenheit

Power Range/Size:

- 1/12 HP to 1250 HP
- 3/4" to 10"
- Fire Pump Systems of 50 GPM up to 2,000 GPM

Additional Offerings:

- Hydraulic Specialties
- Heat Exchangers
- Expansion Tanks
- Glycol Feeders

Parts and Service:

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Contact:

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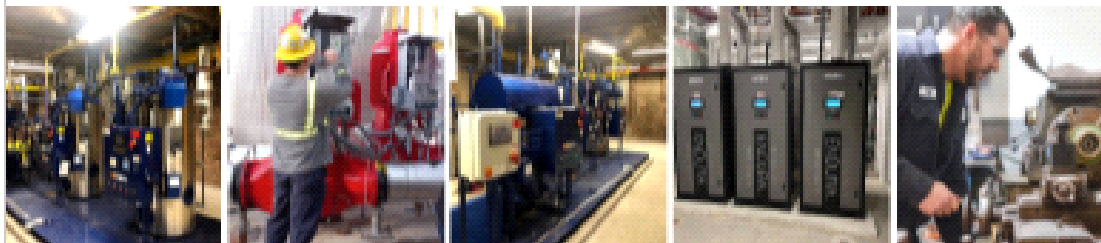
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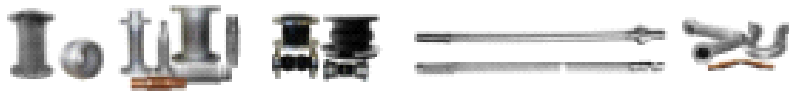
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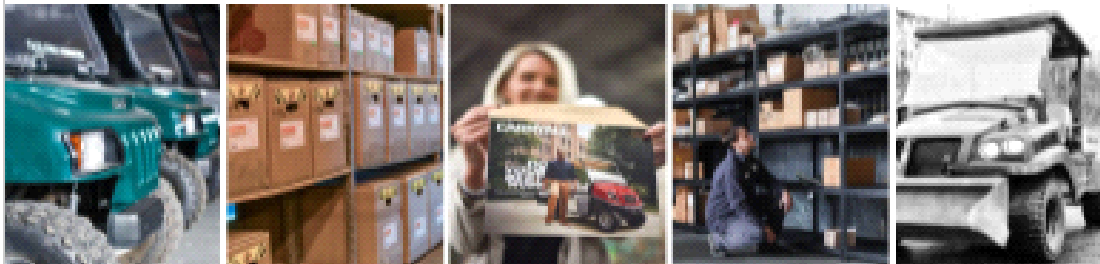
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NHSM Response to RIR 3/7/2024

Wednesday, March 27, 2024 2:03 PM



Andrews, Edward S <edward.s.andrews@wv.gov>

RE: Regulatory Interpretation Request - EGG

2 messages

Morrison, Jacqueline <Morrison.Jacqueline@epa.gov>
To: "edward.s.andrews@wv.gov" <edward.s.andrews@wv.gov>
Cc: "Supplee, Gwendolyn" <Supplee.Gwendolyn@epa.gov>

Thu, Mar 7, 2024 at 3:32 PM

Hi Ed,

Thanks for sending. We will get back to you, but I'd assume that the response from the waste perspective will be similar to when we last considered this facility last year.

Thank you.

Jacque

Jacqueline Morrison

RCRA Programs Section, Hazardous Waste
Land, Chemicals, & Redevelopment Division
Mail Code 3LD31

US EPA Mid-Atlantic Region

Address: Four Penn Center,

1600 John F. Kennedy Boulevard,

Philadelphia, PA 19103-2852

Phone: 215-814-5664

Email: morrison.jacqueline@epa.gov

From: Andrews, Edward S <edward.s.andrews@wv.gov>

Sent: Wednesday, March 06, 2024 3:35 PM

To: Willson, Matthew (he/him/his) <Willson.Matthew@epa.gov>; Morrison, Jacqueline <Morrison.Jacqueline@epa.gov>

Subject: Fwd: Regulatory Interpretation Request - EGG

Caution: This email originated from outside EPA, please exercise additional caution when deciding whether to open attachments or click on provided links.

FYI on a Regulatory Interpretation Request

Thanks,

Ed

----- Forwarded message -----

From: **Andrews, Edward S** <edward.s.andrews@wv.gov>

Date: Wed, Mar 6, 2024 at 2:54 PM

Subject: Regulatory Interpretation Request - EGG

To: Cristina Fernandez <fernandez.cristina@epa.gov>, Supplee, Gwendolyn <supplee.gwendolyn@epa.gov>, Marycate Opila <opila.marycate@epa.gov>

Cc: Beverly D Mckeone <beverly.d.mckeone@wv.gov>, Crowder, Laura M <laura.m.crowder@wv.gov>

Please see the attached file (Reg_Inter_Req_for_EGG) from the WVDEP/DAQ Regulatory Interpretation Request regarding Empire Green Generation's proposed pyrolysis units processing plastic feedstock.

Should you or your staff have any questions about this request, please let me know.

Thanks,

Ed

--

Edward Andrews, P.E.

Engineer

WVDEP/Division of Air Quality

304-926-0499 Ext 41244

[601 57th Street, SE](#)

[Charleston, WV 25304](#)

--

Edward Andrews, P.E.

Engineer

WVDEP/Division of Air Quality

304-926-0499 Ext 41244

601 57th Street, SE

Charleston, WV 25304

Andrews, Edward S <edward.s.andrews@wv.gov>

Fri, Mar 8, 2024 at 7:38 AM

Draft To: "Morrison, Jacqueline" <Morrison.Jacqueline@epa.gov>

Cc: "Supplee, Gwendolyn" <supplee.gwendolyn@epa.gov>, Krystal <Stankunas.Krystal@epa.gov>

Please keep in mind that the facility has elected to change from processing medical waste to plastic feedstock. I have been trying to get them to perform a waste/non-waste determination of this plastic feedstock in accordance with Part 241.

EGG keeps pushing back with a State House bill that defined "advance recycling" and "high temperature under WV

[Quoted text hidden]

Follow-up on Timing for the RIR 3/7/2024

Wednesday, March 27, 2024 2:02 PM



Andrews, Edward S <edward.s.andrews@wv.gov>

Re: Regulatory Interpretation Request - EGG

1 message

Andrews, Edward S <edward.s.andrews@wv.gov>

Thu, Mar 7, 2024 at 2:32 PM

To: "Supplee, Gwendolyn" <Supplee.Gwendolyn@epa.gov>

Cc: Beverly D Mckeone <beverly.d.mckeone@wv.gov>, "Stankunas, Krystal" <Stankunas.Krystal@epa.gov>

Gwen,

We understand that this is fairly complicated and that we expected multiple different offices to be called on to provide input to our request.

Please keep in mind that EGG has a permit to process medical waste using the same equipment. The other real changes to the process is adding the HCI tanks and loadout station.

So, EGG will be applying pressure at some point to get their application moving forward in the near future.

We expect to send EGG an additional information request based on your office's response.

At some point, we will have to move this application forward to some sort of decision with or without a response.

Please provide us some sort of response by no later than March 29.

Thanks,
Ed

On Thu, Mar 7, 2024 at 1:11 PM Supplee, Gwendolyn <Supplee.Gwendolyn@epa.gov> wrote:

Hi Ed,

We need to coordinate with our RCRA program as well as potentially Headquarters (either for RCRA or CAA requirements) and that coordination can take some time. Since the application hasn't been deemed complete yet, would it be OK if we try to get a response to West Virginia by the end of March? We'll try to get a response sooner if we can.

Many thanks.

-gwen



Air & Radiation Division

Phone 215-814-2763

Email supplee.gwendolyn@epa.gov

From: Andrews, Edward S <edward.s.andrews@wv.gov>
Sent: Thursday, March 07, 2024 10:54 AM
To: Supplee, Gwendolyn <Supplee.Gwendolyn@epa.gov>
Cc: Beverly D Mckeone <beverly.d.mckeone@wv.gov>; Stankunas, Krystal <Stankunas.Krystal@epa.gov>
Subject: Re: Regulatory Interpretation Request - EGG

Caution: This email originated from outside EPA, please exercise additional caution when deciding whether to open attachments or click on provided links.

Gwendolyn,

WVDEP/DAQ received the initial application on December 1, 2023, and a revised application was received January 23, 2024. Currently, this application has not been deemed complete at this time.

Therefore, the regulatory clock has not started yet.

We would like some sort of response from your office within 2 weeks. A response to our request will dictate our future request for additional information from this applicant (EGG).

Should you have any questions about this, please do not hesitate to contact me.

Thanks

Ed

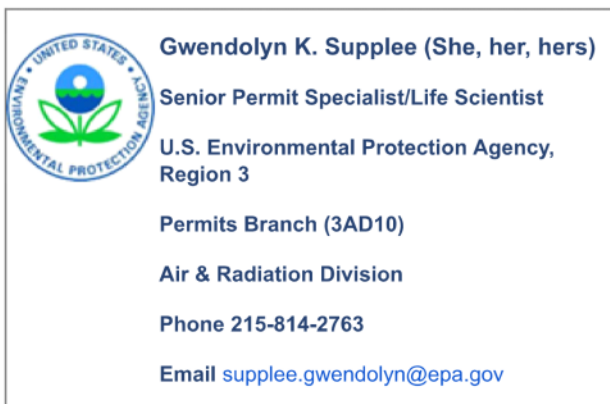
On Thu, Mar 7, 2024 at 10:30 AM Supplee, Gwendolyn <Supplee.Gwendolyn@epa.gov> wrote:

Ed-

Can you tell us when the application was received and what date the permitting decision must be made by? In looking at WV's R13 rule, it looks like a permit must be issued within 90 days of the completeness determination? Is my interpretation correct? We're trying to determine when WV needs a response by.

Many thanks,

-gwen



From: Andrews, Edward S <edward.s.andrews@wv.gov>

Sent: Wednesday, March 06, 2024 2:55 PM

To: Fernandez, Cristina <Fernandez.Cristina@epa.gov>; Supplee, Gwendolyn <Supplee.Gwendolyn@epa.gov>;
Opila, MaryCate <Opila.MaryCate@epa.gov>

Cc: Beverly D Mckeone <beverly.d.mckeone@wv.gov>; Crowder, Laura M <laura.m.crowder@wv.gov>

Subject: Regulatory Interpretation Request - EGG

Caution: This email originated from outside EPA, please exercise additional caution when deciding whether to open attachments or click on provided links.

Please see the attached file (Reg_Inter_Req_for_EGG) from the WVDEP/DAQ Regulatory Interpretation Request regarding Empire Green Generation's proposed pyrolysis units processing plastic feedstock.

Should you or your staff have any questions about this request, please let me know.

Thanks,

Ed

--

Edward Andrews, P.E.

Engineer

WVDEP/Division of Air Quality

304-926-0499 Ext 41244

601 57th Street, SE

Charleston, WV 25304

--

Edward Andrews, P.E.

Engineer

WVDEP/Division of Air Quality

304-926-0499 Ext 41244

601 57th Street, SE

Charleston, WV 25304

Regulatory Interpretation Request 3/6/2024

Wednesday, March 27, 2024 2:00 PM



Andrews, Edward S <edward.s.andrews@wv.gov>

Regulatory Interpretation Request - EGG

1 message

Andrews, Edward S <edward.s.andrews@wv.gov>

Wed, Mar 6, 2024 at 2:54 PM

To: Cristina Fernandez <fernandez.cristina@epa.gov>, "Supplee, Gwendolyn" <supplee.gwendolyn@epa.gov>, Marycate Opila <opila.marycate@epa.gov>

Cc: Beverly D Mckeone <beverly.d.mckeone@wv.gov>, "Crowder, Laura M" <laura.m.crowder@wv.gov>

Bcc: Brian S Tephacock <Brian.S.Tephacock@wv.gov>, Eric Blend <eric.n.blend@wv.gov>

Please see the attached file (Reg_Inter_Req_for_EGG) from the WVDEP/DAQ Regulatory Interpretation Request regarding Empire Green Generation's proposed pyrolysis units processing plastic feedstock.


Should you or your staff have any questions about this request, please let me know.

Thanks,
Ed


--

Edward Andrews, P.E.
Engineer
WVDEP/Division of Air Quality
304-926-0499 Ext 41244
601 57th Street, SE
Charleston, WV 25304

2 attachments

 **R13-3555 Modification_Application_EGG - Redacted_Final Rev.pdf**

22432K

 **Reg_Interp_Request_for_EGG.pdf**

400K



Reg_Interp
_Request...



west virginia department of environmental protection

Division of Air Quality
601 27th Street, SE
Charleston, WV 25304
(304) 236-6173

Harold D. Ward, Cabinet Secretary
dep@wv.gov

March 6, 2023

Ms. Christina Fernandez
Director
U.S. EPA - Region 3
Air and Radiation Division
Four Penn Center
1600 John F. Kennedy Boulevard
Philadelphia, PA 19103-2852

Re: Regulatory Interpretation Request
Empire Green Generation LLC
Facility ID: 009-00141
Permit No.: R13-3555A
Follansbee, WV

Dear Director:

The West Virginia Department of Environmental Protection - Division of Air Quality (DAQ) respectfully requests an regulatory interpretation from the Administrator regarding Empire Green Generation's (EGG) proposed modification of their Follansbee, West Virginia Facility to any regulation developed under Section 129 of the Clean Air Act.

Specifically, the DAQ is requesting an regulatory interpretation as to whether all streams, or only the liquid and solid streams, exiting the pyrolysis process need to be evaluated under 40 CFR 241 to determine applicability under 40 CFR 60, Subpart OOOO, if the plastic feedstock to the pyrolysis process has been determined to be a fuel or raw material under 40 CFR 241.

The DAQ does not believe that EGG pyrolysis trains or downstream emission units (e.g., engines, dryer, and vitrifier) would be affected sources under Subpart AAAA and EEEE because the plastic feedstock does not meet the definition of municipal solid waste and the Follansbee Facility is not an institutional facility.

Promoting a healthy environment.

N8P8 Applicability Determination Request

March 6, 2024

Page 2 of 8

Background Information

In 2022, Empire Green Generation LLC (EGG) proposed to the DAQ to construct and operate two pyrolysis trains with gas cleaning sections to process and convert up to 70 tons per day of medical waste into tar (liquids), char (solids) and synthetic gas.

The DAQ issued Permit R13-3555 to EGG on March 2, 2023. During the DAQ review of the Application, the DAQ determined that EGG's proposed pyrolysis trains meet the criteria of a "pyrolysis unit" as defined in 40 CFR 60.51c and therefore the proposed pyrolysis trains are excluded emission units from Subpart Ec.

Proposed Modification

On December 1, 2023, EGG filed a modification application with the DAQ. EGG proposed to replace the medical waste feedstock with plastic feedstock. In a revised application (January 23, 2024, Submission), EGG noted that the feedstock will be sourced from recyclers, manufacturing, and plastic producers. This pre-processed plastic feedstock will be shipped to EGG's Follansbee, WV facility as feedstock for the pyrolysis trains.

EGG noted that this modification only requires the addition of a hydrochloric acid truck loading facility with associated scrubber system. This feedstock change will allow the pyrolysis trains to generate hydrochloric acid in addition to other products (tars, char, and synthetic gas).

The processing capacity of these pyrolysis trains will remain the same at 35 tons of plastic feedstock per day per pyrolysis train (70 Tons per day total).

Regulatory Considerations

EGG commenced construction of the pyrolysis trains in 2023, which is after the applicability date of Subparts AAAA (August 30, 1999); OOOO (June 4, 2010); and EEEE (December 9, 2004). Therefore, EGG's pyrolysis trains meet the definition of new affected units.

Given the design capacity of the two pyrolysis trains, these units do not meet the capacity criteria of a large municipal waste combustion unit as defined under Subpart Eb and therefore, the units are not subject to Subpart Ec.

The DAQ determined that the four spark ignition engines are affected sources with regard to Subpart JJJJ during the review of Permit R13-3555. However, the DAQ was unable to determine the applicable emission standard to which the permit engines were subject. Condition 5.1.1. of Permit R13-3555 required EGG to seek a determination from the EPA to determine which emission standard would be applicable for these four engines.

The vitrifiers (process heaters) for the pyrolysis trains may be affected sources under Subpart Dc of Part 60 and Subpart JJJJJ of Part 63. Applicability status for these process heaters would be affected by the outcome of this determination. These units are designed to fire gaseous fuel (synthetic gas), liquid fuel (tars), propane for startup operations only, or a combination of synthetic gas and tars with a maximum heat input of 100 MMBtu/hr.

Past Determinations and Other Permitting Actions

Prior to submitting this determination, the DAG searched the Applicability Determination Index (ADI) for similar determinations, and identified the following:

Table 1 - Similar Applicability Determinations			
EPA Office	Control Number	Date	Reference
Region 5	9700062	10/11/1996	60.14, 60.15, 60.5, 60.51b,
Region 6	NR06	02/07/1985	60.21(b), 60.50, 60.51(b)
Region 7	9600086	12/02/1996	60.50b
Region 10	E010	04/12/1977	60.50, 60.51
D88E	E009	01/19/1977	60.50
Region 1	M140002	12/04/2012	40 CFR 60, Subpart EEEE
Region 9	1000019	03/30/2010	40 CFR 60, Subpart AAAA
Region 10	1500025	08/31/2010	40 CFR 60, Subpart AAAA
Region 4	1700010	03/02/2017	40 CFR 60, Subpart CCCC
Region 5	1800003	01/22/2018	40 CFR 60, Subpart CCCC

The DAQ is aware of several determinations by other State Agencies, listed below:

State Agency	Company/Permittee Name	Permit No.	Date	Outcome
Indiana Department of Environmental Management	Fulcrum Centerpoint LLC	089-44042-00660	08/16/2022	Meet exemption under 60.1020(h)
North Carolina Environmental Quality	Eraven Environmental LLC	10672RD0	9/25/2020	No Reference
Indiana Department of Environmental Management	Brightmark Plastics Renewal Indiana 2 LLC	151-45294-00067	06/29/2022	Meet exemption under 60.1020(h)
Ohio Environmental Protection Agency	SOBE Thermal Energy Systems, LLC	PD132799	02/14/2024	Scrap Tires are classified as non-waste per 40CFR241. Not Applicable to any Incinerator rules

The DAQ is aware of EPA's decision to not remove the phrase "pyrolysis/combustion unit" for the definition of municipal waste combustion unit in Subparts A,AAA, and EEEE of Part 60.¹

The DAQ is working under the assumption that EGG's plastic feedstock is non-hazardous per 40 CFR 262.

First Question: Should EGG's plastic feedstock be viewed as waste or non-waste?

The DAQ believes that the proper way to answer this question is for EGG/EGG's plastic feedstock provider to make a waste/non-waste determination of this plastic feedstock in accordance with 40 CFR 241. Based on EGG's application and additional responses regarding this question, EGG may have determined or believe that the plastic feedstock is a non-waste. EGG has not provided any information that DAQ would view as an official determination in accordance with 40 CFR 241.

¹ Standards of Performance for New Stationary Sources and Emission Guidelines for Existing Sources: Other Solid Waste Incineration Units Review; Withdrawal of Proposed Provision Removing Pyrolysis/Combustion Units, 88 Fed. Reg. 26524 (June 5, 2023).

NBPS Applicability Determination Request

March 6, 2024

Page 5 of 8

DAQ looked at the definitions that pertain to waste under potentially applicable subparts, which are as follows:

"Solid waste" is defined under Subpart EEEE as

"means any garbage, refuse, sludge from a waste treatment plant, water supply treatment plant, or air pollution control facility and other discarded material, including solid, liquid, semisolid, or contained gaseous material resulting from industrial, commercial, mining, agricultural operations, and from community activities, but does not include solid or dissolved material in domestic sewage, or solid or dissolved materials in irrigation return flows or industrial discharges that are point sources subject to permits under section 402 of the Federal Water Pollution Control Act, as amended (33 U.S.C. 1342), or source, special nuclear, or byproduct material as defined by the Atomic Energy Act of 1964, as amended (42 U.S.C. 2014)."

Subpart OGGG refers to "solid waste" as defined in 40 CFR 241.2, which refers to 40 CFR 258.2. 40 CFR 258.2 defines to *"means any garbage, or refuse, sludge from a wastewater treatment plant, water supply treatment plant, or air pollution control facility and other discarded material, including solid, liquid, semi-solid, or contained gaseous material resulting from industrial, commercial, mining, and agricultural operations, and from community activities, but does not include solid or dissolved materials in domestic sewage, or solid or dissolved materials in irrigation return flows or industrial discharges that are point sources subject to permit under 33 U.S.C. 1342, or source, special nuclear, or by-product material as defined by the Atomic Energy Act of 1964, as amended (50 Stat. 923)."*

Subparts AAAA do not define "solid waste" or reference waste as determined under 40 CFR 241. This subpart defines *"municipal solid waste or municipal-type solid waste"*

"means household, commercial/retail, or institutional waste. Household waste includes material discarded by residential dwellings, hotels, motels, and other similar permanent or temporary housing. Commercial/retail waste includes material discarded by stores, offices, restaurants, warehouses, nonmanufacturing activities at industrial facilities, and other similar establishments or facilities. Institutional waste includes materials discarded by schools, by hospitals (nonmedical), by nonmanufacturing activities at prisons and government facilities, and other similar establishments or facilities. Household, commercial/retail, and institutional waste does include yard waste and refuse-derived fuel. Household, commercial/retail, and institutional waste does not include used oil; sewage sludge; wood pallets; construction, renovation, and demolition wastes (which include railroad ties and telephone poles); clean wood; industrial process or manufacturing wastes; medical waste; or motor vehicles (including motor vehicle parts or vehicle fluff)."

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March 6, 2024

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Both of these definitions, "solid waste" and "municipal solid waste", contain the phrase "other discarded materials". Therefore, a waste determination must be conducted on the plastic feedstock to be introduced into the pyrolysis trains in accordance with 40 CFR 241.3.

The concern the DAQ has with the plastic feedstock is that EGG did not generate the plastic feedstock and thus, the original end user and/or generator had discarded this plastic material at some point.

Second Question: If the plastic feedstock is determined to be fuel or ingredients in accordance with 40 CFR 241.3, then would the EGG pyrolysis trains be exempt from Section 129 of the CAA (e.g. subject to Subpart AAAA, CCCC, or Subpart EEEE)?

Initially, the DAQ does not believe the pyrolysis trains, (e.g., engines and process heaters) would be subject to Section 129 and therefore Subparts AAAA, CCCC and EEEE would not be applicable to EGG's emission units.

Given the plastic material was discarded by the original end user or generator, the initial waste determination only pertains to the cracking/decomposition of plastic feedstock from the pyrolysis trains and all streams exiting the pyrolysis train should be re-evaluated in accordance with 40 CFR 241.3.

This is the real question: Would EGG need to conduct a waste determination for each of the exiting streams from the pyrolysis trains (e.g., "tars", "oil", "ash and char", and "synthetic gas") in accordance with 40 CFR 241.3?

The definition of "solid waste" at 40 CFR 258.6 states,

"Solid waste means any garbage, or refuse, sludge from a wastewater treatment plant, water supply treatment plant, or air pollution control facility and other discarded material, including solid, liquid, semi-solid, or contained gaseous material resulting from industrial, commercial, mining, and agricultural operations, and from community activities, but does not include solid or dissolved materials in domestic sewage, or solid or dissolved materials in irrigation return flows or industrial discharges that are point sources subject to permit under 33 U.S.C. 1342, or source, special nuclear, or by-product material as defined by the Atomic Energy Act of 1954, as amended (85 Stat. 923)."

Thus, the DAQ does not believe that EGG's cleaned synthetic gas would need to be evaluated because the gas is not stored in a container. EGG's process consumes the synthetic gas as fuel to provide process heat for their process and generates electricity for the facility.

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Third Question: EGG plans to route the ash and char stream to the vitrifier (process heater) to be oxidized into products of combustion. Would the vitrifier be subject to Subpart CCCC or EEEE?

The raw synthetic gas generated will exit the pyrolysis train through off-take with the ash and fixed carbon being collected in a deceleration chamber. The ash and fixed carbon, which is also referred to as "char", will be injected into the vitrifier. The vitrifier is best described as a retort or process heater with the purpose of providing process heat for the respective pyrolysis train. In the process description in EGG's modification application, the high temperatures in the vitrifier should be above the eutectic temperature of the ash and char to be combusted into CO₂ and H₂O.

The DAQ believes the synthetic gas stream and tar stream should be considered a fuel and the chlorine/chloride stream a raw ingredient for the production of hydrochloric acid. However, the injection of the ash and char into the vitrifiers should be viewed as incineration. The question is: because EGG generated the ash and char, would the vitrifiers be subject to Subpart CCCC as a CISM unit.

The DAQ does not believe that the vitrifiers could be classified as an OWSI unit because the Follansbee facility is not an institutional facility generating this waste and the initial plastic feedstock does not meet the definition of "municipal solid waste." Therefore, based on the definitions under 40 CFR 60.2577 the vitrifier(s) is not an "other solid waste incineration unit".

Furthermore, the DAQ does not believe that EGG's vitrifiers qualify for any of the exclusions in Subpart CCCC (e.g. cogeneration facilities, small power production facilities).

Fourth Question, Would the vitrifiers be considered an "energy recovery unit" or a "Commercial and Industrial solid waste Incineration unit" under Subpart CCCC?

The vitrifiers are required to provide process heat for the pyrolysis units with the exhaust used to dry the incoming plastic feedstock of excess moisture in the dryer section. EGG plans to use the generated "tars" and "cleaned synthetic gas" as fuels for the vitrifiers.

If EGG elected not to oxidize the ash and char stream in the vitrifiers (i.e. send the ash & char off-site for proper disposal), then would the vitrifiers be subject to Subpart CCCC?


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March 6, 2024
Page 8 of 8

To aid you and your staff in this determination, a redacted copy of EGG's modification application is attached. The DAQ's permit file for R13-3555, EGG's application on processing medical waste using pyrolysis, can be viewed in our Application Xtender at:
<https://documents.dep.wv.gov/AppXtender/DataSources/DEPAX16/account/login?ref=I-ww>

Instructions on using our Application Xtender are located at:
<https://dep.wv.gov/Data/Documents/AX-Instructions.pdf>.

Should you need to discuss this matter further, please do not hesitate to contact me by email at edward.s.andrews@wv.gov or phone at 304-926-0499 extension 41244.

Sincerely,
**Edward S.
Andrews, P.E.**
Edward S. Andrews, P.E.
Engineer

 Digitally signed by Edward S. Andrews, P.E.
DN: cn = Edward S. Andrews, P.E., email =
edward.s.andrews@wv.gov, o = DEP
WVDEP/Division of Air Quality, ou = Permitting
Date: 2024.03.06 14:22:41 -0500

cc:
Laura Crowder, Director, WV DAQ
Beverly McKeone, NSR Program Manager, WV DAQ
MaryCate Opila, opila.marycate@epa.gov
Gwendolyn Supplee, Supplee.Gwendolyn@epa.gov

HCI Discussion 3/1/2024

Wednesday, March 27, 2024 1:38 PM



Andrews, Edward S <edward.s.andrews@wv.gov>

RE: Empire Discussion

1 message

Wood, Katie <katie.wood@tetrattech.com>
To: "Andrews, Edward S" <edward.s.andrews@wv.gov>
Cc: Farley Wood <fwood@empirede.com>

Fri, Mar 1, 2024 at 10:23 AM

Ed,

We have some answers to your questions below in red. Please let me know if you would like to discuss.

Thanks,

Katie

From: Andrews, Edward S <edward.s.andrews@wv.gov>
Sent: Friday, February 23, 2024 2:24 PM
To: Wood, Katie <katie.wood@tetrattech.com>; Farley Wood <fwood@empirede.com>
Subject: Re: Empire Discussion

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Using Chlorine gas and water reaction to produce HCl also produces a by-product of HOCl (hypochlorous acid). <https://www.bing.com/search?q=chlorine+gas+water+reaction&q=UT&pq=chlorine+gas+water+reaction&sc=10-27&cvid=1D306673308642F9BCEE5D950B9BFB08&FORM=QBRE&sp=1&ghc=1&lq=0> The process does not produce any Chlorine gas, it produces Chloride gas so no HOCL is produced. I have attached a paper provided by Technotherm for further information

Question is: Will Empire separate hypochlorous acid from the HCl or send it out as it? As per above no HOCL will be produced

Will the emissions of HCl go through a separate release point than the RTO stack? If it is a separate stack, I need the stack id and stack parameters of this point. No all emissions will go through the RTO

Also, I will need the calculations to support your emission estimate of HCl and concentration of HCl in the effluent release to the atmosphere from the production/storage/loading out of HCl. Calculations attached, these were included in the last submittal

Ed

On Thu, Feb 22, 2024 at 1:00 PM Wood, Katie <katie.wood@tetrattech.com> wrote:

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KINETIC STUDY OF THERMAL DE-CHLORINATION OF PVC-CONTAINING WASTE

A. Castro^{1,2*}, C. Carneiro¹, C. Vilarinho², D. Soares², C. Maçães², C. Sousa² and F. Castro²

¹ CVR - Centre for Waste Valorization, Guimarães, Portugal

² CT2M - Centre for Mechanical and Materials Technology, Mechanical Engineering Department, University of Minho, Guimarães, Portugal

³ Endutex - textile coatings, SA

* acastro@cvsalduca.pt

ABSTRACT

With the increasing of plastics content in solid waste, both municipal and industrial, also increases the interest in its use as an energy source.

Some of these wastes are an important potential source of energy and might be valorized using the pyrolysis or gasification processes. However, the presence of high chlorine contents in its composition prevents its management by a thermal process, as consequence of toxic compounds production and their release to the atmosphere.

The present work assess a possible process for treating PVC-containing wastes in an environmentally friendly way. It is based on the effective de-chlorination of PVC-containing wastes through a pyrolysis process at low temperature before the carbonaceous residue from PVC-containing wastes being subject to a subsequent thermal treatment for energetic valorization.

Keywords: Pyrolysis, thermal degradation, PVC-containing waste, energy valorization;

INTRODUCTION

The presence of organic compounds on wastes, especially plastics, is considered an important source of energy. However, most of these plastics contain polyvinyl chloride (PVC), causing recycling problems when it is considered a thermal valorization process for its treatment [1], preventing the use of these residues on these processes, which main goal is the energy recovery [2,3]. A possible solution is to remove the chlorine from PVC-containing waste through a pyrolysis process before being subjected to a thermal treatment, for energetic valorization.

Pyrolysis is one of the applied techniques for energetic valorization and is defined as a process of irreversible chemical modification of compounds under the action of heat and in the absence of oxygen, causing thermal degradation [1]. The reaction involved in this process is endothermic and the characteristics of the obtained products are function of the waste composition and of several operating factors, such as the temperature, pressure and residence time in the pyrolysis reactor.

The pyrolysis process is considered by several authors [2 - 8] as a possible technique for the

operating factors, such as the temperature, pressure and residence time in the pyrolysis reactor.

The pyrolysis process is considered by several authors [2 - 8] as a possible technique for the energy recovery from PVC-containing wastes, through the thermal degradation of the chlorine

molecule. PVC pyrolysis involves significant cross-linked reactions with the formation of polyaromatic structures (possibly chlorinated) and a carbonaceous residue (char) [9]. Thus, it is possible to break down this molecule, allowing the chlorine recovery as hydrochloric acid or chloride [10], with potential economic gains.

However, the presence of poly(vinyl chloride) in wastes composition confines their management by thermal valorization processes as consequence of environmental problems and corrosion of the equipment. In fact, high levels of chlorine in wastes composition are responsible for the formation of hydrochloric acid, chlorine gas and dioxins [11]. Therefore, a previous thermal treatment by a pyrolysis process to remove the chlorine from PVC-containing wastes will be a suitable step if done prior to an energy recovery process to produce a synthesis gas.

Considering thermogravimetric analysis, it is assumed that the degradation of PVC occurs between 200 and 400 °C [2, 12]. At 250 °C, the decomposition of PVC has already been initiated, reaching a maximum at approximately 300 °C. At 350 °C the amount of chlorine present in PVC waste is less than 0.1%, which means that at this temperature, 99.9% of the whole chlorine has already been released [2]. At the end of the process of chlorine removal, a residual amount of chlorine remains on the waste [9].

The C-Cl bonds in the structure of PVC have a relatively lower binding energy than the C-C and C-H bonds, which justifies that the bonds of chlorine are the first to be broken, thus starting the thermal degradation of PVC. The de-chlorination of PVC is a free radicals chain reaction therefore requiring low activation energy to start, occurring at low temperatures [4, 7].

De-chlorination of PVC wastes is a mandatory step for any treatment process, able to recover energy from these wastes. In fact, from the decomposition of PVC, one polymeric fraction can be obtained with high energetic value:



As a matter of fact, products from the decomposition will be of the type C_nH_m .

This work aims the contemplation of new valorization processes and use of PVC-containing wastes. For the PVC molecule de-chlorination, tests were performed at low temperature pyrolysis and subsequent gasification of the remaining fraction in order to produce a synthesis gas with high energetic potential.

EXPERIMENTAL WORK

In this work, the kinetics of thermal de-chlorination has been studied, by simultaneous DTA/TGA determinations, under inert atmosphere. With all the experimental data obtained a multivariate

In this work, the kinetics of thermal de-chlorination has been studied, by simultaneous DTA/TGA determinations, under inert atmosphere. With all the experimental data obtained a multivariate regression of $\ln(r)$ has been performed in function of $1/T$ and $\ln([HCl])$. The kinetic model has been calculated just for points where temperature was lower than 340 °C, and the obtained model is:

$$\ln r = 31,3 - \frac{16130}{T} + 1,020 \ln([HCl]) \quad \text{with } r^2 = 0,9912$$

This allows considering that reaction as a first order one with activation energy of 133800 J/mol \pm 760 J/mol.

For the kinetic study, a DTA/TGA (SDT 2960 from TA Instruments) testing at different temperatures has been carried out in order to determine the relationship between the rate of PVC de-chlorination and the temperature of the thermal treatment, under an inert atmosphere. It was used a commercial pure PVC powder with the chemical formula C_2H_3Cl , in which 56,7% is chlorine. The reference is VICIR S 960 and it is a vinyl chloride homopolymer produced by a suspension polymerization process.

Experiments have conducted up to 5 different maximum temperatures: 250, 275, 300, 325 and 400 °C, with a heating rate of 10 °C per minute until the desired temperature is reached. After reaching this temperature, a stage has been done during 360 minutes. Heat flux (weight corrected heat flow in W/g) and weight of sample, has been continuously recorded.

DTA/TGA testing performed indicates that the temperature of 340 °C enables the removal of 88 % of the chlorine present in the PVC material. The resulting de-chlorinated fraction, carbonaceous residue, has also been characterized and it is mainly constituted by carbon. This carbonaceous residue was testing up to 500°C in DTA/DTA and was verified that the combustion reaction of the carbonaceous material is complete at 490°C demonstrating potential as a fuel source to a following gasification in order to produce a synthesis gas with high energetic potential.

To characterize the sample of PVC used and the carbonaceous residue formed, it has been used an TruSpec Elemental Determinator, model TruSpec CHN, of Leco with a burn time of 452 seconds and an Philips Analytical sequential X-ray fluorescence (XRF) Spectrometer model X'Unique II.

Table 1. Comparison between the chemical composition (in wt%) of PVC sample used and the carbonaceous residue obtained from pyrolysis at 340 °C.

	PVC (initial sample)	PVC (Carbonaceous residue)
Carbon	38,4	89

	Carbon (wt %)	Chlorine (wt %)
Carbon	38,4	89
Hydrogen	4,9	7
Chlorine	58,7	0,07

Through table 1, we are able to conclude that the de-chlorinated fraction obtained at 340°C is mainly constituted by carbon presenting residual chlorine content, 0,07 %.

Tests were performed in the laboratory and pilot scale, where the variables temperature, pressure and residence time inside the reactor were studied, as well as its influence on the reaction products obtained.

The pilot plant consists in a reactor where the pyrolysis occurs, with a stainless steel body heated by electrical resistance and a column of water where the gas is bubbled, as exemplified in figure 1. Measuring instruments such as thermocouples and pressure gauges are used to control the conditions (temperature and pressure) inside the reactor.

The fixation of the released chlorine is obtained by water absorption, forming HCl (hydrochloric acid), CaCl₂ (calcium chloride) and also NaCl (sodium chloride), when the aqueous solution, containing CaO (calcium oxide) or NaOH (sodium hydroxide), respectively.

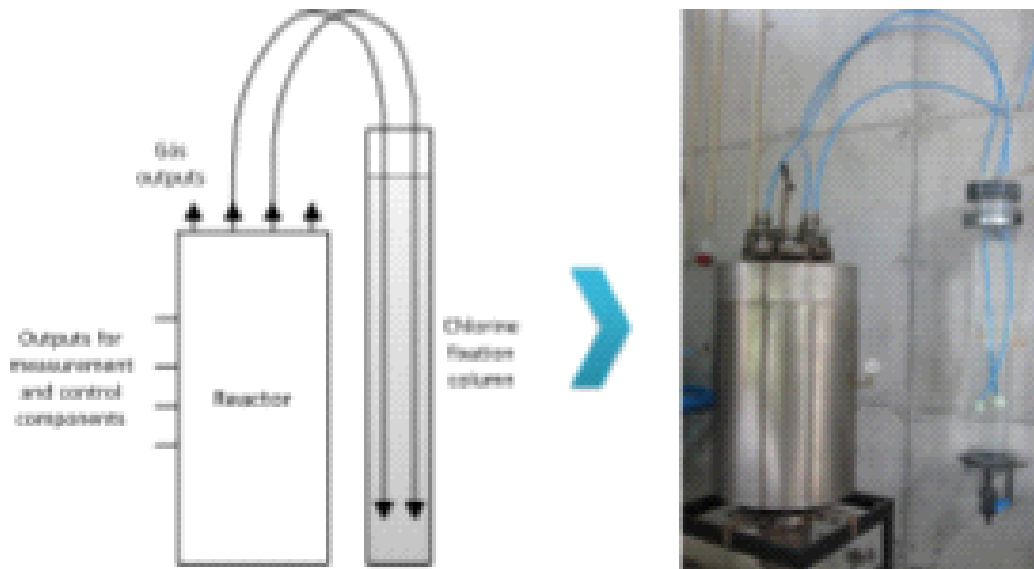


Figure 1: Scheme of pilot scale used for the tests.

Initially, a slight vacuum is created to remove the oxygen inside the reactor, then the test is initiated and divided into two stages. The first stage is a low-temperature pyrolysis or carbonization, where the de-chlorination of the PVC-containing waste reaction occurs, the released chlorine is recovered in the column by the absorption of hydrochloric acid, sodium chloride or sodium hydroxide. After this first

initially, a slight vacuum is created to remove the oxygen inside the reactor, then the test is initiated and divided into two stages. The first stage is a low-temperature pyrolysis or carbonization, where the de-chlorination of the PVC-containing waste reaction occurs, the released chlorine is recovered in the column in the form of hydrochloric acid, sodium chloride or calcium chloride. After the first stage is completed, second stage, takes place, then the carbonaceous residue (without chlorine) resulting from the first stage, is energetically valorized, at temperatures above 550 ° C.

The residence time in the reactor was tested by the pressure differences, viewed through the pressure gauge in the reactor. Thus, the reaction starts in vacuum and is assumed as completed when the pressure reaches zero, i.e. atmospheric pressure. During the reaction, when the pressure is 0,5 bar above atmospheric pressure, it is enough for the syngas formed inside the reactor can bubble in column. After all the gas is released, then the pressure drops to zero on the gauge, i.e., atmospheric pressure, thus giving the information that the reaction is complete.

All materials used in building a pilot plant must be well chosen, because of corrosion of materials and isolation. The absence of leakage or entry of gases must also be controlled, since the produced gases are toxic and cannot leak to the atmosphere, and also because as pyrolysis is a process that must take place in anoxic environment, thus it should be affected by any entry of oxidizing agents.

The main reaction product is a synthesis gas for burning to produce heat.

CONCLUSIONS

In this work, the kinetics of the reaction of thermal decomposition of PVC were studied, leading to the development of a kinetic model, with the expression $\ln r = 31,3 - 16100/T + 1,020 \ln C \text{ (HCl)}$. This model was obtained for the decomposition temperatures lower than 340 °C, in which almost all chlorine is removed from the pure PVC through the chemical reaction described, with an activation energy of 133800 J/mol, value very close to the one obtained by others researchers [4].

The kinetic model was verified in laboratorial trials, and it was observed a reduction of 58 % of the chlorine contained in PVC, making it suitable to be used in a recovery process to obtain a synthesis gas.

During the pyrolysis treatment, released chlorine can be fixed in the form of aqueous solution of hydrochloric acid, calcium chloride or sodium chloride. This process shall constitute an attractive route, envisaging environmental benefits, thereby avoiding deleterious effects of toxic gas emissions.

In this study, we propose a methodology to remove chlorine from PVC-containing wastes allowing the valorization of the chlorine-free remaining fraction. A double benefit can thus be achieved as it not only saves the cost of landfilling but also produces an value added syngas.

It is concluded that for PVC-containing waste, the solution can pass through a full treatment

not only saves the cost of landfilling but also produces an value added syngas.

It is concluded that for PVC-containing waste, the solution can pass through a full treatment consisting of two phases. Where the first is to remove the chlorine from the PVC molecule and the second is to valorize the remaining fraction.

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HCI Production Questions 3/1/2024

Wednesday, March 6, 2024 1:48 PM



Andrews, Edward S <edward.s.andrews@wv.gov>

RE: Empire Discussion

1 message

Wood, Katie <katie.wood@tetrattech.com>
To: "Andrews, Edward S" <edward.s.andrews@wv.gov>
Cc: Farley Wood <fwood@empirede.com>

Fri, Mar 1, 2024 at 10:23 AM

Ed,

We have some answers to your questions below in red. Please let me know if you would like to discuss.

Thanks,

Katie

From: Andrews, Edward S <edward.s.andrews@wv.gov>
Sent: Friday, February 23, 2024 2:24 PM
To: Wood, Katie <katie.wood@tetrattech.com>; Farley Wood <fwood@empirede.com>
Subject: Re: Empire Discussion

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Using Chlorine gas and water reaction to produce HCl also produces a by-product of HOCl (hypochlorous acid). <https://www.bing.com/search?q=chlorine+gas+water+reaction&q=UT&pq=chlorine+gas+water+reaction&sc=10-27&cvid=1D306673308642F9BCEE5D950B9BFB08&FORM=QBRE&sp=1&ghc=1&lq=0> **The process does not produce any Chlorine gas, it produces Chloride gas so no HOCL is produced. I have attached a paper provided by Technotherm for futher information**

Question is: Will Empire separate hypochlorous acid from the HCl or send it out as it? **As per above no HOCL will be produced**

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Ed

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Edward Andrews, P.E.

Engineer

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A. Castro^{1,2*}, C. Carneiro¹, C. Vilarinho², D. Soares², C. Maçães², C. Sousa² and F. Castro²

¹ CVR - Centre for Waste Valorization, Guimarães, Portugal

² CT2M - Centre for Mechanical and Materials Technologies, Mechanical Engineering Department, University of Minho, Guimarães, Portugal

³ Endutex - textile coatings, SA

* acastro@cvrnestduca.pt

ABSTRACT

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Some of these wastes are an important potential source of energy and might be valorized using the pyrolysis or gasification processes. However, the presence of high chlorine contents in its composition prevents its management by a thermal process, as consequence of toxic compounds production and their release to the atmosphere.

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molecule. PVC pyrolysis involves significant cross-linked reactions with the formation of polyaromatic structures (possibly chlorinated) and a carbonaceous residue (char) [9]. Thus, it is possible to break down this molecule, allowing the chlorine recovery as hydrochloric acid or chloride [10], with potential economic gains.

However, the presence of poly(vinyl chloride) in wastes composition confines their management by thermal valorization processes as consequence of environmental problems and corrosion of the equipment. In fact, high levels of chlorine in wastes composition are responsible for the formation of hydrochloric acid, chlorine gas and dioxins [11]. Therefore, a previous thermal treatment by a pyrolysis process to remove the chlorine from PVC-containing wastes will be a suitable step if done prior to an energy recovery process to produce a synthesis gas.

Considering thermogravimetric analysis, it is assumed that the degradation of PVC occurs between 200 and 400 °C [2, 12]. At 250 °C, the decomposition of PVC has already been initiated, reaching a maximum at approximately 300 °C. At 350 °C the amount of chlorine present in PVC waste is less than 0.1%, which means that at this temperature, 99.9% of the whole chlorine has already been released [2]. At the end of the process of chlorine removal, a residual amount of chlorine remains on the waste [9].

The C-Cl bonds in the structure of PVC have a relatively lower binding energy than the C-C and C-H bonds, which justifies that the bonds of chlorine are the first to be broken, thus starting the thermal degradation of PVC. The de-chlorination of PVC is a free radicals chain reaction therefore requiring low activation energy to start, occurring at low temperatures [4, 7].

De-chlorination of PVC wastes is a mandatory step for any treatment process, able to recover energy from these wastes. In fact, from the decomposition of PVC, one polymeric fraction can be obtained with high energetic value:



As a matter of fact, products from the decomposition will be of the type C_nH_m .

This work aims the contemplation of new valorization processes and use of PVC-containing wastes. For the PVC molecule de-chlorination, tests were performed at low temperature pyrolysis and subsequent gasification of the remaining fraction in order to produce a synthesis gas with high energetic potential.

EXPERIMENTAL WORK

In this work, the kinetics of thermal de-chlorination has been studied, by simultaneous DTA/TGA determinations, under inert atmosphere. With all the experimental data obtained a multivariate regression of $\ln(-r)$ has been performed in function of $1/T$ and $\ln([HC\ell])$. The kinetic model has been calculated just for points where temperature was lower than 340 °C, and the obtained model is:

regression of $\ln(r)$ has been performed in function of $1/T$ and $\ln([HCl])$. The kinetic model has been calculated just for points where temperature was lower than 340 °C, and the obtained model is:

$$\ln r = 31,3 - \frac{16139}{T} + 1,029 \ln[HCl] \quad \text{with } r^2 = 0,8912$$

This allows considering that reaction as a first order one with activation energy of $163800 \text{ J/mol} \pm 760 \text{ J/mol}$.

For the kinetic study, a DTA/TGA (SDT 2960 from TA Instruments) testing at different temperatures has been carried out in order to determine the relationship between the rate of PVC de-chlorination and the temperature of the thermal treatment, under an inert atmosphere. It was used a commercial pure PVC powder with the chemical formula C_2H_2Cl , in which 58,7% is chlorine. The reference is VICIR S 950 and it is a vinyl chloride homopolymer produced by a suspension polymerization process.

Experiments have conducted up to 5 different maximum temperatures: 250, 275, 300, 325 and 400 °C, with a heating rate of 10 °C per minute until the desired temperature is reached. After reaching this temperature, a stage has been done during 360 minutes. Heat flux (weight corrected heat flow in W/g) and weight of sample, has been continuously recorded.

DTA/TGA testing performed indicates that the temperature of 340 °C enables the removal of 88 % of the chlorine present in the PVC material. The resulting de-chlorinated fraction, carbonaceous residue, has also been characterized and it is mainly constituted by carbon. This carbonaceous residue was testing up to 500°C in DTA/DTA and was verified that the combustion reaction of the carbonaceous material is complete at 400°C demonstrating potential as a fuel source to a following gasification in order to produce a synthesis gas with high energetic potential.

To characterize the sample of PVC used and the carbonaceous residue formed, it has been used an TruSpec Elemental Determinator, model TruSpec CHN, of Leco with a burn time of 452 seconds and an Philips Analytical sequential X-ray fluorescence (XRF) Spectrometer model X'Unique II.

Table 1. Comparison between the chemical composition (in wt%) of PVC sample used and the carbonaceous residue obtained from pyrolysis at 340 °C.

	PVC (initial sample)	PVC (Carbonaceous residue)
Carbon	38,4	88
Hydrogen	4,9	7
Chlorine	58,7	0,07

Chlorine	58,7	0,07
----------	------	------

Through table 1, we are able to conclude that the de-chlorinated fraction obtained at 340°C is mainly constituted by carbon presenting residual chlorine content, 0,07 %.

Tests were performed in the laboratory and pilot scale, where the variables temperature, pressure and residence time inside the reactor were studied, as well as its influence on the reaction products obtained.

The pilot plant consists in a reactor where the pyrolysis occurs, with a stainless steel body heated by electrical resistance and a column of water where the gas is bubbled, as exemplified in figure 1. Measuring instruments such as thermocouples and pressure gauges are used to control the conditions (temperature and pressure) inside the reactor.

The fixation of the released chlorine is obtained by water absorption, forming HCl (hydrochloric acid), CaCl₂ (calcium chloride) and also NaCl (sodium chloride), when the aqueous solution, containing CaO (calcium oxide) or NaOH (sodium hydroxide), respectively.

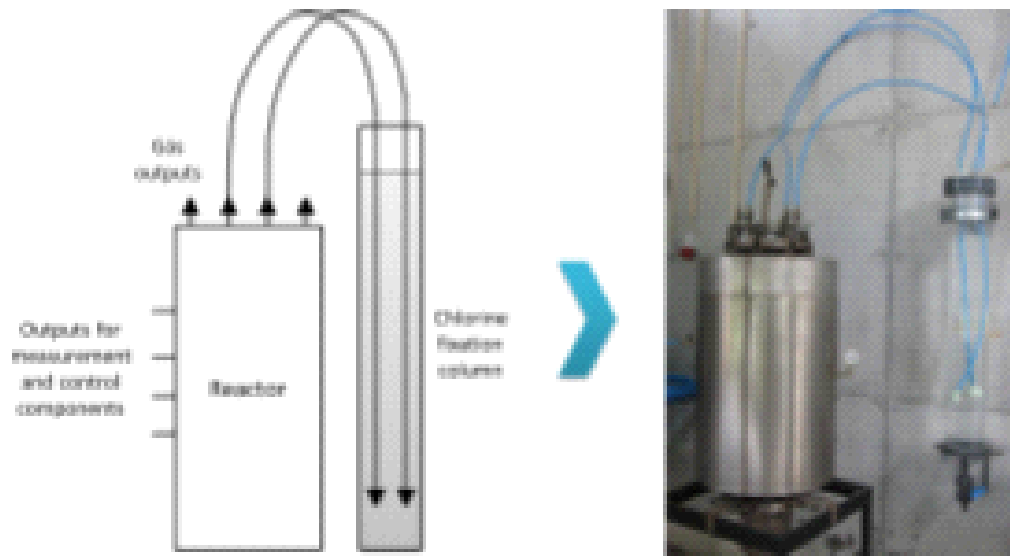


Figure 1: Scheme of pilot scale used for the tests.

Initially, a slight vacuum is created to remove the oxygen inside the reactor, then the test is initiated and divided into two-stages. The first stage is a low-temperature pyrolysis or carbonization, where the de-chlorination of the PVC-containing waste reaction occurs, the released chlorine is recovered in the column in the form of hydrochloric acid, sodium chloride or calcium chloride. After the first stage is completed, second stage, takes place, then the carbonaceous residue (without chlorine) resulting from the first stage, is energetically valorized, at temperatures above 550 °C.

The residence time in the reactor was tested by the pressure differences, viewed through the pressure gauge in the reactor. Thus, the reaction starts in vacuum and is assumed as completed when the pressure reaches zero, i.e. atmospheric pressure. During the reaction, when the pressure

The residence time in the reactor was tested by the pressure differences, viewed through the pressure gauge in the reactor. Thus, the reaction starts in vacuum and is assumed as completed when the pressure reaches zero, i.e. atmospheric pressure. During the reaction, when the pressure is 0,5 bar above atmospheric pressure, it is enough for the syngas formed inside the reactor can bubble in column. After all the gas is released, then the pressure drops to zero on the gauge, i.e., atmospheric pressure, thus giving the information that the reaction is complete.

All materials used in building a pilot plant must be well chosen, because of corrosion of materials and isolation. The absence of leakage or entry of gases must also be controlled, since the produced gases are toxic and cannot leak to the atmosphere, and also because as pyrolysis is a process that must take place in anoxic environment, thus it should be affected by any entry of oxidizing agents.

The main reaction product is a synthesis gas for burning to produce heat.

CONCLUSIONS

In this work, the kinetics of the reaction of thermal decomposition of PVC were studied, leading to the development of a kinetic model, with the expression $\ln r = 31,3 - 16100/T + 1,020 \ln C$ (HC). This model was obtained for the decomposition temperatures lower than 340 °C, in which almost all chlorine is removed from the pure PVC through the chemical reaction described, with an activation energy of 133800 J/mol, value very close to the one obtained by others researchers [4].

The kinetic model was verified in laboratorial trials, and it was observed a reduction of 88 % of the chlorine contained in PVC, making it suitable to be used in a recovery process to obtain a synthesis gas.

During the pyrolysis treatment, released chlorine can be fixed in the form of aqueous solution of hydrochloric acid, calcium chloride or sodium chloride. This process shall constitute an attractive route, envisaging environmental benefits, thereby avoiding deleterious effects of toxic gas emissions.

In this study, we propose a methodology to remove chlorine from PVC-containing wastes allowing the valorization of the chlorine-free remaining fraction. A double benefit can thus be achieved as it not only saves the cost of landfilling but also produces an value added syngas.

It is concluded that for PVC-containing waste, the solution can pass through a full treatment consisting of two phases. Where the first is to remove the chlorine from the PVC molecule and the second is to valorize the remaining fraction.

References

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References

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HCl Spec

Friday, February 23, 2024 4:18 PM

Product Specifications

luriatic Acid, 20 DEG Baume (HCL 32)

Item	Specification
	APHA
Color, Max	15.0
Degrees Baume DEG Baume @60F	20.0-20.8
	Parts Per Million
Arsenic, Max (AS)	0.1
Bromide, Max (BR)	50.0
Calcium, Max (CA)	2.0
Free Chlorine, Max (CL2)	3.0
Fluoride, Max (F)	2.0
Iron, Max (FE)	0.5
Non-Volatile Residue, Max (NVR)	15.0
Organics, Max	1.0
Lead, Max (PB)	0.2
Sulfate, Max	10.0
	Percent by Weight
Hydrogen Chloride	31.5-32.9

Approved 07.12.1996

Response to HCl questions 2/21/2024

Wednesday, February 21, 2024 8:49 AM



Andrews, Edward S <edward.s.andrews@wv.gov>

FW: Empire's Revised Modification App

1 message

Wood, Katie <katie.wood@tetrattech.com>
To: Edward Andrews <edward.s.andrews@wv.gov>

Wed, Feb 21, 2024 at 8:33 AM

Ed,

Please see responses below for the HCL for the process. I will follow up on the plastic feedstock as fuel here shortly.

Thanks,

Katie

From: Farley R. Wood, P.E. <fwood@empirede.com>
Sent: Tuesday, February 20, 2024 2:14 PM
To: Wood, Katie <katie.wood@tetrattech.com>
Subject: RE: Empire's Revised Modification App

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Katie,

Please see below:



Farley R. Wood, P.E.
Vice President of Engineering

Main Office (304) 935-5851
Mobile: (304) 650-2023
Teams: [Click Here](#)

fwood@empirede.com
www.empirediversifiedenergy.com

From: Wood, Katie <katie.wood@tetrattech.com>
Sent: Tuesday, February 20, 2024 11:36 AM

To: Farley R. Wood, P.E. <fwood@empirede.com>
Subject: FW: Empire's Revised Modification App

You don't often get email from katie.wood@tetrattech.com. [Learn why this is important](#)

From: Andrews, Edward S <edward.s.andrews@wv.gov>
Sent: Tuesday, February 20, 2024 10:16 AM
To: Wood, Katie <katie.wood@tetrattech.com>
Subject: Re: Empire's Revised Modification App

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Katie,

I will need additional information about the HCl production/storage/loading out rack to develop the appropriate permit requirements.

How will the HCl be produced and at what concentration? **Absorption and cooling of chlorine gas into demineralized water in a spray tower. The target concentration of hydrochloric acid is 31.45% (20° Baume').**

How much HCl be stored on site (identify the tanks and dimensions of the tanks)? **We will have two 10,305 gallon tanks for HCl storage. One for in spec HCl and one for out of spec HCL. Tank dimensions are 11' 11" diameter by 14' high.**

The goal is to make in spec product, so the out of spec tank will hopefully be nearly empty most of the time. We will have a 1,500 gallon production tank (7'2" diameter by 5' 11" high) witch will be where quality analysis samples are regularly taken.

Based on the analysis the HCl will be routed to the in spec or out of spec tanks. HCl will be removed from the tanks on a daily basis by FSTI, an onsite tenant of the Port. Production and shipments will be roughly equal to minimize stored product.

Is the proposed scrubber going to be used to control the storage and loadout racks or just the loading rack? **Both**

Will the pyrolysis units need to process feedstock that contains PVC (polyvinyl chloride) type of plastic material to produce HCl? **Yes, we have the ability to not produce HCl by keeping PVC out of the feed material.**

Also, the reference to House Bill 4048 does not help me justify why the plastic feedstock should be treated as fuel?

Specifically, I need sufficient information from the application to indicate the plastic feedstock is not considered waste and therefore the facility (pyrolysis units) are not subject to Subpart AAAA, CCCC, and Subpart EEEE because the feedstock material is not waste.

Ed

On Wed, Feb 7, 2024 at 8:55 AM Wood, Katie <katie.wood@tetrattech.com> wrote:


Thanks Ed,

Let me dive into this a little more and talk to them and I will reach out to discuss.

Thanks,

Katie

From: Andrews, Edward S <edward.s.andrews@wv.gov>
Sent: Tuesday, February 6, 2024 9:08 AM
To: Wood, Katie <katie.wood@tetrattech.com>
Subject: Empire's Revised Modification App

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Katie,

It is still unclear to me how HCl is going to be generated and separate from the pyrolysis process.

I am not sure if I can just accept WV House Bill 4048 as proof that the material being processed is non-waste.

I would like to discuss these issues further in the near future.

Ed

--

Edward Andrews, P.E.

Engineer

WVDEP/Division of Air Quality

304-926-0499 Ext 41244

[601 57th Street, SE](#)

[Charleston, WV 25304](#)

--

Edward Andrews, P.E.

Engineer

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[601 57th Street, SE](#)

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Notary Public 2/5/24

Monday, February 5, 2024 1:43 PM



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Commission:	HH 186700	Email:	P.BAEZ820@GMAIL.C
RON Issue Date:	6/7/2023	Phone:	407-491-3362
Expire Date:	10/14/2025		
Service Provider			
Name	Start Date	End Date	
NOTARIZE, INC.	Jun 01, 2023		
Secure Repository			
Name	Start Date	End Date	Preferred Contact



Questions or comments? Please contact us

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R.A. Gray Building
500 South Bronough St
Tallahassee FL, 32399-0250
850-245-6500



Andrew, Edward S <edward.s.andrews@va.gov>

RE: Incomplete App Email for Permit App R13-3555A

1 message
Tue, Jan 23, 2024 at 4:28 PM

From: **Wood, Katie** <katie.wood@va.gov>
To: "Andrew, Edward S" <edward.s.andrews@va.gov>
Cc: Beverly D Osborne <beverly.d.osborne@va.gov>; Brian S Tophaback <brian.s.tophaback@va.gov>; Eric Blend <eric.blend@va.gov>; Kenneth Brown <kenneth.brown@va.gov>; Tanya Wood <tanya.wood@va.gov>

Hi,
Please find attached the revised notification application for Empire Green Generation. Please find the response to your comments below in blue. Please don't hesitate to reach out if you have any questions.
Thank you.

Katie Wood | Environmental Scientist
Direct: +1 (703) 296-9662 | Mobile: +1 (204) 598-9999 | katie.wood@va.gov
Contract: Public Policy, (please only email change)
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From: Andrew, Edward S <edward.s.andrews@va.gov>
Sent: Wednesday, December 20, 2023 2:34 PM
To: Edward Brown <edward.brown@va.gov>; Tanya Wood <tanya.wood@va.gov>; Wood, Katie <katie.wood@va.gov>
Cc: Beverly D Osborne <beverly.d.osborne@va.gov>; Brian S Tophaback <brian.s.tophaback@va.gov>; Eric Blend <eric.blend@va.gov>
Subject: Incomplete App Email for Permit App R13-3555A

CAUTION: This email originated from an external sender. Verify the source before opening links or attachments.

RE: Application Status: Incomplete
Empire Green Generation
Permit Application No. R13-3555A
Plant ID No. 889-00141

Hi Brian:

Your application for a modification permit for a plastic recycling by pyrolysis facility was received by this Division on December 1, 2023, and assigned to the writer for review. Upon initial review of said application, it has been determined that the application as submitted is incomplete based on the following items:

1. Affidavit of Publication of Class I Legal Ad. Included in pdf
2. Discussion of the proposed physical changes, if applicable, and/or change to the method of operation. Specifically, this discussion needs to either outline or go into detail regarding proposed regulatory changes to the permit, if applicable; type and amount/weight of the plastic feedstock going to be processed by the facility; any processing/pre-treatment being going to be conducted on the plastic prior to being introduced into the pyrolysis vessel; if there processing is going to occur off-site, it will need to be identified and discussed; a discussion how the facility will switch back and forth in processing medical waste and plastic feedstock, such a discussion why processing the proposed plastic feedstock is not viewed as waste disposal through incineration or the context of the Clean Air Act and how the criteria in a facility within the requirements and provisions of an EPA 241 - SOLID WASTES UNDER FEDERAL OR STATEMENTS IN COMBINATION UNITS. The only physical change to the in feedstock, there is no change in the process. Additional information about the pre-processing handling was added to section C process description. There will be no switching back and forth of feedstock, the facility will only be recycling plastics. A discussion of the regulatory requirements has been added to section D regulatory discussion.
3. The plan (Attachment E) needs to be updated to identify emission units and emission points. Emission units include in pdf plan
4. Attachment J needs to be complete for each emission point. No additional emission points have been added
5. Attachment K and L need to be completed. The potential for leaking equipment (e.g., valves, pumps, compressors, emissions, pressure relief devices) needs to be quantified and documented in their respective attachments. Attachments completed and incorporated in PDF
6. Each of the subject pages that contain confidential business information (CBI) needs to be marked "redacted copy - claim of confidentiality" in accordance with 29 USC 171.4. Redacted pages included in confiders

In addressing items 2 through 6 need to be reflective within the subject application as a single PDF file.

The emission estimates appear to be identical to the emission estimates for processing medical waste with the same process unit. Please review these estimates and review as necessary within permit unit. Any changes will not change given the change in feedstock. The estimates are not expected to change, the majority of the medical waste feedstock was anticipated to be plastics. There is not change to the process, just feedstock so the emission estimates remain the same.

Please advise the status of this application in writing by no later than January 18, 2024. Application review will not commence until the application has been deemed to be technically complete. Failure to respond to this request in a timely manner may result in the denial of the application.


Should you have any questions, please contact Ed Andrews at (304) 626-9899 ext. 41244 or reply to this email.

Edward Andrews, P.E.
Engineer
WDEP/Division of Air Quality
304-626-9899 Ext 41244
601 57th Street, SE
Chickasha, WV 25304

R13-3555 Modification_Application_EG0 - Redacted_Final Rev.pdf
2/24/24



NSR (45CSR13) APPLICATION FORM

 WEST VIRGINIA DEPARTMENT OF ENVIRONMENTAL PROTECTION DIVISION OF AIR QUALITY 401 2 nd Floor, SE Charleston, WV 25304 (304) 526-6175 www.dep.wv.gov		APPLICATION FOR NSR PERMIT AND TITLE V PERMIT REVISION (OPTIONAL)	
PLEASE CHECK ALL THAT APPLY TO NSR (45CSR13) OF ANOVA: <input type="checkbox"/> CONSTRUCTION <input checked="" type="checkbox"/> MODIFICATION <input type="checkbox"/> RELOCATION <input type="checkbox"/> CLASS I ADMINISTRATIVE UPDATE <input type="checkbox"/> TECHNOLOGY <input type="checkbox"/> AFTER-FIELD FACT		PLEASE CHECK TYPE OF 45CSR13 (TITLE V) REVISION (IF ANY): <input type="checkbox"/> ADMINISTRATIVE INDEPENDENT <input type="checkbox"/> MINOR MODIFICATION <input type="checkbox"/> SIGNIFICANT MODIFICATION IF ONLY SIGN. MOD. IS CHECKED, INCLUDE TITLE V REVISION INFORMATION IN ATTACHMENT 1 TO THIS APPLICATION	
FOR TITLE V FACILITIES ONLY: Please refer to "Title V Facility Checklist" in order to determine what Title V Facility action (Appendix A, "Title V Permit Revision Flowchart") and ability to operate with the changes requested in this Permit Application			
Section I. General			
1. Name of applicant (as registered with the WV Secretary of State's Office): Crown Glass Services LLC		2. Facility Employer ID No. (FID#): 00-0121225	
3. Name of facility (if different from above): Crown Glass Operations		4. The applicant is the: <input type="checkbox"/> Owner <input type="checkbox"/> Operator <input checked="" type="checkbox"/> Both	
5A. Applicant's mailing address: 1600 Van Drive, Fortranova, WV 26027		5B. Facility's street address: 601 Rogers Rd, Fortranova, WV 26027	
6. Does Virginia Business Registration, in the applicant's account of the State of West Virginia? <input checked="" type="checkbox"/> Yes <input type="checkbox"/> No - If YES, provide a copy of the Certificate of Incorporation/Partnership (one page) including any name change amendments or other Business Registration Certificates as Attachment A. - If NO, provide a copy of the Certificate of Authority/Authority of L.L.C./Registration (one page) including any name change amendments or other Business Certificates as Attachment A.			
7. If applicant is a subsidiary corporation, please provide the name of parent corporation.			
8. Does the applicant own, lease, have an option to buy, or otherwise have control of the proposed site? <input checked="" type="checkbox"/> YES <input type="checkbox"/> NO - If YES, please explain: Own and Operate			
- If NO, you are not eligible for a permit for this source.			
9. Type of start or facility (select any source) to be constructed, modified, relocated, administratively updated or temporarily permitted (e.g., one production unit, primary burner, etc.). Please Recycle By Purchase.		10. North American Industry Classification System (NAICS) code for the facility: 332222	
11A. DAD Plant ID No. (for existing facilities only): 000-02141		11B. List all current 45CSR13 and 45CSR12 Title V permit numbers associated with this process (for existing facilities only): 81-0000	
All of the required forms and additional information can be found under the Permitting Section of DAD's website, or requested by phone.			

12A. For Modifications, Administrative Updates, or Temporary permits at an existing facility, please provide a location to the nearest location of the facility from the nearest state road.
- For construction or replacement permits, please provide a location to the proposed new site location from the nearest state road. Include a WVI as Attachment B.

Turn in of 11A-2 and 12A (if 12A is used). Turn right into Rogers Road (0.2 mile). Facility location will be on the right.

12B. New site address (if applicable):	12C. Nearest city or town: Fortranova	12D. County: Boone
12E. UTM Northing (NAD 83): 40 038880	12F. UTM Easting (NAD 83): 465 808408	12G. UTM Zone: 17T

13. Shall I receive the proposed changes at the facility?
The facility will be required to receive the proposed changes of process (make as original) permitted:
- If you are an affected third party applicant, provide the date upon which the proposed change will occur: 10/2010

14. Provide the date of anticipated installation or change: 10/2010
- If you are an affected third party applicant, provide the date upon which the proposed change will occur: 10/2010

14C. Provide a description of the process, installation of changes to and start-up of each of the units proposed in the permit application as Attachment C (if more than one unit is involved).

15. Provide permit-required Operating Schedule of unit(s) to be installed in the application:
- Check the box (or check the boxes): Change the facility

16. Is a description or physical relocation of an existing facility involved? YES NO

17. Does Management Term, if the facility is subject to 11.2(c) of the 1990 CAA, or will become subject due to proposed changes to the facility, that are not eligible for grandfathering under the Management Term 11.2(c) of the 1990 CAA, apply? YES NO

18. Regulatory Discussion. List all Federal and State air pollution control regulations that you believe are applicable to the proposed process (if any). A list of these applicable requirements is also included in Attachment B of the application (Title V Permit Revision instructions). Discuss applicability and proposed administrative or compliance (if any). Provide the information as Attachment D.

Section II. Additional attachments and supporting documents.

19. Include a check (check to WV DEP - Division of Air Quality with the appropriate application fee per 45CFR102 and 45CFR103).

20. Include a table of Contents as the first page of your application package.

21. Provide a Title Plan, as a signed original, showing the location of the property on which the stationary permitted unit is to be located as Attachment E (State or Federal Government).
- Include the location of the nearest public roadway, if applicable, street, highway, business, residential.

22. Provide a detailed process flow diagram(s) showing all units proposed or modified with their unit, emission point and control (check as Attachment F).

23. Provide a Notice of Construction as Attachment G.
- Also describe and quantify in the permit package all changes made to the facility since the last permit renewal (if applicable).

All of the required forms and additional information can be found under the Permitting Section of DAD's website, or requested by phone.

24. Provide Material Safety Data Sheets (MSDS) for all materials processed, used or produced as Attachment K.
 Do NOT include MSDSs for any hazardous materials or any of the following:
 25. Provide the MSDSs for the following materials as Attachment L:
 26. Provide the MSDSs for the following materials as Attachment M:
 27. Provide the following information as Attachment N:
 28. Check all applicable items that apply to the facility as Attachment O:
 Air Liquid Transfer Operations New Road Structures Quarry
 Chemical Processes Hot Air Asphalts Cold Inerted Storage, Handling and Storage
 Concrete Batch Plant Inherently Hazardous Fuel Gas
 On-Off and Seal Facility Inherently Hazardous Storage Tanks
 Inerted Storage Other
 29. Provide the MSDSs for the following materials as Attachment P:
 30. Check all applicable Air Pollution Control Device (APCD) as Attachment Q:
 Absorption Systems Baghouses Flare
 Adsorption Systems Condensers Thermal Oxidizer
 Electrostatic Precipitator Other
 Other Control Agency Other Control Agency
 31. Provide all Supporting Information Calculations as Attachment R, or attach the calculations directly to the forms listed in items 21 through 23.
 32. Monitoring, Recordkeeping, Reporting and Testing: Attach process monitoring, recordkeeping, reporting and testing data in order to demonstrate compliance with the applicable emissions rules and operating parameters in this permit application. Provide the information as Attachment S:
 Permittee has a permit to be credited with a monitoring system or the applicant chooses to propose such a system. Attachment S, Part I, may not be able to accept all facilities proposed by the applicant. If none of these items are proposed, the applicant must provide such information as Attachment T.
 33. Public Notice: At the time that the application is submitted, please a Class I Large Amendment in a newspaper of general circulation in the area where the source is or will be located. See 40 CFR 192.2 through 40 CFR 192.9 and Bureau's Legal Advice/Notice to the Public. Please submit the Affidavit of Compliance as Attachment U (Increase, Add, Amend).
 34. Business Confidentiality: Check the application include confidential information (see 40 CFR 192.207).
 Yes No
 35. If this facility is a permit for a Class I Large Amendment, attach a Confidentiality Statement as Attachment V.
 36. Check all applicable Authority as Attachment W. Do not check more than one authority for the applicable authority for the application.
 Authority of Corporation or Other Business Entity Authority of Partnership
 Authority of Government Agency Authority of Limited Partnership
 Submit completed and signed Authority forms as Attachment W.
 All of the required forms and additional information can be found under the Permitting Section of BDD's website, or requested by phone.

37. Certification of Information: To verify the permit applicant's Representative Official (see 40 CFR 192.22 and 40 CFR 192.23) is an Authorized Representative and that the information provided is true and accurate, the permit applicant must complete the following:
 38. Certification of Truth, Accuracy, and Completeness
 The undersigned Authorized Representative hereby certifies that all information contained in this application and any supporting information submitted herewith, in this application, and attached based on information received from other responsible parties, is true and accurate to the best of the permit applicant's knowledge and belief, and that the information is complete and accurate. I understand that providing false information in this application is a violation of the Clean Air Act and may result in the permit applicant being held liable for the information provided. I understand that providing false information in this application is a violation of the Clean Air Act and may result in the permit applicant being held liable for the information provided. I understand that providing false information in this application is a violation of the Clean Air Act and may result in the permit applicant being held liable for the information provided.
 39. Signature: Permittee's Name DATE: 11/01/2011
 40. Printed name of signee: Permittee's Name TITLE: Chief Operating Officer
 41. E-mail: Permittee's Email TEL: Phone: 801-987-6543 SEC: FAX:
 42. Printed name of contact person (if different from above): TITLE:
 43. E-mail: TEL: Phone: SEC: FAX:

PLEASE CHECK ALL APPLICABLE ATTACHMENTS INCLUDED WITH THIS PERMIT APPLICATION:
 Attachment D: Business Confidentiality Attachment E: Air Pollution Control Device Summary
 Attachment G: MSDS for all materials processed, used or produced Attachment H: MSDS for the following materials
 Attachment I: MSDS for the following materials Attachment J: MSDS for the following materials
 Attachment K: MSDS for the following materials Attachment L: MSDS for the following materials
 Attachment M: MSDS for the following materials Attachment N: MSDS for the following materials
 Attachment O: Air Pollution Control Device Summary Attachment P: MSDS for the following materials
 Attachment Q: Air Pollution Control Device Summary Attachment R: Supporting Information Calculations
 Attachment S: Monitoring, Recordkeeping, Reporting and Testing Data Attachment T: Confidentiality Statement
 Attachment U: Affidavit of Compliance Attachment V: Confidentiality Statement
 Attachment W: Authority forms Attachment X: Confidentiality Statement
 Attachment Y: Confidentiality Statement Attachment Z: Confidentiality Statement
 Please affix an original and three copies of the completed application to the permit application and submit to the Permitting Section, at the address listed on the front page of this application. Please DO NOT permit application.

 DEPARTMENT OF ENVIRONMENTAL PROTECTION DIVISION OF AIR QUALITY 50117' South St. QUINCY, ILLINOIS 62414 PHONE: 618-243-4000 FAX: 618-243-4000 WWW.PSD.IL.GOV/DAQ		PERMIT APPLICATION FORM (PDF) FOR AGENCY USE ONLY: PLANT ID # _____ PERM # _____	
1. NAME OF APPLICANT AS REGISTERED WITH THE ILLINOIS SECRETARY OF STATE'S OFFICE: <u>Brain Corporation, LLC</u>			
2. NAME OF FACILITY (IF DIFFERENT FROM ABOVE): <u>Polymers Operations</u>			
3A. MAILING ADDRESS: <u>1400 Van Buren, Colfax, IL 62420</u>		3B. PHYSICAL ADDRESS: <u>801 Rogers Rd, Polk, Illinois 62450</u>	
3A. DIRECTIONS TO FACILITY PLEASE PROVIDE MAP AS ATTACHMENT A: <u>Turn left on 11th St. (11th St. is 1/4 mile N. of 11th St. on the right side of the road. Turn right onto Rogers Rd. (11th St. is on the right side of the road.)</u>			
3C. NEAREST ROAD: <u>11th St.</u>	3D. NEAREST CITY OR TOWN: <u>Polk, Illinois, IL</u>	3E. COUNTY: <u>Polk</u>	
3F. STRAIGHTENING ZONE: <u>40'±</u>	3G. 1/4 MILE EASTING ZONE: <u>40'±</u>	3H. LTR ZONE: <u>1/4</u>	
3I. PERSONNEL: INDICATE IF MORE INFORMATION IS REQUIRED: <input type="checkbox"/> None <input type="checkbox"/> Other <u>None</u> Name: _____ Title: _____ E-mail: _____ Phone: _____			
3J. BACKUP/PLANT ID NO. (FOR AN EXISTING FACILITY ONLY): <u>000 - 00141</u>			
3K. PLEASE USE ALL CURRENT APPLICANTS TO OBTAIN AGENCY USE TO OBTAIN A NEW AGENCY NUMBER WITH THIS PROCESS (FOR AN EXISTING FACILITY ONLY): <u>None</u>			
3L. IS THE PDF BEING SUBMITTED AS THE RESULT OF AN ENVIRONMENTAL ACTION? IF YES, PLEASE LIST: <input type="checkbox"/> YES <input type="checkbox"/> NO			
3M. TYPE OF ORIGIN SOURCE (CHECK ONE): <input checked="" type="checkbox"/> NEW SOURCE <input type="checkbox"/> ADMINISTRATIVE SOURCE <input type="checkbox"/> YES <input type="checkbox"/> NO <input type="checkbox"/> MODIFICATION <input type="checkbox"/> OTHER PLEASE EXPLAIN IN YES: _____			
3N. IS DESTRUCTION/PHYSICAL ABANDONMENT OF ALL EXISTING FACILITY PROPOSED? <input type="checkbox"/> YES <input checked="" type="checkbox"/> NO			
3O. DATE OF ANTICIPATED INSTALLATION OR CHANGE: <u>1/15/2012</u> NO. DATE OF ANTICIPATED START-UP: <u>1/15/2012</u>			
3P. PLEASE PROVIDE A DETAILED PROCESS FLOW DIAGRAM SHOWING ALL INPUTS/OUTPUTS/CONVERSION PROCESSES (SEE PERMITS SECTION OF ATTACHMENT D):			
3Q. PLEASE PROVIDE A DETAILED PROCESS DESCRIPTION AS ATTACHMENT E:			
3R. PLEASE PROVIDE MATERIAL SAFETY DATA SHEETS (MSDS) FOR ALL MATERIALS PROCESSED, USED OR PRODUCED AS ATTACHMENT K. FOR CHEMICAL PROCESSES, PLEASE PROVIDE A MSDS FOR EACH COMPONENT IDENTIFIED TO A.D.			

10. REGULATED AIR POLLUTANT EMISSIONS:
 a. FOR A NEW FACILITY, PLEASE PROVIDE PLANT WIDE EMISSIONS BASED ON THE POTENTIAL TO EMIT (PTE) FOR THE FOLLOWING AIR POLLUTANTS USING THE FOLLOWING PROCEDURE:
 b. FOR AN EXISTING FACILITY, PLEASE PROVIDE THE EMISSIONS CHANGE IN EMISSIONS BASED ON THE PTE OF ALL FACILITY CHANGES FOR THE FOLLOWING AIR POLLUTANTS
 c. FOR A GREEN FACILITY, PLEASE PROVIDE THE EMISSIONS CHANGE IN EMISSIONS BASED ON THE PTE OF ALL FACILITY CHANGES FOR THE FOLLOWING AIR POLLUTANTS
 THE DATA IS TO BE REPORTED IN TABLES A, B, C, D, E, F, G, H, I, J, K, L, M, N, O, P, Q, R, S, T, U, V, W, X, Y, Z AND IS COLLECTED BASED ON THE FOLLOWING DATA: CAPACITY OF PROCESS EQUIPMENT

POLLUTANT	MOBILE PTE (LB/DAY)	YEARLY PTE (TONS/YR)
PM ₁₀	5.7	20.0
PM _{2.5}	3.00	14.0
VOCs	5.45	24.0
CO	25.00	90.0
NO _x	5.45	24.0
SO _x	5.90	20.0
PH ₃	NA	NA
HAZ* (ACCORDANT AGENCY)	0.05	0.0
HAZ* (NON-ACCORDANT)	0.05	0.0
OTHER (ACCORDANT AGENCY)	0.0	0.0

11. PLEASE PROVIDE ALL SUPPORTING CALCULATIONS AS ATTACHMENT C
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ATTACHMENT A

Business Certificate

**City of Follansbee
CITY LICENSE**



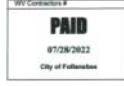
Follansbee, West Virginia

This is to certify that the undersigned, in pursuance of the authority vested in him by law has this day granted to:

Name of Establishment: **EMPIRE GREEN GENERATION LLC**
 Address: **1400 MAIN ST**
FOLLANSBEE WV 26037-1218
 Name of Owner: **FRANK ROSSO**
 Address: **401 EAST LAS OLAS BLVD SUITE 1400**
FORT LAUDERDALE FL 33301-2218

a license to engage in, conduct or operate the business of, or devices for which license tax has been assessed and paid as shown in license schedule herein.

Date Issued: **28-Feb-2022**
 Expiration Date: **30-Jun-2023**



Any automatic device licensed herein is that which is not a gambling device under city ordinance or the laws of the State of West Virginia

LICENSE No: 288

John G. McIntosh
City Manager

DISPLAY IN A CONSPICUOUS PLACE

State of West Virginia

Certificate

I, Mac Warner, Secretary of State of the State of West Virginia, hereby certify that

EMPIRE GREEN GENERATION, LLC

was duly authorized under the laws of this state to transact business in West Virginia as a foreign limited liability company on December 02, 2021.

The company is filed as an at-will company, for an indefinite period.

I further certify that the company has not been revoked or administratively dissolved by the State of West Virginia and the West Virginia Secretary of State issued a Certificate of Cancellation or Termination to the company.

Accordingly, I hereby issue this Certificate of Authorization

CERTIFICATE OF AUTHORIZATION

Validation ID: SWYJR_YA8BM

Given under my hand and the Great Seal of the State of West Virginia on this day of January 07, 2022

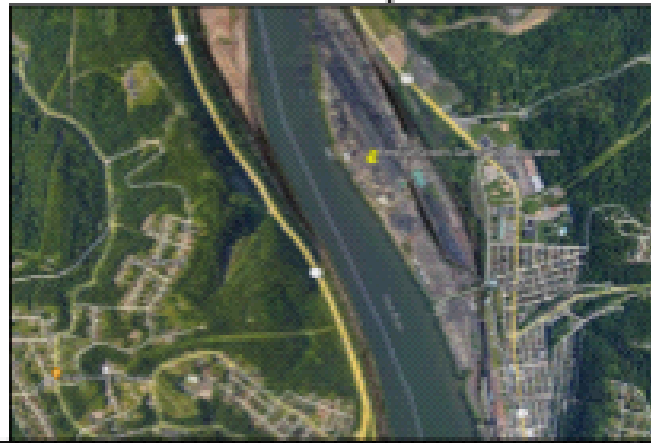


Mac Warner

Secretary of State

ATTACHMENT B

Maps



EMPIRE GREEN GENERATION, LLC
FOLLANSBEE, WY
SITE MAP AND LAYOUT

ATTACHMENT C
Installation and Startup Schedule



ATTACHMENT C: INSTALLATION AND START UP SCHEDULE

Unit	Start of installation	Approximate Start of Operations
BOG & DBO (Terminal Division and Steam Reform)	October 2022	February 2024

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ATTACHMENT D

Regulatory Discussion

15

1.1 West Virginia State Requirements

The Facility will be a minor source of emissions under the NBR Program as well as the Title V Operating Permit program under §45C-2-2. However, the potential uncontrolled emissions for the Facility will exceed the permitting thresholds of 6 pounds per hour (lb/hr) and/or 144 pounds per day (lb/day) in accordance with WVDEP §45C-2-2.24. Accordingly, Entire Green Generation, LLC is submitting this application for a minor source permit to install and operate.

In addition to regulations, state regulations that pertain to this Facility are listed in Table 1-1. Titles shown in capital letters in the table are permits, notifications, and/or reports that will be needed for construction and operation of the Facility.

Federal authority is delegated to the State of West Virginia, and all permit applications will be submitted to West Virginia Department of Environmental Protection (WVDEP). The following list of air permits is applicable to the proposed facility:

Table 1-1 West Virginia DEP Applicable Regulations

Rule	Description
45C-2-2	Control of visible and particulate emissions from stationary sources
45C-2-2a	Ambient Air Quality Standards
45C-2-12	General emission limit provisions for sulfur dioxide
45C-2-11	Prevention Of Air Pollution Emergency Episodes
45C-2-13	Permit-to-Install New Sources and Permit-to-Install and Operate Program
45C-2-17	Restrictions of emissions of fugitive dust
45C-2-21	Control of emissions of VOCs from stationary sources

1.1.1 Permit Applicability

Air pollution control regulations have been established by the WVDEP for air emissions associated with stationary sources and fugitive emissions resulting from material transfer activities.

To determine permit applicability for the Facility's emission sources, the Potential-to-Emit (PTE) emissions have been presented in Attachment 4 and Permit Determination Form. The proposed Facility will be considered a minor source with potential uncontrolled PM emissions greater than 25 tons per year (tpy) and less than major source thresholds. Therefore, the Facility will need to obtain a permit to construct and operate. Applicable federal regulations present in Table 1-2 below.

Entire Green, Inc. [Application Fee](#)

Table 1-2 Federal Applicable Regulations

Rule	Description
40 CFR Part 61 Subpart JJJJ	Standards of Performance for Stationary Spark Ignition Internal Combustion Engines
40 CFR 61 Subpart A	General Provisions
40 CFR 60.1E	General control device and work practice requirements

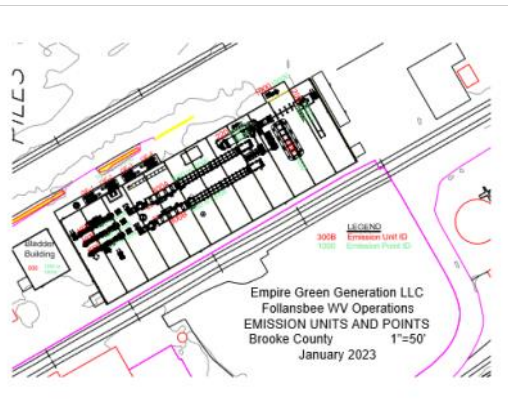
1.1.2 Criteria For Fuel

Processing the plastic food waste is not viewed as a waste digested through incineration, the plastic are being used as a fuel to create a syngas. The process for producing the syngas is through pyrolysis which is classified as an advanced recycling in West Virginia's House Bill 4048.

Terra Tech, Inc. Application Fee

ATTACHMENT E

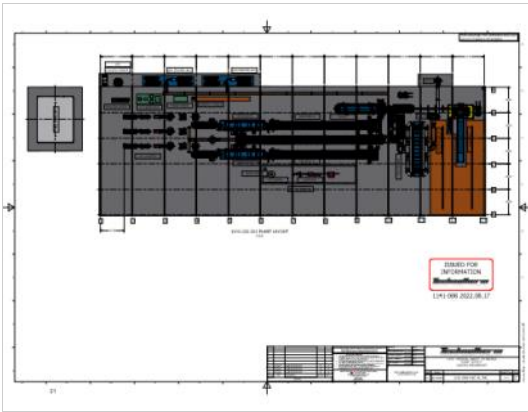
Plot Plan



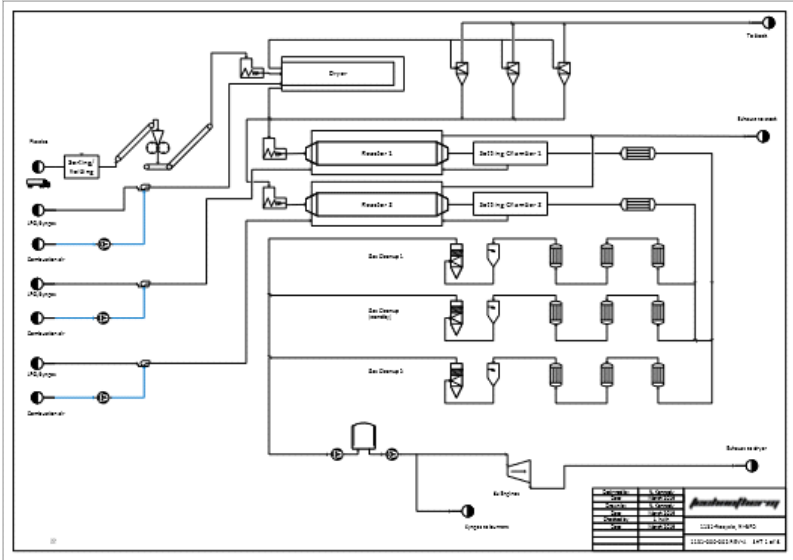
ATTACHMENT F

Process Flow Diagram

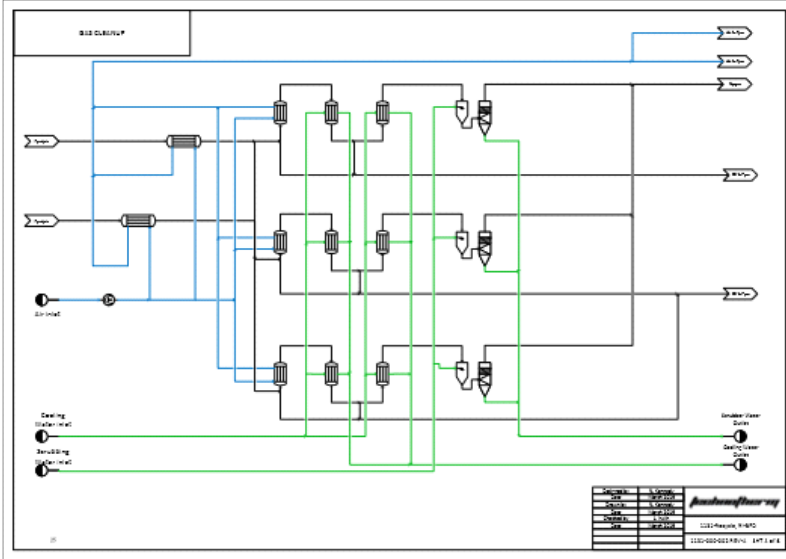
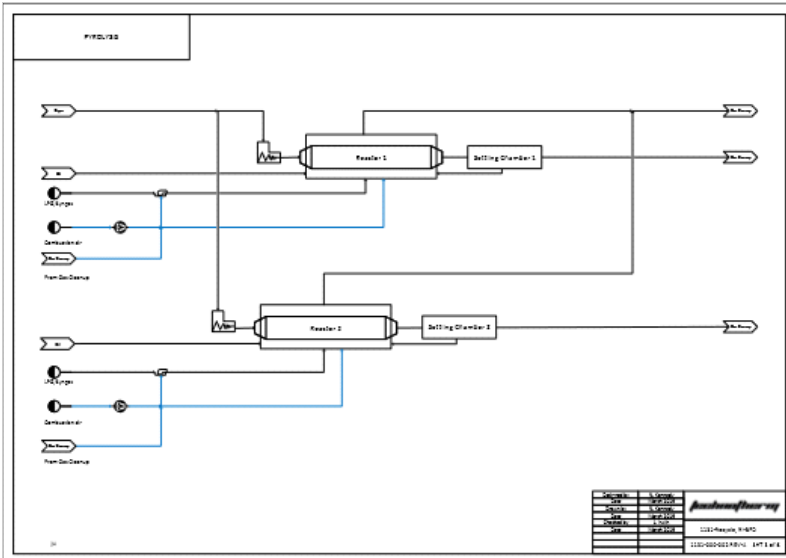
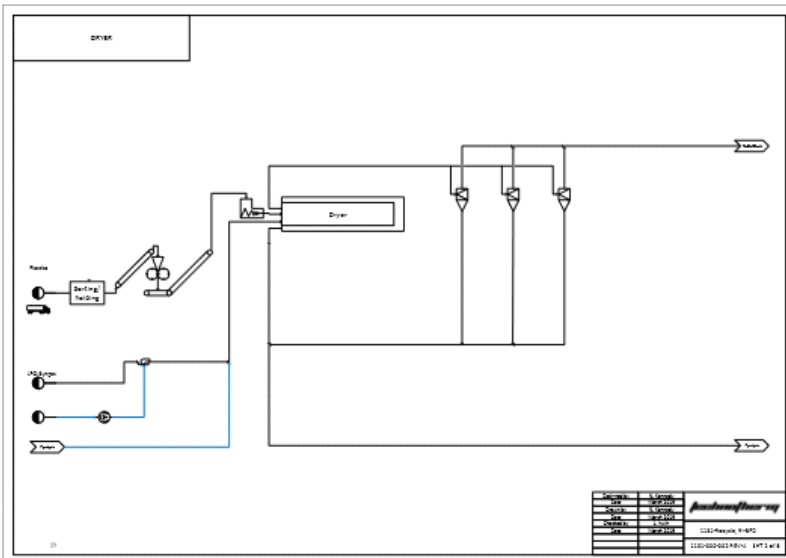
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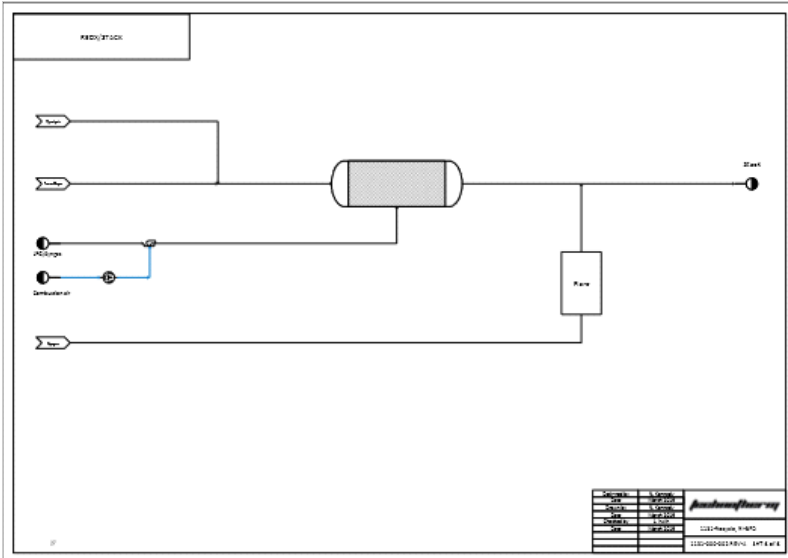
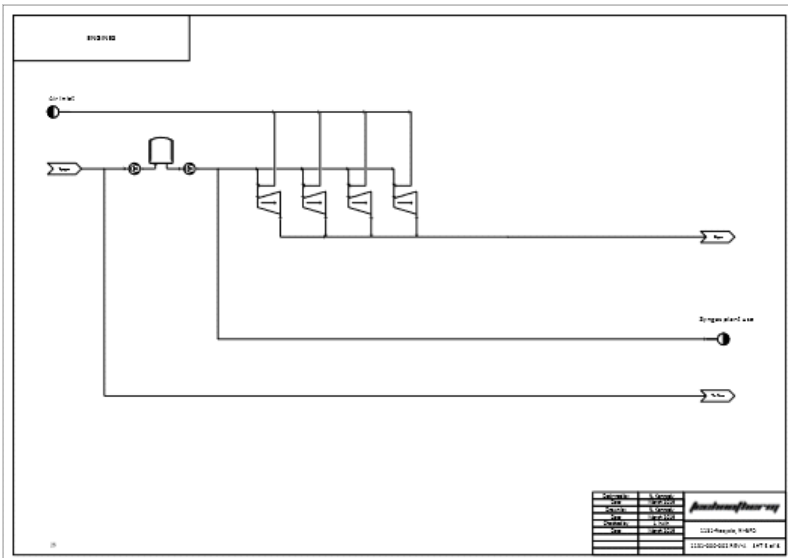


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ATTACHMENT G

Process Description

SUB-SYSTEM OPERA HOW PRIME EQUIPMENT DETAIL

Prime Equipment & Systems
The following description supplements the Process Flow Diagrams (PFDs) shown in Figure 1.

1. Delivery of Reactor

The material will be offloaded from the (3) helicopters per day and topped into the receiving container with the use of air topping stations as shown in Figure 2.



Figure 2: Illustration of typical air topping station.

The crane operator will report the weight of each load prior to topping. Additionally, a weighbridge operator records the weight of the loads as they enter and exit the plant.

The material will automatically load the Reactor as required and topped by the plant control philosophy.

An extraction system is built into the Reactor with an extraction fan pulling from the top of the Reactor at a rate of 1.0 m³/sec. Reactor will be under negative pressure of -0.025 mPa at all times during operations.



Figure 4: Negative Pressure from Reactor to Thermal Oxidizer.
Note all areas prior to thermal oxidation are sealed and under negative pressure.

Further detail, including details of prepared sealed containers are delivered to the plant and placed on an input conveyor. After passing on the conveyor system has no further movement into the Reactor.

Material input conveyor

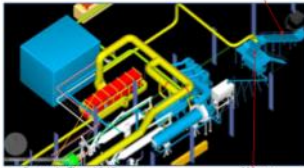


Figure 5.

2. Materials input conveyor to Reactor under negative pressure to Thermal Oxidizer.

The reactor / oxidizer has a single coating that is sealed. Around the valve is what is presented through a single entry point under suction that is closed when valve is not presented for an acknowledgment of during shut down mode.

The entire facility is sealed to the operating environment and operates under a negative pressure (0.025 mPa) ensuring no access of toxic or pathogenic.

Transfer from the Reactor to the dryer and from the dryer to the thermal oxidizer is also sealed to the environment operating under a negative pressure.

Progression to the high temperature cyclone unit (Fig. 7) is again sealed.

The material that is entering a negative pressure is an enclosed Dry Cell. This cell is connected to the regenerative thermal oxidizer from the thermal oxidizer to the Transition Dryer and all material interfaces.

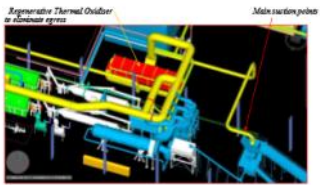


Figure 6: The Regenerative Thermal Oxidizer is maintained at 850°C.

2.1 Interlocks: The plant cannot be energized unless the (3) fan is on and the Thermal oxidizer at temperature. There is no possibility of start.

If the Thermal Oxidizer goes below a predetermined temperature, the plant goes into shut down mode. In the event of a power failure the main control facilities are powered via UPS. At this time a normally closed air vent valve will have closed the only access on the plant except for when the lock is activated.

Once the air vent of an (3) fan before power has been shut the plant will again default to shut down mode.

2.2 Macerator/Grapple Process

Macerator material (Seed stock) is conveyed to the thermal dryer feed hopper. Figure 7 shows a typical macerator system.



Figure 7. Photo of a typical Macerator/Grapple system.

3. Thermal Dryer

The drying of the feed stock is carried out in a direct heated, parallel flow, rotary drum type dryer using a combination of engine exhaust and, if necessary, firing oil/coal. Spinning sandstone, sponges and, as a last resort, propane.

Seed stock is transported from the dryer feed hopper into the dryer by means of a screw conveyor. Upon entering the dryer from the rear, the moist feed stock comes into direct contact with the parallel stream of gas.

Flue gases

Flare and progression gases achieve intimate contact between the feed stock and flue gas therefore facilitating efficient drying, evaporation and movement of feed stock across the drum. Once the feed stock and flue gases reach the closed end of the dryer they are discharged from the inner conveyor into the outer conveyor and return to the rear end of the dryer, discharging 10% moisture wet feed stock into an expansion chamber. Coarser dry feed stock falls to the bottom of the chamber forming a heap on the last conveyor loaded belt each.

The flue gas exhaust, comprising of light particulate feed stock material, is also discharged from the expansion chamber and cooled in a bank of cyclones where separation occurs. Fine particulate falls to the bottom of each cyclone and is discharged to hopper, all else into a common space containing the water separator. The flue product into the heat recovery during the dryer will maintain and the product. The particulate in the common dry feed stock stream into a water separator feeding an intermediate storage hopper that feeds into Pyrolysis.

The cooled flue gas stream from the cyclones is directed to the Flare.

Overall, Figure 8 shows process of a typical dryer in operation.



Figure 8. Photo of a typical dryer in operation from first floor level.



Figure 9. Photo of a typical dryer in operation from ground floor level.

4. Cyclones & Water/Separation Furnace

Overview

The product from a feed stock feed stock from a thermal dryer as described in the previous section. The Pyrolysis feed consists of fine particles of sponges. The unit the output of moist feed is conveyed to the wet flue gas from a combustion furnace located between the cyclones bank. These hot flue gases and the product flue gas and are conveyed to the product gas heat exchanger Thermal Dryer and during start up Surge Cooler and Fan Condenser. Subsequent heating of the product water is being provided by being a portion of the flue gas. The gas (100% is available for use should it be required where the flue gas is available. After passing through the cyclones the gas is progressed to the Flare/Gas.

Overall Description

Feed stock is transferred from the dryer to a live bottom screw hopper, which feeds an inner hopper complete with horizontal material feed screw. Material is fed from a gas-light, storage hopper into the horizontal, vertically angled, rotary drum Pyrolysis Reactor by a rotary screw.

As the material passes through the pyrolysis reactor, it undergoes thermal degradation releasing volatile organic sulfur compounds that end up in the sulfur in the sulfur-bearing liquid.

The heavy carbonaceous components of the gas and liquid streams, along with a specially designed high temperature desulfurization chamber where the particles are collected and returned to the distillation column.

The Pyrolyzers must be designed and arranged such that no process, for handling operations and also under any load condition and shall be complete with all auxiliary equipment for safe, reliable and efficient operation, and be of proper design capable of the required continuous, consistent and consistent operation and be suitable for its intended location. The design and materials of construction shall take into account the location.

Air and nitrogenous species produced by the Pyrolyzers escape off the fuel from the distillation desulfurization chamber where sulfur compounds, together with the heavy residue collected from the base of the Pyrolyzer into a liquid, a reservoir, that further treat to recover and clean the sulfur in the sulfur-bearing equipment. The heat recovered by burning the gas and oil is sufficient to heat the ash from the Pyrolyzer units above their ambient temperature with residue, produced or to burn off the ash. The ash is completely contained into CO2 and H2O.

Figure 10. 11 and 2 show process of Pyrolyzers in operation.



Figure 10. Photo A of typical high temperature pyrolysis unit.



Figure 11. Photo B of typical high temperature pyrolysis unit.



Figure 12. Photo of typical low temperature pyrolysis unit.

3. Sulfur Cleanup

Distillation Column

Raw sulfur is removed from the Pyrolysis Reactor, as described above, and passes through a desulfurization chamber and then hot sulfur. The sulfur is then stored for further use to ensure maximum efficiency during production and shipment as the final product. The steps to identify designed hot sulfur containers and from there is related to the distillation column described below.

Sulfur Storage

The sulfur cleaned, all hot, the gas flows into stainless steel sulfur storage. The sulfur is in essence a heat exchanger which indirectly transfers heat from sulfur to the combustion air stream.

Hot Containers

The sulfur from the cooler described above flows to a stainless-steel steel & tube heat exchanger which is cooled by an air stream system. The air is compressed and then the heated through the heat exchanger which is a single exchanger, not compressed as a product in the distillation column described above. The distillation column steel set of columns, each engine effluent flanges to heat the heat exchanger and thereby make the air to drop into the heated through below.

Oil Condensate

The syngas from the air condensers described above flows to a small 2 inch "Heat Exchanger" cooler and are cooled to near freezing levels. Condensate oils, which also contain condensed water, are collected and is purged to the incinerator furnace described previously. The carbon material after a set of cyclone gas syngas without flue gas to heat the heat exchangers and thereby cause the oils to enter the heated syngas below.

Water Scrubber

From the oil condensers the syngas flows through a high pressure fine Venturi Scrubber to remove any remaining SO₂.

Figure 13 and 14 show photos of a typical gas cleanup system in operation.



Figure 13. Photo A of a typical gas cleanup system in operation



Figure 14. Photo B of a typical gas cleanup system in operation

Gas Blower (Syngas Storage Tank)

The syngas storage tank provides surge capacity of cleaned syngas to allow for air and composition variations. It also has a blower connected with a downstream pipe. The blower will operate with an intake pressure of 20 to 45 inches gauge.

Figure 15 shows a typical gas blower in operation.



Figure 15. Picture of typical gas blower

Stack

The tail gas passes through the stack into the atmosphere after passing through the Thermal Oxidizer (discussed in next section).

6. Thermal Oxidizer

After passing through a scrubber the gas passes into a Thermal Oxidizer consisting of a rectangular duct shaped furnace. The internal dimensions are determined by the total volume gas needs to be heated to 850 °C and maintained for 2 seconds.

Figure 16 shows a typical Thermal Oxidizer in operation. Please refer to the technical specification file for more details if required.



Figure 16 Photo of a typical Thermal Oxidizer in operation

7. Surge Engine

Each surge engine shall be a fully packaged unit complete with all associated components and auxiliary. These engines are of robust design and have been proven in oil and medium density, heavy gas fuels.

The engine package will be complete to allow the engine to start, synchronize, operate continuously at base or peak load and shut down.

The surge engine shall be assembled in containers as indicated on the plant layout. The containerized engine shall conform to a sound pressure level of 80 dBA (2000 hours Regulatory Off-Axis) level as measured one meter from the enclosure at one meter above floor level.

Contaminating the surge clean up equipment and systems described in the Surge Clean up System Manual, the surge engine exhaust systems shall be designed and installed such that they meet emission standards as of the Commissioning Date.

The engine loading will be by means of external relations, they shall be designed and constructed with sufficient margin and spare surface area for the maximum rated reaction duty under all operating conditions.

The reactors shall incorporate features to minimize corrosion and erosion on the air and capabilities and suitable provisions for cleaning and core replacement.

The reactors and all of their component parts shall be of proven design and arranged so as to minimize maintenance work.

Figure 17 and Figure 18 illustrate the containerized engine in production.

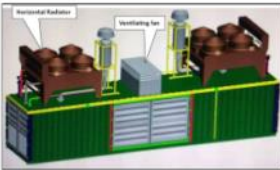


Figure 17. Illustration A of containerized engine in production

ATTACHMENT H

Material Data Safety Sheets (MSDSs)
Safety Data Sheets (SDSs)



Section 1: Identification

Product Identifier
Product Name - Synthetic Natural Gas
Relevant identified uses of the substance or mixture and uses advised against
Recommended use - Fuel for combustion applications, see material for chemical reactions
Details of the supplier of the safety data sheet
Manufacturer
Telephone (General)
Emergency Contact Information
Supplier
Manufacturer
UNSPSC

Section 2: Hazard Identification

United States (US)
According to USHA 29 CFR 1910.1200 HCS
Classification of the substance or mixture
USHA HCS 2012
Label elements
USHA HCS 2012
DANGER
Hazard statements

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Precautionary statements
Response
Other hazards
USHA HCS 2012
GHS07
GHS09

Classification of the substance or mixture
Label elements
Other hazards

Section 3: Composition/Information on Ingredients

Table with 5 columns: CASRN, Inertness, %, SOLUBLE, Composition Accounting, Comments. Row 1: Inertness, 44.2 to 49.0, 0, N/A, GDS 1000 99.99, None, See 1. Press. See 1.000, N/A.

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Table with 5 columns: Harmonized System, Inertness, %, SOLUBLE, Composition Accounting, Comments. Row 1: Harmonized System, 2826 1000, 1, N/A, GDS 1000 99.99, None, See 1. Press. See 1.000, N/A.

Section 4: First-Aid Measures

Description of first aid measures
Skin
Eyes
Ingestion
Most important symptoms and effects, both acute and delayed
Indication of any immediate medical attention and special treatment needed
Other information

Section 5: Fire-Fighting Measures

Extinguishing media
Special hazards arising from the substance or mixture
Exposure Pre and Explosion Hazards

Product Name: Synthetic Natural Gas
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Burns with a pale, fairly luminous flame, or containing more than 14% methane burns without noise.

Hazardous Consequences • No data available
Precautions

Advice for firefighters

- Structural firefighters protective clothing provides limited protection in fire situations ONLY if it is not affected in self situations where direct contact with the substance is possible.
- FIRE FIGHTING: Do NOT EXTINGUISH A LEAKING GAS FINE LINE LEAK LEAK CAN BE STOPPED. About containers from the area if you can do it without risk.
- FIRE: If safe, seal gas or tank track in a tank in a fire. ISOLATE for 100 meters (1 mile) in all directions, also consider other evacuation for 100 meters (1 mile) in all directions.
- FIRE: ALWAYS TANKS: ALWAYS stay away from tanks engulfed in the fire.
- FIRE: ALWAYS TANKS: Fight fire from maximum distance or use unmanned hose holders or remote nozzles.
- FIRE: ALWAYS TANKS: Withdraw immediately in case of rising sound from venting safety devices or discoloration of tank.
- FIRE: ALWAYS TANKS: Cool containers with flooding quantities of water until well after the fire is out.
- FIRE: ALWAYS TANKS: Do not direct water at source of leak or safety devices, using high pressure water.
- FIRE: ALWAYS TANKS: For massive fire, use unmanned hose holders or remote nozzles; if this is impossible, withdraw from area and let fire burn.

Section 6 - Accidental Release Measures

Personal precautions, protective equipment and emergency procedures

Personal Precautions • Do not touch damaged containers or spilled material unless wearing appropriate protective clothing. Do not walk through spilled material. Transfer the area before entry.

Emergency Procedures • ELIMINATE all ignition sources (no smoking, flames, sparks or fires in immediate area). As an immediate precautionary measure, isolate spill or leak area for at least 100 meters (330 feet) in all directions. Stop leak if you can do it without risk. Keep unauthorized personnel away. Stay out of low areas. Stop spill, LARGE SPILL, Consider initial containment for at least 100 meters (330 feet) if practicable. Use of equipment which cannot be equipped by switching off the current appropriate valve or main supply valve (control), however a gas for evacuation and to fully contact the local fire department.

Environmental precautions

- Prevent spreading of leaks through sewers, ventilation systems and confined areas.

Methods and material for containment and cleaning up

Containment/Clean-up • All equipment used when handling the product must be grounded. Stop use if you can do it without risk. If possible, turn leaking containers so that gas escapes rather than leaks. Use water spray to reduce vapors, do not get water directly on leak, spill area or inside container. Do not direct water at spill or source of leak. Isolate area until gas has dispersed.

Section 7 - Handling and Storage

Precautions for safe handling

Handling • Stay away from heat and ignition sources - No Smoking. Take precautionary measures against static charges. All equipment used when handling the product must be grounded. Use only non-sparking tools. Use only with adequate ventilation. Ventilate closed spaces before entering. Be aware of any signs of dizziness or fatigue, especially if work is done in poorly ventilated areas, exposure to high concentrations of this gas mixture could result without any significant warning symptoms, due to effects of fatigue or oxygen deficiency. Cylinders should be firmly secured to prevent falling or being knocked-over. Use explosion-proof electrical, ventilating and lighting equipment. Do not attempt to repair, adjust, or in any other way modify cylinders. If there is a malfunction on another type of operational product, contact nearest distributor immediately. Empty cylinders return product residue and can be hazardous. Do not heat, weld, puncture or incinerate container.

Conditions for safe storage, including any incompatibilities

Storage • Cylinders should be stored in dry, well-ventilated areas away from sources of heat, ignition and direct sunlight. Do not allow areas where cylinders are stored to exceed 50°C (120°F). Cylinders must be protected from the environment and particularly heat or high temperature approximately 25°C (77°F). Punctured cylinders appear physical damage. Cylinders should be firmly secured to prevent falling or being knocked-over. Block locked up.

Section 8 - Exposure Controls/Personal Protection

Control parameters

Material	Exposure Limits/Conditions		
	OSHA	NIOSH	IDLH
Carbon Dioxide (CO2) (gas)	3500 ppm (50 mg)	3500 ppm (50 mg)	3000 ppm (45 mg)
Hydrogen Sulfide (H2S) (gas)	10 ppm (1.4 mg)	10 ppm (1.4 mg)	100 ppm (14 mg)
Hydrogen Cyanide (HCN) (gas)	10 ppm (1.1 mg)	10 ppm (1.1 mg)	10 ppm (1.1 mg)

Exposure controls

Engineering • Good general ventilation should be used. Ventilation rates should be matched to conditions. If applicable, use process enclosures, local exhaust ventilation, or other engineering controls to maintain airborne levels below recommended exposure limits. If exposure limits have not been established, maintain airborne levels to an acceptable level. Use explosion-proof electrical, ventilating and lighting equipment.

Personal Protective Equipment

Respiratory • Follow the OSHA respiratory protection hierarchy in 29 CFR 1910.134. Use a NIOSH-approved respirator if exposure limits are exceeded or symptoms are experienced.

Safety • Wear safety glasses.

Handbook • Wear rubber gloves when handling cylinders.

Environmental • Follow best practice for site management and disposal of waste. Controls should be engineered to prevent release to the environment, including procedures to prevent spills, atmospheric release and release to wastewater.

Hygiene measures

Additional measures: Shower or decontaminate exposed regions. Wash immediately and thoroughly after use. Use of personal protective equipment.

Section 9 - Physical and Chemical Properties

Information on Physical and Chemical Properties

Physical Properties	
Physical Form	Gas
Color	Colorless
Color (solution)	Colorless
Odor	None

Chemical Properties	
Boiling Point	-89.3°C (-128.7°F)
Freezing Point	-109.9°C (-165.8°F)
Density (gas)	1.529 g/L (at 25°C and 101.325 kPa)
Density (liquid)	0.816 g/cm³ (at 25°C)
Specific Gravity (liquid)	0.504
Relative Vapor Density	1.529 (air = 1)
Water Solubility	Not applicable
Acidic Properties	Not applicable
Basic Properties	Not applicable
Reactivity	Non-reactive
Stability	Stable
Flammability	Highly flammable
Explosive Limits	5% - 15% (LEL - UEL)
Flash Point	-109.9°C (-165.8°F)
Autoignition Temperature	500°C (932°F)
Decomposition Temperature	Not applicable
Stability	Stable
Reactivity	Non-reactive
Stability	Stable

Section 10 - Stability and Reactivity

Reactivity

• No dangerous reaction known under conditions of normal use.

Chemical stability

• Stable

Possibility of hazardous reactions

• Hazardous decomposition will not occur.

Conditions to avoid

• Incompatible materials: avoid contact with heat and ignition sources. Excess heat.

Incompatible materials

• Reacts violently with powerful oxidizers (e.g. bromine pentafluoride, chlorine trifluoride, chlorine, fluorine, chlorine monofluoride, bromine trifluoride, bromine pentafluoride, chlorine dioxide, nitrogen dioxide, liquid oxygen).

Hazardous decomposition products

• No data available.

Section 11 - Toxicological Information

Information on toxicological effects

Composition	
Hydrogen (H ₂)	99.999% (mass fraction) (99.999% by volume)
Carbon Dioxide (CO ₂)	0.001% (mass fraction) (0.001% by volume)

Health Hazard	Classification
Acute Toxicity	Not classified
Chronic Toxicity	Not classified
Reproductive Toxicity	Not classified
Developmental Toxicity	Not classified
Respiratory Irritation	Not classified
Eye Irritation	Not classified
Skin Irritation	Not classified
Skin Sensitization	Not classified
Respiratory Sensitization	Not classified
Systemic Toxicity	Not classified

Potential Health Effects

Information

- Acute (Immediate)**
 - This material is a simple asphyxiant. May displace or reduce oxygen available for breathing especially in confined spaces. If this material is released in a small, poorly ventilated area (i.e. an enclosed or confined space), an oxygen-deficient environment may cause individuals breathing such an atmosphere may experience symptoms which include headache, dizziness, fatigue, loss of consciousness, unconsciousness, nausea, vomiting, and depression of the senses. Under some circumstances of over-exposure, death may occur. The following effects associated with decreased levels of oxygen: increase in breathing rate, pulse rate, emotional upset, abnormal fatigue, headache, weakness, collapse, loss of consciousness, convulsive movements, respiratory collapse and death.
- Chronic (Delayed)**
 - None
- Subacute**
 - None
- Subchronic**
 - None
- Chronic (Delayed)**
 - None

- Ingestion (Immediate)**
 - Under normal conditions of use, no health effects are expected.
- Chronic (Delayed)**
 - None

Section 12 - Ecological Information

- Toxicity**
 - Material data lacking.
- Persistence and degradability**
 - Material data lacking.
- Bioaccumulative potential**
 - Material data lacking.
- Biodegradability in soil**
 - Material data lacking.
- Results of PBT and vPvB assessment**
 - PBT and vPvB assessment has not been conducted for this material.
- Other adverse effects**
 - No studies have been found.

Section 13 - Disposal Considerations

- Waste treatment methods**
 - Dispose of content and/or container in accordance with local, regional, national, and/or international regulations.
- Packaging waste**
 - Dispose of content and/or container in accordance with local, regional, national, and/or international regulations.

Section 14 - Transport Information

ID#	UN number	UN proper shipping name	Transport hazard class(es)	Flashpoint group	Environmentally hazardous
DOT	1013	Compressed gas, flammable, n.o.s. (Hydrogen and Carbon Dioxide)	2.1	2.1	NEA
TSA	1013	Compressed Gas, Flammable, n.o.s. (Hydrogen and Carbon Dioxide)	2.1	2.1	Flammable Gases, Flammable
IMDG	1013	Compressed gas, flammable, n.o.s. (Hydrogen and Carbon Dioxide)	2.1	2.1	NEA

Special procedures for use: Cylinders should be transported in a secure position, in a well-ventilated vehicle. The transportation of compressed gas cylinders in automobiles or in closely-topped vehicles

can present serious safety hazards. If transporting these cylinders in vehicles, ensure these cylinders are not exposed to extremely high temperatures (as may occur in an enclosed vehicle on a hot day). Additionally, the vehicle should be well-ventilated during transportation.

- Transport in bulk according to Annex B of ADR159, 175 and the IBC Code**
 - None

Section 15 - Regulatory Information

Safety, health and environmental regulations/legislation specific for the substance or mixture

Country	Regulation	Classification
Canada	WHMIS - Hazardous Materials Information System	1000-100-0000
USA	OSHA - Occupational Safety and Health Administration	1000-100-0000
USA	DOT - Department of Transportation	1000-100-0000
USA	EPA - Environmental Protection Agency	1000-100-0000

Canada - WHMIS - Classification of Substances

Hydrogen	1000-100-0000	A, B1
Carbon dioxide	1000-100-0000	A, Compressed gas (see also WHMIS 1000-100-0000)

Canada - WHMIS - Ingredient Disclosure List

Hydrogen	1000-100-0000	Not Listed
Carbon dioxide	1000-100-0000	1%

USA - EPA - Priority Substances List

Hydrogen	1000-100-0000	Not Listed
Carbon dioxide	1000-100-0000	Not Listed

United States

Label - OSHA - Process Safety Management - Highly Hazardous Chemicals

Hydrogen	1000-100-0000	Not Listed
Carbon dioxide	1000-100-0000	Not Listed

Label - OSHA - Section 1910.1200 - Hazardous Chemicals

Hydrogen	1000-100-0000	Not Listed
Carbon dioxide	1000-100-0000	Not Listed

Label - EPA - Clean Air Act - 1990 Interstate Air Pollution

Hydrogen	1000-100-0000	Not Listed
Carbon dioxide	1000-100-0000	Not Listed

Methane	74-82-6	Not Listed
U.S. - CDRLASARA - Hazardous Substances and Their Reasonable Quantities		
Hydrogen	1333-74-0	Not Listed
Carbon dioxide	120-82-9	Not Listed
Methane	74-82-6	Not Listed
U.S. - CDRLASARA - Radioactive and Their Reasonable Quantities		
Hydrogen	1333-74-0	Not Listed
Carbon dioxide	120-82-9	Not Listed
Methane	74-82-6	Not Listed
U.S. - CDRLASARA - Section 301 Emergency Hazardous Substances (EPCRA) RQs		
Hydrogen	1333-74-0	Not Listed
Carbon dioxide	120-82-9	Not Listed
Methane	74-82-6	Not Listed
U.S. - CDRLASARA - Section 301 Emergency Hazardous Substances (EPCRA) TQs		
Hydrogen	1333-74-0	Not Listed
Carbon dioxide	120-82-9	Not Listed
Methane	74-82-6	Not Listed
U.S. - CDRLASARA - Section 302 - Corrosive Reporting		
Hydrogen	1333-74-0	Not Listed
Carbon dioxide	120-82-9	Not Listed
Methane	74-82-6	Not Listed
U.S. - CDRLASARA - Section 303 - PBT Chemical Listing		
Hydrogen	1333-74-0	Not Listed
Carbon dioxide	120-82-9	Not Listed
Methane	74-82-6	Not Listed
Inventory - United States - Section 306 Inventory (TSCA) - DBP Number to EPA Account Number Link		
Hydrogen	1333-74-0	Not Listed
Carbon dioxide	120-82-9	Not Listed
Methane	74-82-6	Not Listed

Product Name: Synthetic Natural Gas Page 16 of 12
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Section 16 - Other Information

Last Revision Date: 02/06/2019
Preparation Date: 01/06/2014

No to Information
GHS-to-MSDS
MSDS-to-GHS

Disclaimer/Statement of Liability
The information contained in this Safety Data Sheet (SDS) is believed to be correct since it was obtained from sources we believe are reliable. However, no representation, guarantee or warranty of any kind is made as to the accuracy, reliability or applicability of any material, conditions or methods, conditions and equipment used to store, handle, or process the material and hazards connected with the use of the material. It is the responsibility of the user and owner of the site to determine compliance with all applicable Federal, State, and local laws and regulations. Compliance with all applicable Federal, State, and local laws and regulations remains the responsibility of the user, and the user has the responsibility to provide a safe work place to ensure all aspects of its operation and to determine if or where precautions, in addition to those described herein, are required.

Product Name: Synthetic Natural Gas Page 11 of 12
Revision Date: 10/02/2019



Safety Data Sheet
Material Name: Propane - UN 1978 Gaseous
SDS ID: 00248010

Section 1 - PRODUCT AND COMPANY IDENTIFICATION

Material Name: Propane - UN 1978 Gaseous

Synonyms: PROPANE; Propane

Chemical Family: Hydrocarbons, Aliphatic

Product Description: Compressed in accordance with Compressed Gas Association standards.

Product Use: Industrial and Specialty Gas Applications.

Manufacturer: MATHESON TECHNOLOGICAL, INC.
300 Lake County Parkway
Lake Forest, IL 60045
Tel: 815-735-7100
Fax: 815-735-7101
Emergency: 1-800-424-4300 (24/7 M-F only)
Outside the US: 708-537-1181 (24/7 Global)

Section 2 - HAZARD IDENTIFICATION

Classification in accordance with paragraph 2.1 of EU CLP Regulation:
Flammable Gas - Category 1
Compressed Gaseous - Gaseous

GHS Label Elements:
Signal Word: Danger

Hazard Statement(s):
H202: Extremely flammable gas.
P201: Read the label and instructions carefully before use.

Precautionary Statement(s):
P201: Read the label and instructions carefully before use.

Response:
R102: Store from incompatible materials - No smoking.





Safety Data Sheet

Material Name: Propane - UN1073 Spacibo SDS ID: 002801010

Leaking gas fire: Do not extinguish unless leak can be stopped safely.
Extinguish all gas fires unless fire is out.

Storage
Store from sunlight. Store in a well-ventilated place.

Disposal
Disposal of contents: container to be accordance with local/regional/national/international regulations.
Other Details
May cause irritation upon sudden release of liquefied gas.

Section 2 - COMPOSITION / INFORMATION ON INGREDIENTS		
CAS	Component Name	Percent
76644	Propane	100

Section 4 - FIRST AID MEASURES

Inhalation
If inhaled, effects acute, remove to uncontaminated area. Give artificial respiration if not breathing. If breathing is difficult, oxygen should be administered by qualified personnel. Get immediate medical attention.

Eye
Irritation or freezing occurs. Immediately flush with plenty of lukewarm water (30-33°C / 86-91°F). DO NOT USE HOT WATER. If warm water is not available, gently wipe affected parts in lukewarm. Get immediate medical attention.

Skin
Wash immediately with water for several minutes. Remove contact lenses, if present and easy to do. Continue rinsing. Then get immediate medical attention.

Ingestion
If swallowed, get medical attention.

Must-Report Symptoms/Effects
None

Notes
No information on significant adverse effects.

Section 5 - FIRE FIGHTING MEASURES

Extinguishing Media
Suitable Extinguishing Media
Alcohol-resistant, water-soluble foam, Large fixed water spray or fog.

Unsuitable Extinguishing Media
Do not stream water at source of leak or safety devices, using any excess.

Special Risks, arising from the Chemical
Severe fire hazard. Severe explosion hazard. Critical situations are explosion. The vapor is heavier than air. Vapors or gases may ignite at distant ignition sources and then back. Backflow discharge may be generated by fire or explosion resulting in ignition or explosion.

Standard Fire-fighting Procedures
Evacuate of carbon.

Fire Fighting Measures
None



Safety Data Sheet

Material Name: Propane - UN1073 Spacibo SDS ID: 002801010

When someone has been in contact with the gas, remove them from the area. Apply water from a protected location or from a safe distance. Stay away from the leak if it leaks. For fire in engine or storage tank: Cool containers with water from a protected location or from a safe distance. Do not touch the tank or safety devices, using any excess water. Do not touch the tank. If possible, turn leaking container so that gas escapes rather than toward. The water spray to reduce vapors or direct spray should help. Avoid directing water toward to contain spilled material. Do not direct water at spill or source of leak. Stop unnecessary people nearby, isolate hazard area and deny entry. Ventilate closed spaces before entering.

Special Fire-fighting Equipment and Procedures for Firefighters
Wear full protective fire fighting gear including self-contained breathing apparatus (SCBA) for protection against possible explosion.

Section 6 - ACCIDENTAL RELEASE MEASURES

Personal Protection, PPE and Equipment and Emergency Procedures
Wear appropriate protective clothing and equipment see Section 8.

Methods and Materials for Containment and Cleanup
Avoid fire, flames, sparks and other sources of ignition. All equipment used when handling the product must be grounded. Remove sources of ignition. Do not touch or walk through spilled material. Stop leak if possible without personnel risk. If possible, turn leaking container so that gas escapes rather than toward. The water spray to reduce vapors or direct spray should help. Avoid directing water toward to contain spilled material. Do not direct water at spill or source of leak. Stop unnecessary people nearby, isolate hazard area and deny entry. Ventilate closed spaces before entering.

Environmental Precautions
Avoid release to the environment.

Section 7 - HANDLING AND STORAGE

Precautions for Safe Handling
Keep away from heat, sparks, open flames, and hot surfaces. No smoking. Avoid breathing dust/fumes/gas/vapors. The only condition is in a well-ventilated area. Wash hands thoroughly after handling. Use only after each use and when empty. Never put cylinders into uncontrolled areas of pressure vessels. Do not use other containers for transport purposes. Use only.

Conditions for Safe Storage, Including any Storage Restrictions
Store from sunlight. Store in a well-ventilated place.

Stability and Reactivity
Stable and non-hazardous in a container with its normal operating conditions and materials. Keep container tightly closed. Grounding and bonding required. Keep separated from incompatible substances.

Incompatible Materials
Combustible materials, oxidizing materials.

Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

Exposure	Control Measure
Propane	None
SDS-001	None
SDS-002	1000 ppm TWA, 1000 ppm STEL
SDS-003	1000 ppm TWA, 1000 ppm STEL
SDS-004	1000 ppm TWA, 1000 ppm STEL
SDS-005	1000 ppm TWA, 1000 ppm STEL



Safety Data Sheet

Material Name: Propane - UN1073 Spacibo SDS ID: 002801010

Section 2 - COMPOSITION / INFORMATION ON INGREDIENTS

Component Name	Percent
Propane	100

ACGIH - Threshold Limit Value - Biological Exposure Indices (BEI)
There are no biological limit values for any of the product's components.

Engineering Controls
Ventilation equipment should be explosion resistant if explosive concentrations of material are present. Provide local exhaust ventilation systems. Ensure compliance with applicable exposure limits. Use back flow prevention device on the PPE.

Individual Protection Measures, such as Personal Protective Equipment
See fire protection.

Eye Protection
For the gas: Eye protection not required, but recommended. For the liquid: Wear splash resistant safety goggles. Contact lenses should not be worn. Provide an emergency eye wash fountain and quick drench shower in the immediate work area.

Skin Protection
For the gas: Protective clothing is not required. For the liquid: Wear appropriate protective, cold insulating clothing.

Respiratory Protection
The following exposure and maximum use concentrations are derived from NIOSH and/or OSHA 29 CFR. Any certified respirator. Any self-contained breathing apparatus with a full facepiece. Emergency or standby entry into unknown concentrations or IDLH conditions. Any self-contained breathing apparatus that has a full facepiece and is operated in a pressure-demand or other positive-pressure mode. Any tight-fitting respirator with a full facepiece that is operated in a pressure-demand or other positive-pressure mode in combination with an auxiliary self-contained breathing apparatus operated in a pressure-demand or other positive-pressure mode. Except: Any appropriate escape-type, self-contained breathing apparatus.

Glove Recommendations
Wear insulated gloves.

Section 6 - ACCIDENTAL RELEASE MEASURES

Personal Protection, PPE and Equipment and Emergency Procedures
The following exposure and maximum use concentrations are derived from NIOSH and/or OSHA 29 CFR. Any certified respirator. Any self-contained breathing apparatus with a full facepiece. Emergency or standby entry into unknown concentrations or IDLH conditions. Any self-contained breathing apparatus that has a full facepiece and is operated in a pressure-demand or other positive-pressure mode. Any tight-fitting respirator with a full facepiece that is operated in a pressure-demand or other positive-pressure mode in combination with an auxiliary self-contained breathing apparatus operated in a pressure-demand or other positive-pressure mode. Except: Any appropriate escape-type, self-contained breathing apparatus.

Section 7 - HANDLING AND STORAGE

Precautions for Safe Handling
Keep away from heat, sparks, open flames, and hot surfaces. No smoking. Avoid breathing dust/fumes/gas/vapors. The only condition is in a well-ventilated area. Wash hands thoroughly after handling. Use only after each use and when empty. Never put cylinders into uncontrolled areas of pressure vessels. Do not use other containers for transport purposes. Use only.

Conditions for Safe Storage, Including any Storage Restrictions
Store from sunlight. Store in a well-ventilated place.

Stability and Reactivity
Stable and non-hazardous in a container with its normal operating conditions and materials. Keep container tightly closed. Grounding and bonding required. Keep separated from incompatible substances.

Incompatible Materials
Combustible materials, oxidizing materials.

Section 8 - EXPOSURE CONTROLS / PERSONAL PROTECTION

Exposure	Control Measure
Propane	None
SDS-001	None
SDS-002	1000 ppm TWA, 1000 ppm STEL
SDS-003	1000 ppm TWA, 1000 ppm STEL
SDS-004	1000 ppm TWA, 1000 ppm STEL
SDS-005	1000 ppm TWA, 1000 ppm STEL

Safety Data Sheet

Material Name: Propane - UN1075 Spacelac SDS ID: 00249810

Flamer Solubility:	(Very slightly) soluble	Flammability Coefficient: 0-100	Not available
Viscosity:	Not available	Reactivity: 0-5	Not available
Solubility (Other):	Not available	Stability:	Not available
Log KOW:	2.76	Physical Form:	Liquefied gas
Molecular Formula:	C ₃ H ₈	Molecular Weight:	44.10
Critical Temperature:	96.74 °C		

Safety Hazards: (check all that apply, otherwise remove application)

Section 10 - STABILITY AND REACTIVITY

Stability: No stability hazard is expected.
Chemical Stability: Stable at normal temperatures and pressures.
Reactivity with Water: No reaction.
Reactivity with Acids: No reaction. Gases, vapors and other sources of ignition. Minimum contact with material. Containers may rupture or explode if exposed to heat.
Incompatible Materials: Oxidizable materials, reducing materials.
Hazardous decomposition products: Oxides of carbon.

Section 11 - TOXICOLOGICAL INFORMATION

Information on Likely Routes of Exposure: Inhalation.
Routes: Irritation, vomiting, singed hair/beard, headache, dizziness, drowsiness, disorientation, mood swings, loss of coordination, confusion, unconsciousness, coma.
Skin Contact: Irritation, dryness.
Eye Contact: Irritation, blurred vision.
Ingestion: Irritation of gas in vehicle.
Acute and Chronic Toxicity: Component Categories: L200, L200, L200.
 The components of this material have been reviewed in various sources and the following selected endpoints are published:
Propane (74-98-6): Inhalation LC50 (Rat): 100000 ppm (41 min).

Safety Data Sheet

Material Name: Propane - UN1075 Spacelac SDS ID: 00249810

Flamer Solubility:	(Very slightly) soluble	Flammability Coefficient: 0-100	Not available
Viscosity:	Not available	Reactivity: 0-5	Not available
Solubility (Other):	Not available	Stability:	Not available
Log KOW:	2.76	Physical Form:	Liquefied gas
Molecular Formula:	C ₃ H ₈	Molecular Weight:	44.10
Critical Temperature:	96.74 °C		

Safety Hazards: (check all that apply, otherwise remove application)

Stability: No stability hazard is expected.
Chemical Stability: Stable at normal temperatures and pressures.
Reactivity with Water: No reaction.
Reactivity with Acids: No reaction. Gases, vapors and other sources of ignition. Minimum contact with material. Containers may rupture or explode if exposed to heat.
Incompatible Materials: Oxidizable materials, reducing materials.
Hazardous decomposition products: Oxides of carbon.

Section 12 - ECOLOGICAL INFORMATION

Component Analysis - Aquatic Toxicity: No data available.
Permeability and Degradability: No data available.
Biodegradability: No data available.
Accumulation Potential: No data available.
Other Toxicity: No additional information is available.

Section 13 - DISPOSAL CONSIDERATIONS

Disposal Method: Dispose in accordance with all applicable regulations.
Component Waste Disposal:

Safety Data Sheet

Material Name: Propane - UN1075 Spacelac SDS ID: 00249810

Flamer Solubility:	(Very slightly) soluble	Flammability Coefficient: 0-100	Not available
Viscosity:	Not available	Reactivity: 0-5	Not available
Solubility (Other):	Not available	Stability:	Not available
Log KOW:	2.76	Physical Form:	Liquefied gas
Molecular Formula:	C ₃ H ₈	Molecular Weight:	44.10
Critical Temperature:	96.74 °C		

Safety Hazards: (check all that apply, otherwise remove application)

Section 14 - TRANSPORT INFORMATION

U.S. DOT Information:
Shipping Name: PROPANE, LIQUEFIED
Hazard Class: 2.1
UN Number: 1075
Required Label(s): 2.1
MSDS Information:
Shipping Name: PROPANE, LIQUEFIED
Hazard Class: 2.1
UN Number: 1075
Required Label(s): 2.1
Dangerous Goods Chemical Code:
 This material does not contain any chemicals regulated by the IBC Code to be identified as dangerous chemicals in 49 CFR.

Section 15 - REGULATORY INFORMATION

U.S. Federal Regulations:
 Note: The following regulations apply to propane as listed under 49 CFR 192.401, 192.403, 192.405, 192.407, 192.409, 192.411, 192.413, 192.415, 192.417, 192.419, 192.421, 192.423, 192.425, 192.427, 192.429, 192.431, 192.433, 192.435, 192.437, 192.439, 192.441, 192.443, 192.445, 192.447, 192.449, 192.451, 192.453, 192.455, 192.457, 192.459, 192.461, 192.463, 192.465, 192.467, 192.469, 192.471, 192.473, 192.475, 192.477, 192.479, 192.481, 192.483, 192.485, 192.487, 192.489, 192.491, 192.493, 192.495, 192.497, 192.499, 192.501, 192.503, 192.505, 192.507, 192.509, 192.511, 192.513, 192.515, 192.517, 192.519, 192.521, 192.523, 192.525, 192.527, 192.529, 192.531, 192.533, 192.535, 192.537, 192.539, 192.541, 192.543, 192.545, 192.547, 192.549, 192.551, 192.553, 192.555, 192.557, 192.559, 192.561, 192.563, 192.565, 192.567, 192.569, 192.571, 192.573, 192.575, 192.577, 192.579, 192.581, 192.583, 192.585, 192.587, 192.589, 192.591, 192.593, 192.595, 192.597, 192.599, 192.601, 192.603, 192.605, 192.607, 192.609, 192.611, 192.613, 192.615, 192.617, 192.619, 192.621, 192.623, 192.625, 192.627, 192.629, 192.631, 192.633, 192.635, 192.637, 192.639, 192.641, 192.643, 192.645, 192.647, 192.649, 192.651, 192.653, 192.655, 192.657, 192.659, 192.661, 192.663, 192.665, 192.667, 192.669, 192.671, 192.673, 192.675, 192.677, 192.679, 192.681, 192.683, 192.685, 192.687, 192.689, 192.691, 192.693, 192.695, 192.697, 192.699, 192.701, 192.703, 192.705, 192.707, 192.709, 192.711, 192.713, 192.715, 192.717, 192.719, 192.721, 192.723, 192.725, 192.727, 192.729, 192.731, 192.733, 192.735, 192.737, 192.739, 192.741, 192.743, 192.745, 192.747, 192.749, 192.751, 192.753, 192.755, 192.757, 192.759, 192.761, 192.763, 192.765, 192.767, 192.769, 192.771, 192.773, 192.775, 192.777, 192.779, 192.781, 192.783, 192.785, 192.787, 192.789, 192.791, 192.793, 192.795, 192.797, 192.799, 192.801, 192.803, 192.805, 192.807, 192.809, 192.811, 192.813, 192.815, 192.817, 192.819, 192.821, 192.823, 192.825, 192.827, 192.829, 192.831, 192.833, 192.835, 192.837, 192.839, 192.841, 192.843, 192.845, 192.847, 192.849, 192.851, 192.853, 192.855, 192.857, 192.859, 192.861, 192.863, 192.865, 192.867, 192.869, 192.871, 192.873, 192.875, 192.877, 192.879, 192.881, 192.883, 192.885, 192.887, 192.889, 192.891, 192.893, 192.895, 192.897, 192.899, 192.901, 192.903, 192.905, 192.907, 192.909, 192.911, 192.913, 192.915, 192.917, 192.919, 192.921, 192.923, 192.925, 192.927, 192.929, 192.931, 192.933, 192.935, 192.937, 192.939, 192.941, 192.943, 192.945, 192.947, 192.949, 192.951, 192.953, 192.955, 192.957, 192.959, 192.961, 192.963, 192.965, 192.967, 192.969, 192.971, 192.973, 192.975, 192.977, 192.979, 192.981, 192.983, 192.985, 192.987, 192.989, 192.991, 192.993, 192.995, 192.997, 192.999.

Section 16 - OTHER INFORMATION

ATEX Group: Group 2 (I) or 3 (II) depending on the concentration.
Preparation Date: Not available.

Safety Data Sheet	
Effective date: 03/26/2015	according to 29CFR1910.1203 and GHS Rev. 3
Page 6 of 6	
Hydrochloric Acid, ACS	
Reproductive Toxicity:	No additional information.
SECTION 12: Ecological Information	
Ecotoxicity	
7683-83-6: Toxic by test: SC50 - Corrosive effect (Hazardous fish): 202 mg/L, 96 h (fish embryos, acif)	
Persistence and degradability	
Biodegradability potential	
Mobility in soil	
Other adverse effects	
SECTION 13: Disposal considerations	
Waste disposal recommendations	
Do not allow product to reach sewage system or open water. It is the responsibility of the waste generator to properly characterize all waste materials according to applicable regulatory criteria (EPA RCRA/FRC/SLU). Contact a licensed professional waste disposal service to dispose of this material. Dispose of empty containers as above. Product or container must not be disposed together with hazardous wastes. Chemical waste generators must determine whether a chemical chemical is classified as a hazardous waste. Then, if waste generators must also comply with regional and national hazardous waste regulations. Ensure compliance and accurate classification.	
SECTION 14: Transport information	
UN number	
3709	
UN proper shipping name	
HYDROCHLORIC ACID	
Transport hazard class(es)	
Corrosive	
Hazard class(es)	
3	
Packaging group(s)	
Environmentally hazardous	
Transport in bulk	
Special precautions for user	
SECTION 15: Regulatory information	
United States (USA)	
SARA Section 311/312 (Specific toxic chemical hazard)	
None	
SARA Section 313 (Specific toxic chemical hazard)	
None	
TSCA (2002) Hydrochloric Acid	
None of the ingredients is listed.	
TSCA (2002) Substance Control Act	
None of the ingredients is listed.	
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Created by: 184664636 Management, B.V., PO Box 1423478 Eindhoven - van.gelendoorn	

Safety Data Sheet	
Effective date: 03/26/2015	according to 29CFR1910.1203 and GHS Rev. 3
Page 7 of 8	
Hydrochloric Acid, ACS	
CEPCLA (Comprehensive Environmental Response, Compensation, and Liability Act)	
TSCA (2002) Hydrochloric Acid	
None of the ingredients is listed.	
Proposition 65 (California)	
Chemicals known to cause cancer:	
None of the ingredients is listed.	
Chemicals known to cause reproductive toxicity for females:	
None of the ingredients is listed.	
Chemicals known to cause reproductive toxicity for males:	
None of the ingredients is listed.	
Chemicals known to cause developmental toxicity:	
None of the ingredients is listed.	
Canada	
Canadian Domestic Substances List (DSL)	
None of the ingredients is listed.	
Canadian 1984 Ingredient Disclosure List (Book 6, 25)	
None of the ingredients is listed.	
Canadian 1984 Ingredient Disclosure List (Book 25)	
None of the ingredients is listed.	
SECTION 16: Other information	
This product has been classified in accordance with hazard criteria of the Globally Harmonized System and the GHS criteria. All information required by the Globally Harmonized System, including the responsibility to provide a safe workplace, applies to the user. The user should consider the health hazards and safety information contained herein in a safe and effective manner. Precautions required to do so include attention to hazard, employees and develop work practice procedures for safe work environment. For information contained herein to be the basis of safe handling and/or accident response, proper conditions of handling and use are paramount. Contact our office for questions or risks. All products are fully compliant with the use of the material. It is the responsibility of the user to comply with all applicable laws and regulations applicable to the material.	
GHS Add Text Phrases:	
Abbreviations and acronyms:	
MSD: Material Safety Data Sheet for Hazardous Goods	
PPE: Personal Protective Equipment (PPE)	
CFR: Code of Federal Regulations (CFR)	
SARA: Superfund Amendments and Reauthorization Act (SARA)	
NCHA: National Chemical and Hazardous Act (NCHA)	
TSCA: Toxic Substances Control Act (TSCA)	
NPL: National Pollution Release Inventory (National Pollution Release Inventory)	
GHS: Globally Harmonized System of Classification and Labeling of Chemicals	
ACGIH: American Conference of Governmental Industrial Hygienists	
CAS: Chemical Abstracts Service Division of the American Chemical Society	
NTP: National Toxicology Program Accession (NTP)	
16	
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Safety Data Sheet	
Effective date: 03/26/2015	according to 29CFR1910.1203 and GHS Rev. 3
Page 8 of 8	
Hydrochloric Acid, ACS	
HSE: Hazardous Materials Identification System (HMIS)	
WHMIS: Workplace Hazardous Materials Information System (Canada)	
PPE: Personal Protective Equipment (PPE)	
Effective date: 03/26/2015	
Last updated: 03/26/2015	
17	
Created by: 184664636 Management, B.V., PO Box 1423478 Eindhoven - van.gelendoorn	

FUGITIVE EMISSIONS SUMMARY	All Required Pollutants/ Chemical Name/CAS#	Maximum Potential Uncontrolled Emissions ¹		Maximum Potential Controlled Emissions ¹		Est. Unavoidable Emissions ²
		lb/hr	lb/day	lb/hr	lb/day	
Hot Spot/Reel Coil Emissions Pallet Heat Treat						
Unpaved Hot Roads						
Storage Pile Emissions						
Loading/Unloading Operations						
Wastewater Treatment Equipment & Operations						
Equipment Leaks	Simulated Natural Gas 74-80-2 Propylene 74-80-2		20.41			
Sanitary Drainage VOC Emissions						
Other						

¹ List all required air pollutants. Specify VOCs, including all HAPs. Refer chemical name with Chemical Abstracts Service (CAS) number. List Acids, CO, CO₂, VOCs, HAPs, nitrogen oxides, Oxygen, SO₂, SO_x, SO₃, SO_x, all ammonia derivatives. Specify individual VOCs and methane, ethane, DCE, VC, UST, H₂, H₂S, H₂O, H₂, O₂, and other gases.
² Specify rate with no control equipment operating. If emissions occur for less than 1 hr, then report emissions per batch in minutes (e.g. 3 lb / VOC 30 minute batch).
³ Specify rate with proposed control equipment operating. If emissions occur for less than 1 hr, then report emissions per batch in minutes (e.g. 3 lb / VOC 30 minute batch).
⁴ Include method used to determine emission rate as follows: UB = manual; BC = test; ST = short test (give date of test); EA = engineering estimate; O = other (specify).

ATTACHMENT L

Emission Units Data Sheet

Attachment L
EMISSIONS UNIT DATA SHEET
CHEMICAL PROCESS

For chemical processes please fill out this sheet and all supplementary forms (see below) that apply. Please check all supplementary forms that have been completed.

Emergency Unit Summary Sheet
 Leak Survey Data Sheet
 Inventory Data Sheet
 Emission Data Sheet
 Chemical Calculus Data Sheet

1. Chemical process area name and equipment ID number (as shown in Equipment Log Form)

2. Standard Industrial Classification Codes (SICs) for process(es)

3. List type reactions and attach MSDSs

4. List Products and Maximum Production and attach MSDSs

Description and CAS Number	Maximum Hourly (lb/hr)	Maximum Annual (lb/year)

5. Complete the Emergency Unit Summary Sheet for all emergency relief devices.

6. Complete the Leak Survey Data Sheet and describe below in detail to appropiate the leak detection or maintenance program to minimize fugitive emissions. Include detection instruments, calibration gases or methods, planned inspection frequency, and record keeping and similar pertinent information. If subject to a rule requirement (e.g. RCFCRIS, Subpart VV), please list those here.

7. Copy the details below or attach to application Accident Procedures to be followed in the event of an accident spill or release.

B. Maximum Temperature 99 °C 234°F		JA. Maximum Pressure 10.0 bar (145 psig) 10.0 MPa (145 psig)	
C. Output Data - Flow Out:		g/Phr of particles	
Material Name and CAS No.	Process	Specific Gravity	Major Pressure
Varies per feed	SCF	TBD	TBD
	TBD	TBD	TBD
K. Complete the following emission data for equipment connected to a header exhaust system, giving emissions levels <u>below</u> entering header system (a. below control equipment). <input type="checkbox"/> Check here if not applicable Emission Point ID (nearest point of header system): TBD Material Name and CAS No.: TBD Maximum Potential Emission Rate (lb/hr): TBD Method **: TBD			

10. Provide the following information pertaining to each condenser that may be attached to this reactor. Attach additional pages as necessary if more than one condenser is used for this reactor. Complete the Condenser Air Pollution Control Device Sheet if necessary.

Check here if not applicable

10A. Coating material:
10B. Minimum and maximum flowrate of coating material (g/Phr)
10C. Inlet temperature of coating material (°F)
10D. Outlet temperature of coating material (°F)
10E. Pressure drop of gas to be condensed from inlet to outlet (psig)
10F. Inlet temperature of gas stream (°F)
10G. Outlet temperature of gas stream (°F)
10H. Number of passes
10I. Coating surface area

11. Provide the following pertaining to auxiliary equipment that burns fuel (boilers, dryers, etc.):
 Check here if not applicable

11A. Type of fuel and maximum fuel burn rate, per hour.
Name of Gas for storage in 120-Gal or 480, 600-gal or other storage vessel once set up has been completed.

11B. Provide maximum percent sulfur (S), ash content (of fuel), and the energy content using appropriate units.
% S Ash BTU/Lb, etc. BTU/cu. ft (include units)

11C. Theoretical combustion air requirement in SCFD per unit of fuel (include appropriate unit @ 75°F and 14.7 PSIA).
SCFD/ton, SCFD/gal (include units)

11D. Percent excess air: %

11E. Type, amount, and BTU rating of burners and all other firing equipment that was planned to be used.

11F. Total maximum design heat input: x10⁶ BTU/hr.

12. **Process Monitoring, Recordkeeping, Reporting, and Testing**
Please provide monitoring, recordkeeping, and reporting details to demonstrate compliance with the proposed operating parameters. Please provide testing in order to demonstrate compliance with the proposed emissions limit.

MONITORING: Provide and specify inspection according to all applicable and identify inspection along with any manufacturer's manufacturer specifications.	RECORDKEEPING
REPORTING: Any notifications	TESTING: Per report

MONITORING: PLEASE LIST AND DESCRIBE THE PROCESS PARAMETERS AND RANGES THAT ARE PROPOSED TO BE MONITORED IN ORDER TO DEMONSTRATE COMPLIANCE WITH THE OPERATION OF THE PROCESS EQUIPMENT OPERATE UNDER AIR POLLUTION CONTROL DEVICE.

RECORDKEEPING: PLEASE DESCRIBE THE MONITORING RECORDKEEPING THAT WILL ACCOMPANY THE MONITORING.

REPORTING: PLEASE DESCRIBE THE MONITORING FREQUENCY OF REPORTING OF THE RECORDKEEPING.

TESTING: PLEASE DESCRIBE ANY PROPOSED INSPECTION TESTING FOR THE PROCESS EQUIPMENT OR AIR POLLUTION CONTROL DEVICE.

13. Operation of operating ranges and maintenance procedures required by manufacturer to maintain warranty: N/A

DISTILLATION COLUMN DATA SHEET

Identification Number (as assigned on Equipment Use Form):

1. Name and type of equipment

2. Projected actual equipment operating schedule (complete appropriate lines)
 Monday _____ days/year
 Wednesday, both weekend _____ days/year
 (include one)

3. Number of stages (plates), including condenser

4. Number of feed plates and stage location

5. Specify details of any reheating, recycling, or stage conditioning along with the stage location

6. Specify reflux ratio, R, (where R is defined as the ratio of the reflux to the overhead product, given symbolically as $R = D/D_1$, where $D_1 = \text{liquid down column}$, $D = \text{distillation product}$)

7. Specify the fraction of feed which is evaporated, F (where F is the mole fraction of the feed that leaves the feed plate continuously as vapor)

7A. Type of condenser used: heat partial multiple other

7B. For each condenser provide process operating details including all inlet and outlet temperatures, pressures, and compositions

8. Feed Characteristics
 A. Molar composition
 B. Individual vapor pressures of each component
 C. Total feed stage pressure
 D. Total feed stage temperature
 E. Total mass flow rate of each stream into the system

9. Overhead Product
 A. Molar composition of components
 B. Vapor pressure of components
 C. Total mass flow rate of all streams leaving the system as overhead products

10. Bottom Product
 A. Molar composition of all components
 B. Total mass flow rate of all streams leaving the system as bottom products

11. General Information
 A. Distillation column diameter
 B. Distillation column height
 C. Type of column
 D. Plate spacing
 E. Murphree plate efficiency
 F. Any other information necessary to describe the operation of this distillation column

12. Process Monitoring, Recordkeeping, Reporting, and Testing
 Please propose monitoring, recordkeeping, and reporting in order to demonstrate compliance with the proposed operating conditions. Please propose testing in order to demonstrate compliance with the proposed emissions limits.

MONITORING	RECORDKEEPING
REPORTING	TESTING

MONITORING: PLEASE LIST AND DESCRIBE THE PROCESS PARAMETERS AND RANGES THAT ARE MONITORED TO BE SENSITIVE TO OPERATIONAL CHANGES WITH THE UNDERSTOOD AND PROPOSED OPERATING CONDITIONS AND POLLUTION CONTROL DEVICES.

RECORDKEEPING: PLEASE DESCRIBE THE PROPOSED RECORDKEEPING THAT WILL ACCOMPANY THE MONITORING.

REPORTING: PLEASE DESCRIBE THE PROPOSED FREQUENCY OF REPORTING OF THE RECORDKEEPING.

TESTING: PLEASE DESCRIBE ANY PROPOSED EMISSIONS TESTING FOR THIS PROPOSED EQUIPMENT OR AIR POLLUTION CONTROL DEVICES.

13. Describe all operating ranges and maintenance procedures required by manufacturer to maintain air quality.

NOTE: An AIR POLLUTION CONTROL SERVICE SHEET must be completed for any air pollution device(s) (except emergency relief devices) used to control emissions from this distillation column.



Supertherm (Pty) Ltd is a TECHNO THERM
 Head Office: 1001 Phyllis Street, Durban, KwaZulu-Natal, 4001
 Contact: 031 291 3443
 Email: info@technotherm.co.za
 www.technotherm.co.za

1131

TECHNICAL FILE
 (Area-010)

Phase Pyrolyser for Power Plant

PYROLISER PLANT
 (Area 010)

Equipment – MEDRECYCLER – 010

Redacted Copy - Claim of Confidentiality
Pyrolyser Plant
Equipment No. General-D10-001

1131

[REDACTED]

Redacted Copy - Claim of Confidentiality
Pyrolyser Plant
Equipment No. General-D10-001

1131

[REDACTED]

Redacted Copy - Claim of Confidentiality
Pyrolyser Plant
Equipment No. General-D10-001

1131

[REDACTED]

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 Pyrolyser Plant
 Equipment No. General-0001

1131

[Redacted Content]

Redacted Copy - Claim of Confidentiality

[Redacted Content]

REVISION	1	DATE	01/01/2017
BY	SA	APPROVED BY	SA
REVISION	2	DATE	01/01/2017
BY	SA	APPROVED BY	SA
REVISION	3	DATE	01/01/2017
BY	SA	APPROVED BY	SA
REVISION	4	DATE	01/01/2017
BY	SA	APPROVED BY	SA
REVISION	5	DATE	01/01/2017
BY	SA	APPROVED BY	SA

Technoheraq
 Superior Thermal Technologies
 PYROLYSIS PLANT
 PYROLYSER
 GENERAL ARRANGEMENT
 DRG No :- 1131-000 SHT 1 OF 1

Attachment L
 Airman and Dale Street
 INCINERATOR

Control Device ID No. (Must match List Form) 860

Equipment Information

1. Manufacturer: Technoheraq	2. Model No. Regenerative Thermal Oxidizer
3. On a separate sheet sketch or draw the proposed incinerator showing the location and dimensions (inside and out) of (1) the primary combustion chamber, (2) the secondary combustion chamber, (3) the flame port, (4) scrubber (if any), and (5) ductwork with special emphasis on dimensions of the flame port and secondary combustion chamber (if any). Also, indicate the minimum distance the gas travels through the secondary combustion chamber.	
4. Rated capacity of the incinerator for the type of waste to be burned: Maximum: 143,000 lbs/hr Typical: 72,570 lbs/hr Annual: 287,765 tons/yr	
5. By what means is waste charged? <input type="checkbox"/> Batch <input checked="" type="checkbox"/> Continuous <input type="checkbox"/> Periodically	
6. Type: <input type="checkbox"/> Multiple Chamber <input checked="" type="checkbox"/> Single Chamber <input type="checkbox"/> Other, specify _____	
7. Proposed operating schedule: 24 hrs/day 7d/week	

Primary Combustion Chamber

8. Volume: 87 cu ft	9. Effective grate area: 87 sq ft
10. Maximum temperature: 2700 F	11. Burning rate: 143,000 lbs/hr
12. Heat release in primary chamber: 87,133,000 BTU/hr	13. Total heat release in incinerator: 87,133,000 BTU/hr

Secondary Combustion Chamber

14. Volume: 70 cu ft	15. Gross sectional area: 87 sq ft
16. Volume of gas through secondary combustion chamber: ACFM @ 7 F chamber: 87,000	17. Gas velocity through secondary combustion chamber: 87,000 ft/min
18. Minimum gas temperature: 7 F	19. Minimum retention time of gas: 60 sec
20. Minimum distance of gas travel through secondary combustion chamber: 8 ft	21. Location of air admission: _____

Flame Port

22. Flame port area: 87 sq ft	23. Velocity through flame port: 87,000 ft/min
-------------------------------	--

Dampers

24. Type: Periodically operated push block lever	25. Number: 4
26. Diameter: inches	27. Capacity: ACFM @ 7 F

Combustion Air

28. Type of draft: Natural Forced
 Induced Induced
 29. If draft is forced or induced, describe ID fans or blowers:
 Number: 1 fan
 HP rating: 1 HP
 Model/Type: 1.276
 RPM: 1725
 Fan rated draft: 100
 30. Theoretical air/fuel ratio: 14.7
 31. Percent of total air supplied as: outside air recirculated air

Primary Burner

32. Burner type and fuel: NG

33. Primary Burner
 Capacity: 1.74 MBTU/hr
 Number: 2
 Manufacturer: Turbochem
 Model: TSD
 Estimated capacity: 87 L/hr
 Fuel: NG
 Flow controlled? Yes No
 Is there a temperature indicator? Yes No
 (How temperature measured? Thermocouple sensor)

34. Automatic loading device: Yes No
 If yes, describe: Gas flow

35. Spill prevention: Yes No

36. Method of cleaning: Substrate for combustion gases: Air

37. Other interlocking devices or controls: If yes, describe: Maintenance and fire protection

Secondary Burner

38. Secondary Burner
 Capacity: N/A
 Number: MBTU/hr
 Manufacturer:
 Model:
 Estimated capacity: 87 L/hr
 Fuel:
 Flow controlled?
 Is there a temperature indicator? Yes No

39. Flame failure protection equipment: Yes No

40. Method of cleaning secondary or wetting chamber: Describe: N/A

41. Other interlocking devices or controls: If yes, describe: Maintenance and fire protection

Insulation

42. Insulator Installation: Yes No
 If yes, describe method of supplying combustion air:

43. Exhaust Insulation: Yes No

Stack or Vent Data

44. Inside diameter or dimensions: 3.75
 45. Stack exit temperature: 160
 46. Height: 45.0
 47. Stack service: This equipment only
 Other equipment also (indicate type and rating of all other equipment exhausted through the stack or vent)
 48. Gas flow rate: 11,219
 49. Estimated percent of recirculation: %

Process

50. Source of waste: Commercial Hospital Restaurant Store Industry Apartment
 Cemetery Laboratories Public Institution Other, specify:

51. Describe briefly or sketch the composition of waste feed to the incinerator.
 The waste feed will vary and is a combination of all of the processes from the process equipment and engines.

52. Expected BTU/hr as fired: 87 L/hr
 53. Daily amount: lb

54. Does incinerator have a charge hopper? Yes No
 55. What is the volume of the charge hopper? m³

56. Does the charge hopper have automatic control? Yes No
 57. Is the waste charged to the incinerator weighed? Yes No

58. Is the secondary chamber preheated prior to charging waste? Yes No
 59. At what secondary temperature does waste charging begin? °F

60. Is the ash waste quenched? Yes No
 61. Is all the waste burned/gasified on site? Yes No

62. For hazardous waste, is the ash recycled for recognizable combustible components? Yes No
 63. For hazardous waste, are any recognizable combustible components of the ash recycled? Yes No
 64. Is any waste received from outside the local government boundary? Yes No

65. Are hazardous or special waste burned? Yes No
 If yes, please describe:

66. Are potential infectious waste burned? Yes No

67. How will the waste bypass from process air control equipment be disposed of? Through the stack. Gas.

68. Method of charging waste solids:
 Manual
 Manual charge hopper
 Automatic charge hopper
 Other, specify: _____

69. Method of heating liquids: Gas jacket
 Injection as a primary burner fuel
 Injection as a secondary burner fuel
 Other, specify: _____

70. Related steam flow - heat recovery boiler: _____
 71. Related pressure - recovery boiler: _____

Incinerator Emission

Pollutant	Emission Rate per Hour	Emission Rate per Hour				Tons per Year	Tons per Million Gallons
		lb/hr	kg/hr	lb/day	kg/day		
CO							
Hydrocarbons							
NO _x							
PM ₁₀							
SO _x							
HCl							
Other (specify):							

73. If an Air Pollution Control Device is not submitted, the emission rates should be the same as those reported on the Maximum Potential and Maximum Actual Emission on the Emission Points Data Summary Sheet.

74. Emission rates should be submitted by submitting each unit used in their combustion.

Fuel Usage Data

75. Submitter annual fuel cost: \$

76. Fueling rate: Hourly Typical Design
 77. Fuel type: Natural Gas Coal
 Fuel Oil, No. _____
 Other, specify: _____

78. Typical heating content of fuel: _____
 79. Typical fuel sulfur content: wt. %

80. Report fuel cost in terms: \$/Btu
 81. Annual fuel usage: _____

82. Please indicate an Air Pollution Control Device (APCD) for the unit(s) used on this Emission Unit, if applicable.

83. Have you included the air-pollution rates on the Emission Points Data Summary Sheet?

34. Proposed Monitoring, Recordkeeping, Reporting, and Testing
 Please propose monitoring, recordkeeping, and reporting in order to demonstrate compliance with the proposed operating parameters. Please propose testing in order to demonstrate compliance with the proposed emission limits.

MONITORING PLAN: Please list (1) describe the process parameters and how they were chosen (2) the design and how they were established for monitoring to demonstrate compliance with the operation of the process equipment operation or air pollution control device. Monitoring will be accomplished using the 10 standard tests. The facility assumed will conduct monthly inspections for visible stack emissions and colorless. The 10 stack emission observations will not be required to be performed by a person certified as a qualified observer under EPA Method 9 for Visual Observation of Emissions from Stationary Sources.

TESTING PLAN: Please describe any proposed emissions testing for this process equipment or air pollution control device.

RECORDKEEPING: Please describe the proposed recordkeeping that will accompany the monitoring. Record all copies in the system Visual Inspection reports will be maintained on-site in the control room.

REPORTING: Please describe the proposed frequency of reporting of the recordkeeping. Report to WYDOP all copies sent to the file.

35. Please describe all operating ranges and maintenance procedures required by Manufacturer to maintain operation.
 Annual maintenance manual provided to the plant.



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 Superior thermal technologies
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 SRT 488025864 | Reg No: 2010/000941/07
 80 business number 0871 78 111 9218
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1131

TECHNICAL FILE
 (Area-950)

Regenerative Thermal Oxidiser for
 Power Plant

**REGENERATIVE
 THERMAL OXIDISER (Re-Ox)**
 (Area 950)

Equipment – MEDRECYCLER – 950

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Regenerative Thermal Oxidiser
Equipment No. General-002

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Equipment No. General-002

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Equipment No. General-002

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Regenerative Thermal Oxidiser
Equipment No. Genwar200

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Haul Road Fugitive

Attachment L

Screened Unit Data Sheet
(NORMEET HEAT EXCHANGERS)

Control Device ID No. (Must match List Form) 400

Equipment Information	
1. Manufacturer: Technotherm	2. Model No.: Serial No.:
3. Number of units: One destruction chamber with 2 cyclones, two-stage heat exchanger, air coolers, oil condensing scrubbers, Wet Ash/Slag Gas Scrubbers, APC CRT Scrubbing into one scrubber, Cooling	4. Unit Gas Capacity (scfm):
5. Rated Boiler Horsepower: N/A	6. Boiler Name No.:
7. Date constructed:	8. Date of last modification and update:
9. Maximum design heat input per unit: N/A	10. Peak heat input per unit: +10% BTU/hr
11. Steam produced at maximum design output: N/A	12. Projected Operating Schedule: Hours/Day: 24 Days/Week: 7 Weeks/Year: 52
13. Type of firing equipment to be used: <input type="checkbox"/> Pulverized coal <input type="checkbox"/> Spreader stoker <input type="checkbox"/> Circulating <input type="checkbox"/> Natural Gas Burner <input type="checkbox"/> Others, specify	14. Physical type of burners and orientation: <input type="checkbox"/> Vertical <input type="checkbox"/> Front Wall <input type="checkbox"/> Backward <input type="checkbox"/> Tangential <input type="checkbox"/> Others, specify
15. Type of draft: <input type="checkbox"/> Forced <input type="checkbox"/> Induced	16. Percent of ash retained in furnace: %
17. Will flash be retracted? <input type="checkbox"/> Yes <input type="checkbox"/> No	18. Percent of carbon in flyash: %
Stack or Vent Data	
19. Inside diameter or dimensions: N/A	20. Gas exit temperature: 500 °F
21. Height: 8'	22. Stack service: <input type="checkbox"/> This equipment only <input type="checkbox"/> Other equipment also (list type and rating of all other equipment exhausted through this stack or vent)
23. Gas flow rate: #/hr	
24. Estimated percent of incinerate: %	

Fuel Requirements					
Type	Fuel Oil No.	Natural Gas	Gas (Other than NG)	Coal, Type	Other
Quantity of Output	gal/2000 F	lb/hr	lb/hr	lb/hr	
Annually	+10' gal	+10' lb/hr	+10' lb/hr	tons	
Sulfur	Maximum wt. %	g/100 lb	g/100 lb	Maximum wt. %	
	Average wt. %				
BTU Content	BTU/Gal.	BTU/lb	BTU/lb	BTU/lb	
	calories/2000 F				
Source					
Supplier					
Propane (Y/N)					
Coal and Inventory Details					

26. Gas burner mode of control:
 Manual Automatic full range Automatic full modulation Automatic on/off
 OK burner manufacture OK burner manufacture

27. Gas burner manufacture:
 OK burner manufacture OK burner manufacture

28. If fuel oil is used, how is it delivered?
 Pipeline Tanker Compressed Air Other source?

29. Fuel oil preheated? Yes No If yes, indicate temperature. °F

30. Specify the calculated (described) air requirements for combustion of the fuel or mixture of fuels described above actual cubic feet (ACF) per unit of fuel:
 ACF _____ % moisture

31. Excess air or rated capacity: _____ %

32. Percent excess air actually required for combustion of the fuel described: _____ %

33. Sulfur: _____

34. Proximate analysis (dry basis):
 % of Fixed Carbon: _____ % of Sulfur
 % of Moisture: _____ % of Volatile Matter
 % of Ash: _____

Emission Stream						
37. What quantities of pollutants will be emitted from the boiler before control?	Pollutant	Tons per Year	lb/hr	gpm/ACF	SI Y	PSIA
CO						
hydrocarbons						
NO _x						
PM ₁₀						
PM _{2.5}						
SO ₂						
SO _x						
Other (specify)						

38. What quantities of pollutants will be emitted from the boiler after control?	Pollutant	Tons per Year	lb/hr	gpm/ACF	SI Y	PSIA
CO						
hydrocarbons						
NO _x						
PM ₁₀						
PM _{2.5}						
SO ₂						
SO _x						
Other (specify)						

39. How will waste material from the process and control equipment be disposed of?
 Stack 100% Landed?

40. Have you completed an Air Pollution Control Device (APCD) for the controls used on this Emission Unit. Y

41. Have you included the air pollution data on the Emissions Profile Data Summary Sheet?

42. **Process Monitoring, Recordkeeping, Reporting and Testing**
 Please describe monitoring, recordkeeping, and reporting in order to demonstrate compliance with the proposed operating parameters. Please propose testing in order to demonstrate compliance with the proposed emissions limits.

MONITORING PLAN: Please list (1) the process parameters and how they were chosen (2) the ranges and how they were established for monitoring to demonstrate compliance with the operator of the process equipment operator or air pollution control device.
 Weekly and monthly operations according to maintenance specifications. Inspections to assist in preventing leaks.

TESTING PLAN: Please describe any proposed emissions testing for this process equipment or air pollution control device.
 Per routine.

RECORDKEEPING: Please describe the proposed recordkeeping that will accompany the monitoring data.
 Manual.

REPORTING: Please describe the proposed frequency of reporting of the recordkeeping.
 All weekly and monthly operations along with any malfunctions.

43. Describe all operating ranges and maintenance procedures required by manufacturer to maintain warranty.
 The gas cleanup system will be maintained via instrumentation for any errors or Malfunctions requiring maintenance. Requiring regular maintenance procedures to be provided by manufacturer.

Synthesis Gas Cleaning System

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**Attachment L
EMISSIONS UNIT DATA SHEET
GENERAL**

To be used for affected sources other than asphalt plants, furnaces, incinerators, indirect heat exchangers, and boilers.
Identification Number (as assigned on Equipment List Form): 1131

1. Name or type and model of proposed affected source: Dye Furniture
2. On a separate sheet(s), furnish a sketch(es) of this affected source. If a modification is to be made to the source, clearly indicate the change(s). Provide a narrative description of all feature(s) of the affected source which may affect the production of air pollutants.
3. Name(s) and maximum amount of proposed process material(s) charged per hour: 1.112 x 10 ⁶ lb/hr
4. Name(s) and maximum amount of proposed material(s) produced per hour: 7.798 x 10 ⁶ lbs/week
5. Give chemical reactions, if applicable, that will be involved in the generation of air pollutants: N/A

* The identification number which appears here must correspond to the air pollution control device identification number appearing on the List Form.

6. Combustion Data (if applicable): (a) Type and amount in appropriate units of fuel(s) to be burned: 76 cfm of natural gas for 8 hrs during start up then will use syngas
(b) Chemical analysis of proposed fuel(s), excluding coal, including maximum percent sulfur and ash: Natural gas Syngas is 90% dry in composition
(c) Theoretical combustion air requirement (ACF based on 23.1°C): 29440 @ 63 °F and 14.7 psi.
(d) Percent excess air: 100%
(e) Type and BTU/hr of burners and all other firing equipment planned to be used: 1 burner
(f) If coal is proposed as a source of fuel, identify supplier and seams and give sizing of the coal as it will be fired: N/A
(g) Proposed maximum design heat input: 4.75 x 10 ⁷ BTU/hr.
7. Projected operating schedule: Hours/Day: 14 Days/Week: 1 Weeks/Year: 0

8. Projected amount of pollutants that would be emitted from this affected source if no control devices were used:				
#	NA	Y and	ppm	
a. NO _x	NA	NA	lb/hr	NA grams/ACF
b. SO _x	NA	NA	lb/hr	NA grams/ACF
c. CO	NA	NA	lb/hr	NA grams/ACF
d. PM ₁₀	NA	NA	lb/hr	NA grams/ACF
e. Hydrocarbons	NA	NA	lb/hr	NA grams/ACF
f. VOCs	NA	NA	lb/hr	NA grams/ACF
g. Pb	NA	NA	lb/hr	NA grams/ACF
h. Specify other(s)			lb/hr	grams/ACF
			lb/hr	grams/ACF
			lb/hr	grams/ACF
			lb/hr	grams/ACF

NOTE: (1) An Air Pollution Control Device Sheet must be completed for any air pollution devices used to control emissions from this affected source.
(2) Complete the Emission Points Data Sheet.

9. Proposed Monitoring, Recordkeeping, Reporting, and Testing	
Please propose monitoring, recordkeeping, and reporting to demonstrate compliance with the proposed operating parameters. Please propose testing in order to demonstrate compliance with the proposed emission limits.	
MONITORING Provide all essential inspections according to all manufacturer specifications.	RECORDKEEPING Provide all essential and Modify inspection along with any manufacturer specifications.
REPORTING Provide instructions.	TESTING Provide request.
<p>MONITORING. PLEASE LIST AND DESCRIBE THE PROCESS PARAMETERS AND RANGES THAT ARE PROPOSED TO BE MONITORED IN ORDER TO DEMONSTRATE COMPLIANCE WITH THE OPERATION OF THIS PROCESS EQUIPMENT OPERATING/POLLUTION CONTROL DEVICE.</p> <p>RECORDKEEPING. PLEASE DESCRIBE THE PROPOSED RECORDKEEPING THAT WILL ACCOMPANY THE MONITORING.</p> <p>REPORTING. PLEASE DESCRIBE THE PROPOSED FREQUENCY OF REPORTING OF THE MONITORING DATA.</p> <p>TESTING. PLEASE DESCRIBE ANY PROPOSED EMISSIONS TESTING FOR THE PROCESS EQUIPMENT/POLLUTION CONTROL DEVICE.</p> <p>TO DESCRIBE: all operating ranges and maintenance procedures required by manufacturer to maintain safe and healthy.</p> <p>Responsible maintenance (fabrication) do not require shut down. Major maintenance issues to involve a shut-down.</p>	



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TECHNICAL FILE
(Area-030)
Dryer for Power Plant

DRYER PLANT
(Area 030)
Equipment - MEDRECYCLER - 030

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Equipment No. General 030

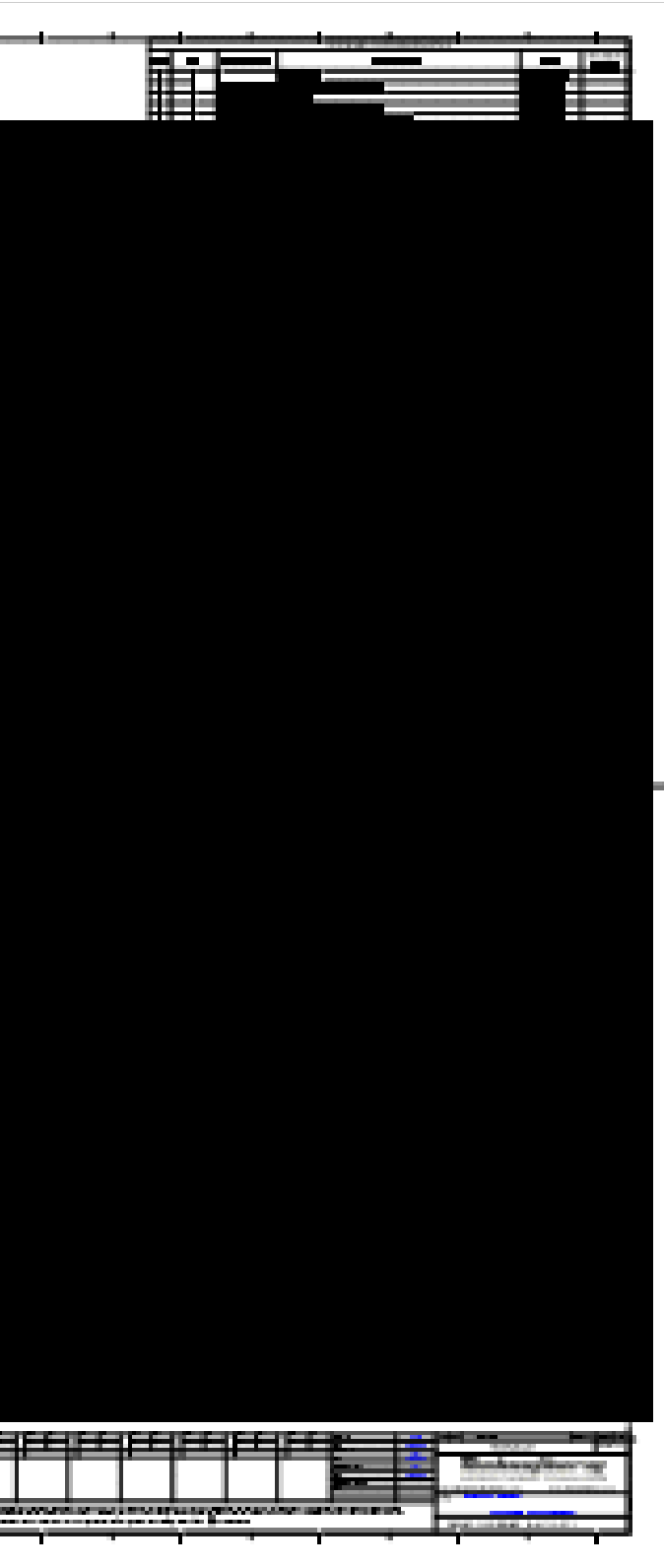
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**Attachment 1
EMISSIONS LIMIT DATA SHEET
GENERAL**

To be used for affected sources other than asphalt plants, foundries, incinerators, indirect heat exchangers, and quarries.
Identification Number (as assigned on Equipment List Form): 100

1. Name or type and model of proposed affected source:
Meyers/Bender
Medical Waste Incinerator by Truhotburn

2. On a separate sheet(s), furnish a sketch(es) of this affected source. If a modification is to be made to the source, clearly indicate the change(s). Provide a narrative description of all features of the affected source which may affect the production of air pollutants.

3. Name(s) and maximum amount of proposed process material(s) charged per hour:
2.54E7 lb/hr

4. Name(s) and maximum amount of proposed material(s) produced per hour:
2.54E7 lb/hr

5. Give chemical reactions, if applicable, that will be involved in the generation of air pollutants:
N/A

* The identification number which appears here must correspond to the air pollution control device identification number appearing on the L&E Form.

6. Combustion Data (if applicable):

(a) Type and amount in appropriate units of fuel(s) to be burned:
N/A

(b) Chemical analysis of proposed fuel(s), excluding coal, including maximum percent sulfur and ash:
N/A

(c) Theoretical combustion air requirement (ACF/unit of fuel):
N/A @ °F and psia.

(d) Percent excess air: N/A

(e) Type and BTU/hr of burners and all other firing equipment planned to be used:
N/A

(f) If coal is proposed as a source of fuel, identify supplier and seams and give sizing of the coal as it will be fired:
N/A

(g) Proposed maximum design heat input: N/A × 10⁶ BTU/hr.

7. Projected operating schedule:

Hours/Day	24	Days/Week	7	Weeks/Year	52
-----------	----	-----------	---	------------	----

8. Projected amount of pollutants that would be emitted from this affected source if no control devices were used:

	N/A	°F and	psia
a. NO _x	N/A	lb/hr	N/A grams/ACF
b. SO ₂	N/A	lb/hr	N/A grams/ACF
c. CO	N/A	lb/hr	N/A grams/ACF
d. PM ₁₀	N/A	lb/hr	N/A grams/ACF
e. Hydrocarbons	N/A	lb/hr	N/A grams/ACF
f. VOCs	N/A	lb/hr	N/A grams/ACF
g. Pb	N/A	lb/hr	N/A grams/ACF
h. Specify other(s)		lb/hr	grams/ACF
		lb/hr	grams/ACF
		lb/hr	grams/ACF
		lb/hr	grams/ACF

NOTE: (1) An Air Pollution Control Device Sheet must be completed for any air pollution device(s) used to control emissions from this affected source.
(2) Complete the Emission Points Data Sheet.

D. Proposed Monitoring, Recordkeeping, Reporting, and Testing

Please propose monitoring, recordkeeping, and reporting in order to demonstrate compliance with the proposed operating parameters. Please propose testing in order to demonstrate compliance with the proposed emissions levels.

<p>MONITORING Wastely and weekly inspection according to manufacturer specifications.</p>	<p>RECORDKEEPING All Weekly and Monthly inspection along with any malfunctions.</p>
--	--

<p>REPORTING Per malfunction.</p>	<p>TESTING Per all emissions associated with equipment.</p>
--	--

MONITORING: PLEASE LIST AND DESCRIBE THE PROCESS PARAMETERS AND RANGES THAT ARE PROPOSED TO BE MONITORED IN ORDER TO DEMONSTRATE COMPLIANCE WITH THE OPERATION OF THIS PROCESS EQUIPMENT OPERATING RANGE POLLUTION CONTROL DEVICE.

RECORDKEEPING: PLEASE DESCRIBE THE RECORDKEEPING THAT WILL ACCOMPANY THE MONITORING.

REPORTING: PLEASE DESCRIBE THE PROPOSED FREQUENCY OF REPORTING OF THE RECORDKEEPING.

TESTING: PLEASE DESCRIBE ANY PROPOSED EMISSIONS TESTING FOR THE PROCESS EQUIPMENT AND POLLUTION CONTROL DEVICE.
TO ENSURE ALL OPERATING RANGES AND MAINTENANCE PROCEDURES REQUIRED BY MANUFACTURER IS PROVIDED VARIABLY.
OPERATION TO TAKE PLACE AT 4000PSIA



<p>DATE: 03/20/17</p>	<p>PROJECT: 133555A</p>	<p>REV: 001</p>	<p>DESCRIPTION: [Redacted]</p>	<p>APPROVED: [Redacted]</p>	<p>DATE: 03/20/17</p>	<p>Technotherm SUPERIOR THERMAL TECHNOLOGY 10011 SOUTHWEST 100TH AVENUE, SUITE 100 MIRAGE RECYCLER - R3 SHREDDING AREA MEDICAL WASTE SHREDDING GENERAL ARRANGEMENT</p> <p>AL 0:50 1031-020-001 1/1 A</p>
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**Attachment L
EMISSIONS UNIT DATA SHEET
BULK LIQUID TRANSFER OPERATIONS**

Furnish the following information for each new or modified bulk liquid transfer area or loading rack as shown on the **Equipment List Form** and other parts of this application. This form is to be used for bulk liquid transfer operations such as to and from drums, marine vessels, rail tank cars, and tank trucks.

Identification Number (as assigned on **Equipment List Form**)

1. Loading Area Name:

2. Type of cargo vessels accommodated at this rack or transfer point (check as many as apply):
 Drums Marine Vessels Rail Tank Cars Tank Trucks

3. Loading Rack or Transfer Point Data:

Number of pumps	
Number of liquids loaded	2
Maximum number of marine vessels, tank trucks, tank cars, and/or drums loading at one time	1

4. Does ballasting of marine vessels occur at this loading area?
 Yes No Does not apply

5. Describe cleaning location, compounds and procedure for cargo vessels using this transfer point.

6. Are cargo vessels pressure tested for leaks at this or any other location?
 Yes No

If YES, describe:

7. Projected Maximum Operating Schedule (for rack or transfer point as a whole):

Maximum	Jan - Mar	Apr - June	July - Sept	Oct - Dec
hours/day				
days/week				

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weeks/quarter				
---------------	--	--	--	--

8. Bulk Liquid Data (add pages as necessary):

Pump ID No:	
Liquid Name:	HCL
Max. duty throughput (1000 gals/hr):	5K
Max. annual throughput (1000 gal/yr):	1500
Loading Method ¹ :	Sub
Max. Fill Rate (gpm/hr):	1000
Average Fill Time (min/loading):	500
Max. Bulk Liquid Temperature (°F):	75
Train Vapor Pressure ² :	
Cargo Vessel Condition ³ :	
Control Equipment or Method ⁴ :	
Minimum capture efficiency (%):	
Maximum Emission Rate:	
Loading (lb/hr):	
Annual (lb/yr):	
Estimation Method ⁵ :	
¹ BF = Bottom Fill SF = Splash Fill SUB = Submerged Fill	
² At maximum bulk liquid temperature	
³ B = Ballasted Vessel; C = Cleaned; U = Uninsulated (dedicated service); O = other (describe)	
⁴ List as many as apply (complete and submit appropriate Air Pollution Control Device): CDSB = Cold Carbon Adsorption LCB = Low-CO Adsorption/CO = Condensation SC = Scrubber (Absorber/DRA = Compression- Refrigeration/Adsorption TD = Thermal Oxidizer or Incinerator CR = Compression-Refuge/adsorption-Condensation VB = Dedicated Vapor Balance (closed system) O = other (describe)	
⁵ EPA = EPA Emission Factor as stated in AP-42	
MS = Material Balance	

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TM = Test Measurement based upon test data submitted
 O = other (describe)

Proposed Monitoring, Recordkeeping, Reporting, and Testing Please propose monitoring, recordkeeping, and reporting in order to demonstrate compliance with the proposed operating parameters. Please propose testing in order to demonstrate compliance with the proposed emissions levels.	
MONITORING	RECORDKEEPING
REPORTING	TESTING

MONITORING: PLEASE LIST AND DESCRIBE THE PROCESS PARAMETERS AND RANGES THAT ARE PROPOSED TO BE MONITORED IN ORDER TO DEMONSTRATE COMPLIANCE WITH THE OPERATION OF THE PROCESS EQUIPMENT OPERATIONAL POLLUTION CONTROL DEVICE.

RECORDKEEPING: PLEASE DESCRIBE THE PROPOSED RECORDKEEPING THAT WILL ACCOMPANY THE MONITORING.

REPORTING: PLEASE DESCRIBE THE PROPOSED FREQUENCY OF REPORTING OF THE RECORDKEEPING.

TESTING: PLEASE DESCRIBE ANY PROPOSED EMISSIONS TESTING FOR THE PROCESS EQUIPMENT/POLLUTION CONTROL DEVICE.

10. Describe all operating ranges and maintenance procedures required by Manufacturer to maintain air quality:

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ATTACHMENT M

Air Pollution Control Device Sheet

101

ATTACHMENT L

FUGITIVE EMISSIONS FROM UNPAVED HIGHWAYS

UNPAVED HIGHWAYS (INCLUDING EQUIPMENT TRAFFIC) IN PAVED AREAS, INTERCHANGES, ETC.

Q = Pavement slip resistance	0.80	0.50
Q = Soil content of road surface material (%)		
Q = Number of days per year with precipitation > 0.25 in.		

Item Number	Description	Material Weight (lb/ft ²)	Mean Vehicle Weight (lb)	Mean Vehicle Speed (mph)	Material Type	Material Type per Foot	Material Type per Year	Control Device #	Control Efficiency (%)
1									
2									
3									
4									
5									
6									
7									
8									

Source: AP-42 5th Edition - 12.2.2 Unpaved Roads
 $E = 1.4 \times 10^{-4} \times W \times (1 + 0.01 \times (W - 10) + 0.0001 \times W^2) \times (1 + 0.01 \times (Q - 10)) \times (1 + 0.01 \times (S - 10))$ to Vehicle Mile Traveled (VMT)
 Where: E = lb/yr

Q = Pavement slip resistance	0.80	0.50
Q = Soil content of road surface material (%)		
Q = Mean vehicle speed (mph)		
W = Mean vehicle weight (lb)		
W = Mean number of axles per vehicle		
Q = Number of days per year with precipitation > 0.25 in.		

For lb/yr: $E = VMT \times [PMF \times W] \times [(1 + 0.01 \times (W - 10))] \times [(1 + 0.01 \times (Q - 10))] \times [(1 + 0.01 \times (S - 10))]$
 For TSP: $E = VMT \times [PMF \times W] \times [(1 + 0.01 \times (W - 10))] \times [(1 + 0.01 \times (Q - 10))] \times [(1 + 0.01 \times (S - 10))]$

SUMMARY OF UNPAVED HIGHWAY EMISSIONS

Item No.	Uncontrolled		Controlled		Uncontrolled		Controlled	
	lb/yr	TPY	lb/yr	TPY	lb/yr	TPY	lb/yr	TPY
1								
2								
3								
4								
5								
6								
7								
8								
TOTALS								

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FUGITIVE EMISSIONS FROM PAVED HIGHWAYS

INDUSTRIAL PAVED HIGHWAYS (INCLUDING EQUIPMENT TRAFFIC) IN PAVED AREAS, INTERCHANGES, ETC.

Q = Industrial augmentation factor (dimensionless)		
Q = Number of traffic lanes		
Q = Surface material soil content (%)		
Q = Surface dust loading (lb/ton)		
Q = Average vehicle weight (lb)		

Item Number	Description	Material Weight (lb/ft ²)	Material Type	Material Type per Foot	Material Type per Year	Control Device #	Control Efficiency (%)
1							
2							
3							
4							
5							
6							
7							
8							

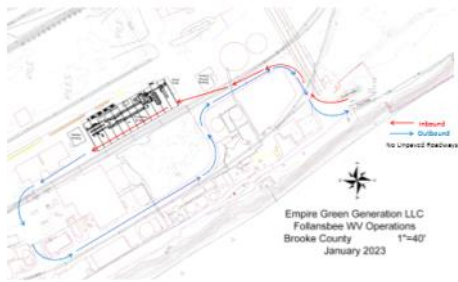
Source: AP-42 5th Edition - 11.2.5 Industrial Paved Roads
 $E = 0.077 \times 1 \times (1 + 0.01 \times (W - 10) + 0.0001 \times W^2) \times (1 + 0.01 \times (Q - 10)) \times (1 + 0.01 \times (S - 10))$
 Where: E = lb/yr

For lb/yr: $E = VMT \times [PMF \times W] \times [(1 + 0.01 \times (W - 10))] \times [(1 + 0.01 \times (Q - 10))] \times [(1 + 0.01 \times (S - 10))]$
 For TSP: $E = VMT \times [PMF \times W] \times [(1 + 0.01 \times (W - 10))] \times [(1 + 0.01 \times (Q - 10))] \times [(1 + 0.01 \times (S - 10))]$

SUMMARY OF PAVED HIGHWAY EMISSIONS

Item No.	Uncontrolled		Controlled	
	lb/yr	TPY	lb/yr	TPY
1				
2				
3				
4				
5				
6				
7				
8				
TOTALS				

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**Attachment B
Air Pollution Control Device Sheet**

Control Device ID No. (must match Emission Units Table): 1006
PLANS 2111103

Equipment Information	
1. Manufacturer: Endurotherm Model No.: To Be Determined	<input type="checkbox"/> Vertical <input checked="" type="checkbox"/> Horizontal <input type="checkbox"/> Other Stack/Exhaustory: None
3. Provide diagram(s) of unit describing capture system with duct arrangement and size of duct, air volume, capacity, temperature of exhaust. If applicable, state dust free velocity and heat recovery efficiency.	
4. Method of system wash: <input checked="" type="checkbox"/> Steam-washed <input checked="" type="checkbox"/> Air-washed <input type="checkbox"/> Physico-chemical <input type="checkbox"/> Non-washed	
5. Maximum capacity of flow: 15,239 scfm scfh	6. Dimensions of duct: Diameter: 2.36 ft Height: 46.71 ft
7. Estimated combustion efficiency (Waste gas destruction efficiency): Estimated: >98 % Minimum guaranteed: >98 %	8. Fuel used in burners: <input checked="" type="checkbox"/> Natural Gas <input type="checkbox"/> Fuel Oil, Number _____ <input type="checkbox"/> Other, specify: But Gas
9. Number of burners: 1 Rating: 12,247 BTU/hr	11. Describe method of controlling flame:
10. Will preheat be used? <input type="checkbox"/> Yes <input checked="" type="checkbox"/> No	12. Natural gas flow rate to three pilot flames per pilot light: scfh 0.20 13.0
12. Flue height: 46.71 ft	14. Will adjustable regulation be used? <input checked="" type="checkbox"/> Yes <input type="checkbox"/> No
13. Flue to inside diameter: 2.36 ft	15. If automatic regulation will be used, describe the method.
14. Number of pilot lights: 1 Type: 12,247 BTU/hr	16. An electrical ignition assembly will be used to ignite the gases and including igniter coils and an electrical control assembly to provide a spark. The assembly will be employed to light the pilot. A pilot flame thermocouple and a stack flame thermocouple will monitor the systems and provide the spark for re-ignition when necessary.
18. Is pilot flame equipped with a monitor? <input checked="" type="checkbox"/> Yes <input type="checkbox"/> No If yes, what type? <input checked="" type="checkbox"/> Thermocouple <input type="checkbox"/> Infrared <input type="checkbox"/> Ultra Violet <input type="checkbox"/> Camera with monitoring control room <input type="checkbox"/> Other, describe:	19. Hours of unit operation per year: 24 hours per day; 7 days per week

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Steam Injection			
20. Will steam injection be used? <input type="checkbox"/> Yes <input checked="" type="checkbox"/> No	21. Steam pressure: 100 psig Steam Control: Manual Inlet location: Top		
22. Total Steam flow rate: 1.874	23. Temperature: 7		
24. Velocity: None	25. Number of jet streams:		
26. Diameter of steam jets: 0	27. Design data for steam injection: 8 steam/18 steam/20 steam		
28. How will steam flow be controlled if steam injection is used?			
Characteristics of the Waste Gas Stream to be Burned			
29. Name	Quantity (lb/hr or 1000 gal)	Quantity (lb/hr or 1000 gal)	Source of Material
Hydrocarbons	THD		
Cyanide Compounds	THD		
Acrolein	THD		
CO	THD		
Hydrogen	THD		
H ₂ S	100%		
30. Calculate total combustible flow: 60,000 gal/hr	31. Calculate total flow rate of waste gas: 1.874	32. Calculate total flow rate to flow including materials to be burned, carrier gases, auxiliary fuel, etc.: 1.874	33. Gross composition of carrier gases: THD 50ppm
34. Temperature of entrance stream: 550.0 °F max	35. Identify and describe all auxiliary fuels to be burned: Not Applicable	36. Identify and describe all auxiliary fuels to be burned: Not Applicable	37. Flue gas heat content: 700 BTU/lb
35. Heating value of waste gas stream: THD 87000	38. Flue gas flow rate: 11,000 scfm	39. Flue gas flow rate: 11,000 scfm	40. Flue gas flow rate: 11,000 scfm
36. Heat rate of waste gas stream: 100 BTU/lb	41. Flue gas flow rate: 11,000 scfm	42. Flue gas flow rate: 11,000 scfm	43. Flue gas flow rate: 11,000 scfm
37. Temperature of flue gas: 712 °F	44. Flue gas flow rate: 11,000 scfm	45. Flue gas flow rate: 11,000 scfm	46. Flue gas flow rate: 11,000 scfm
38. Flue gas heat content: 700 BTU/lb	47. Flue gas flow rate: 11,000 scfm	48. Flue gas flow rate: 11,000 scfm	49. Flue gas flow rate: 11,000 scfm
39. Heat rate of waste gas stream: 100 BTU/lb	50. Flue gas flow rate: 11,000 scfm	51. Flue gas flow rate: 11,000 scfm	52. Flue gas flow rate: 11,000 scfm
40. Flue gas flow rate: 11,000 scfm	53. Flue gas flow rate: 11,000 scfm	54. Flue gas flow rate: 11,000 scfm	55. Flue gas flow rate: 11,000 scfm
41. Flue gas flow rate: 11,000 scfm	56. Flue gas flow rate: 11,000 scfm	57. Flue gas flow rate: 11,000 scfm	58. Flue gas flow rate: 11,000 scfm
42. Flue gas flow rate: 11,000 scfm	59. Flue gas flow rate: 11,000 scfm	60. Flue gas flow rate: 11,000 scfm	61. Flue gas flow rate: 11,000 scfm
43. Flue gas flow rate: 11,000 scfm	62. Flue gas flow rate: 11,000 scfm	63. Flue gas flow rate: 11,000 scfm	64. Flue gas flow rate: 11,000 scfm
44. Flue gas flow rate: 11,000 scfm	65. Flue gas flow rate: 11,000 scfm	66. Flue gas flow rate: 11,000 scfm	67. Flue gas flow rate: 11,000 scfm
45. Flue gas flow rate: 11,000 scfm	68. Flue gas flow rate: 11,000 scfm	69. Flue gas flow rate: 11,000 scfm	70. Flue gas flow rate: 11,000 scfm
46. Flue gas flow rate: 11,000 scfm	71. Flue gas flow rate: 11,000 scfm	72. Flue gas flow rate: 11,000 scfm	73. Flue gas flow rate: 11,000 scfm
47. Flue gas flow rate: 11,000 scfm	74. Flue gas flow rate: 11,000 scfm	75. Flue gas flow rate: 11,000 scfm	76. Flue gas flow rate: 11,000 scfm
48. Flue gas flow rate: 11,000 scfm	77. Flue gas flow rate: 11,000 scfm	78. Flue gas flow rate: 11,000 scfm	79. Flue gas flow rate: 11,000 scfm
49. Flue gas flow rate: 11,000 scfm	80. Flue gas flow rate: 11,000 scfm	81. Flue gas flow rate: 11,000 scfm	82. Flue gas flow rate: 11,000 scfm
50. Flue gas flow rate: 11,000 scfm	83. Flue gas flow rate: 11,000 scfm	84. Flue gas flow rate: 11,000 scfm	85. Flue gas flow rate: 11,000 scfm
51. Flue gas flow rate: 11,000 scfm	86. Flue gas flow rate: 11,000 scfm	87. Flue gas flow rate: 11,000 scfm	88. Flue gas flow rate: 11,000 scfm
52. Flue gas flow rate: 11,000 scfm	89. Flue gas flow rate: 11,000 scfm	90. Flue gas flow rate: 11,000 scfm	91. Flue gas flow rate: 11,000 scfm
53. Flue gas flow rate: 11,000 scfm	92. Flue gas flow rate: 11,000 scfm	93. Flue gas flow rate: 11,000 scfm	94. Flue gas flow rate: 11,000 scfm
54. Flue gas flow rate: 11,000 scfm	95. Flue gas flow rate: 11,000 scfm	96. Flue gas flow rate: 11,000 scfm	97. Flue gas flow rate: 11,000 scfm
55. Flue gas flow rate: 11,000 scfm	98. Flue gas flow rate: 11,000 scfm	99. Flue gas flow rate: 11,000 scfm	100. Flue gas flow rate: 11,000 scfm

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<p>16. Proposed Monitoring, Recordkeeping, Reporting, and Testing Please describe monitoring, recordkeeping, and reporting in order to demonstrate compliance with the proposed operating parameters. Please propose testing in order to demonstrate compliance with the proposed emission limits.</p>	
<p>MONITORING: Monitoring will be accomplished using the control panel. The facility personnel will conduct weekly inspections for visible stack emissions and odors. The stack emission observations will not be required to be performed by a person certified as a qualified observer under 40 CFR Method for Visual Determination of Visibility of Emission from Stationary Sources.</p>	<p>RECORDKEEPING: Record all spots on the system Visual inspection reports will be maintained on site in the control room.</p>
<p>REPORTING: Report to WYDEP all spots sent to the State.</p>	<p>TESTING: TBD</p>
<p>MONITORING: Please list and describe the process parameters and ranges that are proposed to be monitored in order to demonstrate compliance with the operation of the process equipment or air control device. RECORDKEEPING: Please describe the proposed recordkeeping that will accompany the monitoring. REPORTING: Please describe any proposed emissions testing for the process equipment or air pollution control device. TESTING: Please describe any proposed emissions testing for the process equipment or air pollution control device.</p>	
<p>15. Manufacturer's Guaranteed Capture Efficiency for each air pollutant (40%):</p>	
<p>16. Manufacturer's Guaranteed Control Efficiency for each air pollutant (40%):</p>	
<p>17. Describe all operating ranges and maintenance procedures required by Manufacturer to maintain warranty. See attached document.</p>	



Ionisationspilotsbrenner
 Ionization pilot burner
 Brûleur pilote à ionisation
 ZAI

Kromschroder

Ionisationspilotsbrenner ZAI

7 von 10 Bildern (1) - 10 Bilder
 1 von 10 Bildern (1) - 10 Bilder
 2 von 10 Bildern (1) - 10 Bilder

Anwendung
 Gasbrenner, Gasrohr und Air-Strahlrohr
 Anwendung in den Abgasanlagen, Gas-Boiler, etc. Anwendung für Industrie, Gas-Boiler, Gas-Heizung, etc.

Ionization pilot burner ZAI

7 von 10 Bildern (1) - 10 Bilder
 1 von 10 Bildern (1) - 10 Bilder
 2 von 10 Bildern (1) - 10 Bilder

Application
 Gas burner, Gas pipe and Air jet
 Application in the exhaust systems, Gas boiler, etc. Application for industry, Gas boiler, Gas heating, etc.

Brûleur pilote à ionisation ZAI

7 von 10 Bildern (1) - 10 Bilder
 1 von 10 Bildern (1) - 10 Bilder
 2 von 10 Bildern (1) - 10 Bilder

Application
 Gaz brûleur, Gaz tuyau et Air jet
 Application dans les systèmes d'échappement, Gaz boiler, etc. Application pour l'industrie, Gaz boiler, Gaz chauffage, etc.

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Introduction

Project Overview

The Project is a commercial Plant using advanced conversion technology in the nature of a waste acceptance facility using plastics with moisture content of 35% from which a clean syngas is produced and subsequently combusted in reciprocating engine generator sets to produce electricity for export.

The Project comprises of a macerator, rotary thermal dryer, 2 sets of pyrolyzers, 3 sets of syngas clean-up equipment, a hydrochloric acid recovery system, gas stealer (barges accumulators), reciprocating engine generators, thermal oxidiser, a stack and flare. The Facility is a 3 shift plant using 24000 tonnes of plastic to renewable energy power plant producing baseload electricity for export to the grid. The Project is operated on a continuous basis and is designed with sufficient plant redundancy to negate any single point issue (SPP) failures.

Waste heat is recovered from the exhaust of the engines and the Pyrolyser to dry the plastic once shredded. Site available natural gas provides start-up and standby thermal energy. The plant design and configuration is comprised of equipment that results in low life cycle costs, high operational efficiency and operational flexibility consistent with the results of the design reviews, RAH and HAZOP studies conducted during the project execution.

Supplier

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Plant and Equipment

This instruction manual relates to a WASTE TO ENERGY PLANT Scheme.

The design is for 800°C maximum process temperature.

It essentially comprises of a

- Shredder
- Dryer
- Pyrolyzers
- Hydrochloric acid recovery system
- Gas clean-up system
- Gas temporary storage
- Engines
- Regenerative thermal oxidiser
- Stack and flare

Operating Philosophy

Overall process takes plastics, received by a transporting company, and thermally processes it in a pyrolysis system operating at 800°C - 900°C (1,472°F - 1,652°F). Organic matter from the plastics is evaporated forming a syngas that can directly be used as a fuel source for electrical generating engines. Oil and tar are produced where the oil is recycled through the pyrolysis system to make more syngas, and the tar is used to heat a vitrification system where solids from the process are vitrified and made inert. Exhaust from the engines are sent to a drying unit, where the plastics are dried prior to being introduced into the pyrolysis system. All gases are sent to a Thermal Oxidiser operating at 850°C (1,562°F), after which they are conditioned for release to atmosphere via a stack.

Shredder Section

Plastic received at the facility will be in sealed containers about 2 feet square and they are not opened. Each box has an inventory, so Empire Green Generation knows what is in the box prior to being fed into the Pyrolysis System. Plastics in boxes are not stored on site but are processed as they arrive after unloading the waste.

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Sealed containers will be fed into the Macerator. In the Macerator, operating under negative pressure, the plastics in sealed containers will be reduced to 25 mm or less.

Drying Section

The drying of the feedstock is carried out in a direct heated, parallel flow, rotary twin drum type dryer using a combination of engine exhaust and, if necessary during start-up or unusual operating conditions, syngas and, as a last resort, natural gas.

Feedstock is transported from the dryer feed hopper into the dryer by means of a screw conveyor. Upon entering the dryer inner rotator, the moist feedstock comes into direct contact with the parallel stream of hot flue gases.

Liners and progression plates ensure intimate contact between the feedstock and flue gas therefore facilitating efficient drying and movement of feedstock along the rotator.

Once both feedstock and flue gases reach the closed end of the dryer they are discharged from the inner rotator into the outer rotator and return to the entry end of the dryer, discharging 10% moisture level feedstock into an expansion chamber. Coarse dry feedstock falls to the bottom of the chamber forming a heap on the conveyor located beneath.

The flue gas exhaust, contaminated with light particulate feedstock material, is also discharged from the expansion chamber and ducted to a bank of cyclones where separation occurs. Fine particulate falls to the bottom of each cyclone and is discharged via rotary valves into a common screw conveyor. The screw conveyor discharges the product onto the conveyor joining the dryer exit material and the product. This conveyor transfers the combined dry feedstock streams onto a conveyor feeding an intermediate storage hopper that feeds both pyrolyzers.

HCl Recovery System

This system is fed with dried feedstock from a thermal dryer via a live bottom screw hopper. Material is fed from a gas-tight, storage hopper into the horizontal pre-pyrolyser rotator by a rotary screw. The feedstock passes through a pre-pyrolyser chamber where it is

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heated in an inert environment to allow gradual release of Chlorine. The gas is removed and condensed to form hydrochloric acid where after it is stored.

Pyrolyser

The pyrolyser train is fed dried and partially reconstructed feedstock from the HCl recovery system as described in the previous section. The pyrolysis train consists of two identical pyrolysers. For unit, the source of indirect heat is primarily hot exhaust flue gas from a furnace located beneath the pyrolysis retort. These hot flue gases exit the pyrolysis retorts and then progress to the medium grade heat applications (Thermal dryer and during start up syngas cooler and tar condenser). Supplemental heating of the pyrolysis retort is being provided by firing a portion of the cleaned syngas. Natural gas is available for initial start-up or any start-up where insufficient syngas is available. After passing through the dryers the flue gas is progressed to the thermal oxidiser.

As the material passes through the pyrolysis retort, it undergoes thermal degradation releasing volatile organic syngas compounds that is discharged from the retort. The crude syngas off-takes are collected into a common manifold that transfers the syngas to the syngas cleaning system.

The heavier particles, mainly consisting of ash and fixed carbon, collect in a specially designed high temperature de-acceleration chamber where the particles are collected and returned to the furnace for energy recovery.

Ash and carbonaceous residue produced by the pyrolysers drops off the dust from the aforementioned de-acceleration chamber screw conveyors, together with the main residue collected from the base of the pyrolyser into a refractory lined furnace fired by recovered tars (described below in the syngas cleaning equipment). The heat (provided by burning the tars and gas) is sufficient to heat the ash from the pyrolysis units above their eutectic temperature with oxides, preheated air to burn off the tars. The char is completely combusted into CO₂ and H₂O.

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Syngas Clean-up

Particulate Matter Collection

Raw syngas is removed from the pyrolysis retorts, as described above, and passes through a de-acceleration chamber screw conveyors, together with the main residue collected from the base of the pyrolyser into a refractory lined furnace fired by recovered tars (described below in the syngas cleaning equipment). The heat (provided by burning the tars and gas) is sufficient to heat the ash from the pyrolysis units above their eutectic temperature with oxides, preheated air to burn off the tars. The char is completely combusted into CO₂ and H₂O.

Syngas Coolers

The partially cleaned, still hot, flue gas flows next through stainless steel tubular syngas coolers. The cooler is in essence a heat exchanger which indirectly transfers heat from syngas to the combustion air heaters.

Tar Condensers

The syngas from the coolers described above flows to a stainless-steel shell & tube heat exchanger/cooler that is cooled by an air blower system. Tars are condensed out and drop into heated troughs, the heat source of which is engine exhaust. Hot condensed tar is pumped to the purification furnaces described above. The common insulated spore set of coolers take engine exhaust flue gas to heat the heat exchangers and thereby cause the tars to drop into the heated trough below.

Oil Scrubbers

The syngas from the tar condensers described above flows to a scrubber with inter-connected oil and water separator.

Washing Scrubbers

From the oil scrubber the syngas flows through a high pressure drop venturi scrubber, which is kept at a set pH to neutralise the gas before progressing to temporary storage.

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Gas Bladder (Syngas Storage Tank)

The syngas storage tank provides surge capacity of cleaned syngas to level out flow and composition variations. The bladder is contained within a demarcated area. The bladder will operate with an internal pressure of 30 to 40 millibar gauge.

Thermal Oxidiser

All flue gases enter a thermal oxidiser comprising of a rectangular box shaped furnace. The internal dimensions are determined by the total volume that needs to be raised to 820°C and maintained for 2 seconds.

Stack and Flare

The stack and flare comprises of the following:

1. Induced Draft Fan
2. Flare Stack (combined with plant stack) and
3. Plant Stack (5 m above nearest building x 700 mm dia.)

The treated hot gases progress through the stack and disperse into the atmosphere after passing through the thermal oxidiser.

Gas Engines

Each syngas engine is a fully packaged unit complete with all associated components and auxiliaries. These engines are of robust design and have been proven on low and medium calorific value gas fuels.

The engine package allows the engine to start, synchronise, operate continuously at base or part load and shut down.

The syngas engines are situated in containers as indicated on the plant layout and engine cooling will occur by means of external radiators.

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Maintenance

Scheduled Maintenance

It is assumed that the plant will be subject to the same general routine maintenance discipline, in respect of cleanliness, readiness control etc. as the other plant and equipment in the factory.

General

- Work to be done must be cleared with the operator or his designee before commencement.
- Ensure the area is clean and free of contamination.
- Inspect labels and warning signs location, clear visibility and damage. Repair / replace if necessary.
- Inspect the equipment for any signs of built up or deposits.
- Check that all fasteners and mounting hardware is in place.
- Always stand to the side when observing interior or opening the Pyrolyser to avoid sudden exposure to heat.
- Frequent visual inspection of the equipment should be done. Any leaks, in piping, tanks, equipment casings, covers and all associated equipment or loose connections must be reported.
- If any fault occurs, analyse and permanently remove the cause. Do not remain on temporary repair.
- Immediately report any signs of abnormal equipment operation or unusual instrument readings.

Workshop Report 001

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- After maintenance, ensure that all bolts, fittings, guards and other fasteners are correctly tightened.
- Metal parts must be painted to avoid corrosion. Where painting is not possible suitable oil or grease must be utilized.

Maintenance Tasks

The maintenance schedule specifies the frequency of the inspections and checks that are expected under normal operating conditions. In the event that the prevailing conditions are abnormal, appropriate adjustments could be expected.

Weekly Maintenance

- Check seal integrity on front and rear pre-Pyrolyser and Pyrolyser bellows
- Check seal integrity on knife gate valves
- Check limit switches and/or proxy switches ensuring holding bolts are tight
- Check strike arms on limit switches are secure
- Check thermocouples are secure
- Check for uneven movements / misalignment of mechanisms
- Check temperature controller and over temperature controller for proper operation
- Check retort rollers are tight and no excessive wear is taking place
- Grease wheels with high temperature graphite grease
- Listen for undue mechanical noise from the installation, investigate and repair if necessary
- Check all seals and gaskets for possible leaks on the hydraulic system.
- Check that all bearings are properly greased and operating correctly.

Monthly Maintenance

- Repeat weekly scheduled maintenance
- Check all bolts are tight, tighten if necessary

Workshop Report 001

Technotherm

SUPERIOR THERMAL TECHNOLOGY

Supertherm Pyrolytic TECHNOHERM

Headquarter Office Park - 11 Pölla (topographisch) Street
Grazed House, Block A, Meppenfeld 1498
E-mail: info@technotherm.com | Fax: +43 (0) 684 3403
URL: www.technotherm.com | Reg. No.: 201426643/2
GR Trade Number: 2017 76 311 4334

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- Ensure all guards, louvers, brackets are in place and secure.
- Visually inspect modules and insulation blankets are properly in position and secure, repair / replace if necessary.
- Check ducts for foreign materials causing obstruction.
- Check structure steelwork for signs of corrosion and paint damage. Metal parts must be painted to avoid corrosion.
- Check blower impeller by hand to ensure free rotation. Check that there is not fouling between rotating and stationary components
- Check that all blower fasteners are secure and that all components are in good order at the cooling section.
- Ensure that the blower guard is in place and secure.
- Keep electric motors' air inlets and outlets free and clean. The air blown out by the motor shall not enter again. The distance between the air inlet and the wall must be approximately 1/4 of the inlet opening diameter.
- Check retort mechanism main track roller for alignment
- Check all pre-Pyrolyser and Pyrolyser in feed system for any possible obstructions and charge cleanliness
- Check knife gate valves and proxy switch positions
- Check all pre-Pyrolyser and Pyrolyser front and rear door seal
- Check all booster fan cleanliness and rotation direction.

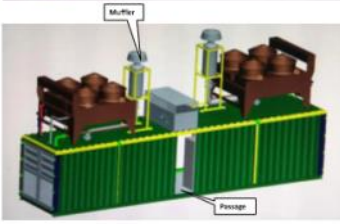
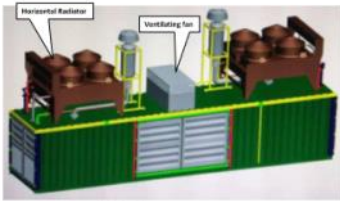
Three Monthly Maintenance

- Repeat Monthly Maintenance
- Clean booster fan and clean all interconnecting pipes

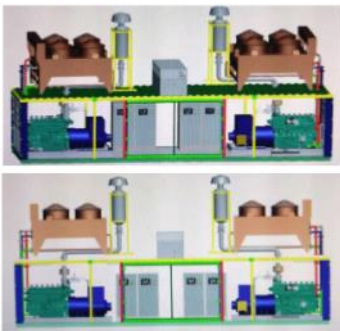
Six Monthly Maintenance

- Repeat Three Monthly Maintenance
- Booster Fan Maintenance:

Workshop Report 001



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Container size (L*W*H) 11.5m*2.2m*2.45m
3 containers in total.

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**Attachment to
Air Pollution Control Device Sheet
(NET CO2 EMISSION SYSTEMS)**

Control Device ID No. (Must match Emission Units Table)

Manufacturer: Pulp Processing	Equipment Information			
	1. Wet/dry	2. Packed Bed	3. Spray Tower	4. Venturi
Model No. 7' X 7' Scrubber	<input checked="" type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
		<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
		<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>
		<input type="checkbox"/>	<input type="checkbox"/>	<input type="checkbox"/>

3. Provide diagram(s) of unit describing wet/dry systems with duct arrangement and type of duct, air volume, capacity, horsepower of motors. If applicable, state flow face velocity and filter collection efficiency.

6. Provide a scale diagram of the collector showing internal construction. Please include ducting type and size, spray configurations, baffle plates, and mist eliminators.

8. Describe the scrubber's construction materials. Polyethylene (PE) tank with PVC electrical and control piping. Water with NaOH is added to the vessel above the wet scrubber. Clean gas from the PE tank passes through the gas diffuser where HCl fumes are neutralized before the scrubbed gas is cooled from the top.

7. What will be the power requirements of the collector?
Size: 7x7 ft. Motor: 1/2 HP

8. What type of filter(s) will be used?
Type of filter: None. Number of tanks: None. Diameter of tank: None in. Also specify a filter class for each fan to be used.

9. Estimated gas pressure drop of measure flow rate: 1 in. H₂O

10. Scrubbing Liquid Characteristics

Composition	Weight %	11. Scrubbing liquid inlet temperature, °F	12. Liquid pressure to scrubber: 8.25 PSIG
1. Water	96	65	13. Pressure drop through scrubber: 6 in. H ₂ O
2. NaOH	4		
3.			
4.			
14. Source of liquid (scrubber): Sump liquid added to tank		15. Liquid flow rate to scrubber: Design maximum: 1,100 gal/min Average expected: 500 gal/min	

16. Describe systems to be used to supply liquid to collector. Manual draw and no. of pumps.

17. Give the expected sulfur content of the liquor. No suspended solids will be present. The liquor is 10 percent HCl gas vapor will react with NaOH to form water (H₂O) and salt (NaCl). Water will be charged out with before NaOH concentration reaches saturation.

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18. If the liquor is to be recirculated, describe any treatment performed:

19. Data for Venturi Scrubber:
 Throat Diameter: NA
 Density: NA
 Throat Velocity: NA ft/sec

20. Data for Packed Towers:
 Type of Packing: NA
 Superficial Gas Velocity through Bed:

Gas Stream Characteristics

21. Gas flow into the collector:
 Into ACF @ 30 °F and 34.2 PSIA

22. Gas stream temperature:
 Inlet: surface °F
 Outlet: surface °F

23. Gas flow rate:
 Design Maximum: 110 ACFM
 Average Expected: 47 ACFM

24. Particulate Grain Loading in grains/ft³:
 Inlet: NA
 Outlet: NA

25. Emission rate of each pollutant (quantity) into and out of collector:

Pollutant	In		Out		Guaranteed Minimum Collection Efficiency
	lb/hr	grams/ft ³	lb/hr	grams/ft ³	
A131	0.0000		0.0000		99
B					
C					
D					
E					

26. Type of pollutant(s) controlled: SO_x NO_x
 Particulate Matter Other: HCl

27. Is what method used for the uncontrolled emissions calculation? Material Balance Stack Test
 Pilot Test Other:

28. Dimensions of stack: Height: 4'-0" ft. Diameter: 0.3' ft.

29. Supply an equilibrium curve and/or solubility data (at various temperatures) for the proposed system.

30. Supply a curve showing proposed collection efficiency versus gas volume from 25 to 100 percent of design rating of collector.

Particulate Distribution

31. Complete the table:

Particulate Size Range (microns)	Particulate Size Distribution of Inlet to Collector Height h ₁ for Size Range % versus total unfiltered emission	Fraction Efficiency of Collector Height h ₂ for Size Range
0-2		
2-4		
4-6		
6-8		
8-10		
10-15		
15-20		
20-30		
30-45		
45-60		
60-80		
80-100		
100-150		
150-200		
200-300		
300-400		
400-500		
500-600		
600-800		
800-1000		
>1000		

32. Describe any air pollution control device that will utilize gas conditioning processes (e.g., gas cooling, gas reheating, gas humidification).

33. Describe the collection material disposal system.
 Should HCl gas will become a salt water that will be disposed of in accordance with local, state (PA/DEP), and federal regulations.

34. Have you included a **Wet Collecting (Scrubber) Control Device** in the Emissions Points Data Summary Sheet? Yes

35. **Process Monitoring, Recordkeeping, Reporting, and Testing**
 Please propose monitoring, recordkeeping, and reporting in order to demonstrate compliance with the proposed operating parameters. Please propose testing in order to demonstrate compliance with the proposed emissions limits.

MONITORING:	RECORDKEEPING:
REPORTING:	TESTING:

MONITORING: Please list and describe the process parameters and ranges that are proposed to be monitored in order to demonstrate compliance with the operation of this process equipment or air control device.

RECORDKEEPING: Please describe the proposed recordkeeping that will accompany the monitoring.

REPORTING: Please describe any proposed emissions testing for this process equipment or air pollution control device.

TESTING: Please describe any proposed emissions testing for this process equipment or air pollution control device.

36. Manufacturer's **Guaranteed Capture Efficiency** for each air pollutant.

37. Manufacturer's **Guaranteed Control Efficiency** for each air pollutant.

38. Describe of operating ranges and maintenance procedures required by Manufacturer to maintain warranty.

**Attachment L
EMISSIONS UNIT DATA SHEET
BULK LIQUID TRANSFER OPERATIONS**

Provide the following information for each new or modified bulk liquid transfer area or loading rack as shown on the **Equipment List Form** and other parts of this application. This form is to be used for bulk liquid transfer operations such as to and from drums, marine vessels, rail tank cars, and tank trucks.

Identification Number (as assigned on **Equipment List Form**)

1. Loading Area Name: _____

2. Type of cargo vessels accommodated at this rack or transfer point (check as many as apply)
 Drums Marine Vessels Rail Tank Cars Tank Trucks

3. Loading Rack or Transfer Point Data:

Number of pumps	
Number of liquids loaded	2
Maximum number of marine vessels, tank trucks, tank cars, and/or drums loading at one time	1

4. Does ballasting of marine vessels occur at this loading area?
 Yes No Does not apply

5. Describe cleaning location, compounds and procedure for cargo vessels using this transfer point.

6. Are cargo vessels pressure tested for leaks at this or any other location?
 Yes No

If YES, describe:

7. Projected Maximum Operating Schedule (for rack or transfer point as a whole)

Maximum	Jan - Mar	Apr - June	July - Sept	Oct - Dec
Hours/Day				
Days/Week				

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Weeks/quarter				
---------------	--	--	--	--

8. Bulk Liquid Data (600 pages at necessary):

Pump ID No.					
Liquid Name	HCL				
Max. daily throughput (1000 gal/day)	5k				
Max. annual throughput (1000 gal/yr)	1300				
Loading Method ¹	sub				
Max. Fill Rate (gal/min)	1000				
Average Fill Time (min/loading)	500				
Max. Bulk Liquid Temperature (°F)	75				
Tran Vapor Pressure ²					
Cargo Vessel Condition ³					
Control Equipment or Method ⁴					
Minimum control efficiency (%)					
Maximum Emission Rate	<table border="1"> <tr><td>Loading (lb/hr)</td><td></td></tr> <tr><td>Annual (lb/yr)</td><td></td></tr> </table>	Loading (lb/hr)		Annual (lb/yr)	
Loading (lb/hr)					
Annual (lb/yr)					
Estimation Method ⁵					

¹ BF = Bottom Fill; SP = Sprayer Fill; SLB = Submerged Fill

² At maximum bulk liquid temperature

³ B = Ballasted Vessel; C = Cleaned; U = Uncleaned (dedicated service); O = other (describe)

⁴ List as many as apply (complete and submit appropriate Air Pollution Control Device Check) CA = Carbon Adsorption; COA = Laminar Airflow/COA = Condensation; SC = Scrubber (Absorber)/CRA = Compressor-Radiant-Adsorption; CD = Thermal Oxidation or Incineration; CRC = Compressor-Radiant-Combustion; VSB = Dedicated Vapor Balance (closed system) or other (describe)

⁵ EPA = EPA Emission Factor as stated in AP-42
 MB = Material Balance

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TM = Test Measurement based upon test data submitted
 O = other (describe)

9. Proposed Monitoring, Recordkeeping, Reporting, and Testing

Please provide monitoring, recordkeeping, and reporting in order to demonstrate compliance with the proposed operating parameters. Please propose testing in order to demonstrate compliance with the proposed emission limits.

MONITORING	RECORDKEEPING
REPORTING	TESTING

MONITORING: PLEASE LIST AND DESCRIBE THE PROCESS PARAMETERS AND RANGES THAT ARE PROPOSED TO BE MONITORED IN ORDER TO DEMONSTRATE COMPLIANCE WITH THE OPERATION OF THIS PROCESS EQUIPMENT OPERATIONAL POLLUTION CONTROL DEVICE.

RECORDKEEPING: PLEASE DESCRIBE THE PROPOSED RECORDKEEPING THAT WILL ACCOMPANY THE MONITORING.

REPORTING: PLEASE DESCRIBE THE PROPOSED FREQUENCY OF REPORTING OF THE RECORDKEEPING.

TESTING: PLEASE DESCRIBE ANY PROPOSED OPERATIONAL TESTING FOR THIS PROCESS EQUIPMENT/ AIR POLLUTION CONTROL DEVICE.

10. Describe all operating ranges and maintenance procedures required by Manufacturer to maintain warranty:

Page ___ of ___ WYDOP-DAQ Revision 03-2007

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151 Falmouth Avenue Pottsville, PA 17242 Email: SALES@VSYSTEMS.COM Telephone: 412-228-9200 Fax: 412-228-9182

Quote #2023-560-H-R

July 28, 2023

Mr. Faley R. Wood, P.E. Vice President of Engineering Empire Diesel/Fuel Energy 1400 Main Street Falmouth, WV 26037

Subject: Chemical System - Scrubber - Quote #2023-560-H-R

Dear Mr. Wood,

Please find attached our proposal for the above referenced equipment project. We appreciate the opportunity to provide a quote for the opportunity.

You will also find our most recent line card attached for your reference. I hope you will think of us during your next project. If you would have any questions or require additional information, please give us a call at 412-228-9200.

Sincerely,

Russell C. Huffstader President & CEO V-Systems, Inc. 151 Falmouth Avenue Pottsville, PA 17242

Enclosure
enr

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151 Falmouth Avenue Pottsville, PA 17242 Email: SALES@VSYSTEMS.COM Telephone: 412-228-9200 Fax: 412-228-9182

Quotation:

Project Name: Chemical System - Scrubber Contact Name: Faley, Wood
Company Name: Empire Diesel/Fuel Energy Email: russell@empirediesel.com
Phone: 304-243-2600
Address/Street: 1400 Main Street Date: July 21, 2023 Revised July 28, 2023
City/State/Zip: Falmouth, WV 26037 Quote Number: 2023-560-H-R

Thank you for the opportunity to provide you with the following quote:

Table with 4 columns: Quantity, Item Description, Unit Price, Total Price. Includes items like Polyethylene Scrubber Tank and 10,000-Gallon Vertical Tank.



151 Falmouth Avenue Pottsville, PA 17242 Email: SALES@VSYSTEMS.COM Telephone: 412-228-9200 Fax: 412-228-9182

Table with 4 columns: Quantity, Item Description, Unit Price, Total Price. Includes items like March Model #TS-10K-MD SPN and Kymen Variable Frequency Drive.

QUOTED BY: RUSSELL HUFFSTADER
Total Quoted Amount: \$195,093.00

If you need further information concerning the products that have been included in the quote, please feel free to contact me at 412-228-9200 and/or russell@empirediesel.com.

We appreciate the opportunity to provide you with this quote and look forward to working with you on this important project.

Thank you,

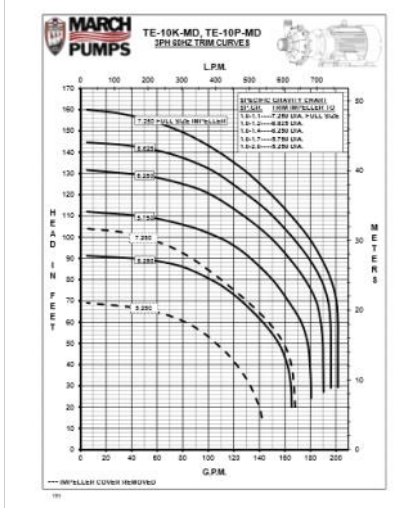
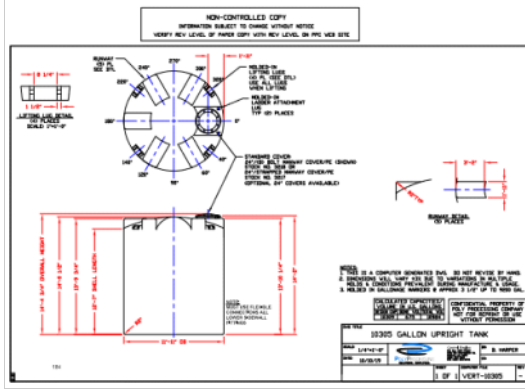
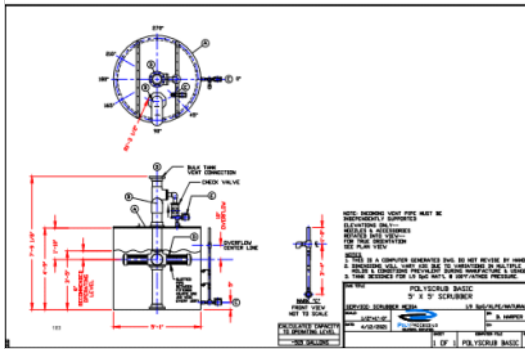
Russell C. Huffstader President & CEO enr

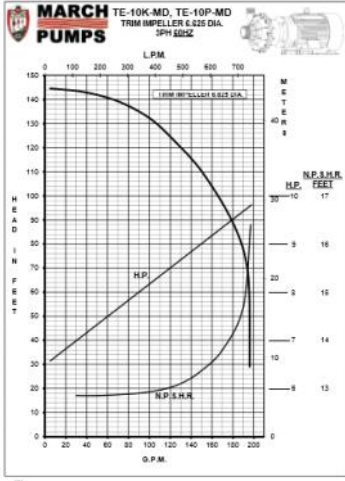
THIS QUOTATION ON SELLER'S ACCEPTANCE OF THIS ORDER IS EXPRESSLY LIMITED TO, AND EXPRESSLY MADE CONDITIONAL ON BUYER'S ACCEPTANCE OF THE V-SYSTEMS TERMS, STANDARD TERMS AND CONDITIONS OF SALE. A COPY OF THESE TERMS AND CONDITIONS IS AVAILABLE AT <http://www.vsystems.com/Products/ProductsPages.aspx>. SELLER OBJECTS TO ANY DIFFERENT OR ADDITIONAL TERMS.

Special Conditions
Noting Buyer's Sales & Service Terms and Conditions if the following are met:
 • Equipment installed per industry standards and manufacturer instruction manual.
 • Operation of equipment in accordance with manufacturer instruction manual.
 • Maintenance and lubrication per manufacturer instruction manual. Note, maintenance log showing dates required.
 • Equipment must be stored per manufacturer instruction manual and protected from the weather.
 Payment terms apply. V-Systems needs to be contacted in writing before any repairs are made, whereas a mutual course of action will be performed. Equipment cannot be disassembled without V-Systems being present.
 Acknowledged and Accepted by Buyer:

Name _____ Title/Company _____ Yes _____ No _____
 Signature _____ PO # _____
 Date _____ Ship To _____

IF APPLICABLE, please send a copy of your company's tax-exempt form. Otherwise, our accounting department will assume that the order is taxable.





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 - Flexible Design
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- Additional Series:**
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 - Continuous, Non-Controlling, & Non-Controlling Series
 - Flexible & Versatile Design
 - Motorized, Hydraulic, 42 Fuel Oil, Three-Phase Fuel Oil & 42 Oil
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- Additional Options:**
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 - Acoustic Enclosures
 - Custom Engineered Systems

Lochinvar

- Micro Series:**
 - Models from 25,000 to 60,000 Btu/hr
 - Up to 95% Thermal Efficiency
 - Up to 20:1 Turndown Ratio
 - Remote CO₂ & O₂ Sensors
 - Connectivity and SmartModem™ Operating Control
- Domestic Industrial Series:**
 - Models from 100,000 to 1,000,000 Btu/hr
 - Up to 95% Thermal Efficiency
 - Up to 20:1 Turndown Ratio
 - Direct Vent Fuel Efficiency up to 95.5%
 - SmartModem™ Operating Control and CO₂ & O₂ Sensor Control
- Additional Options:**
 - High Efficiency Condensing Tube Bundle
 - Non-Condensing Burner
 - Pool Heaters
 - Water Pools Storage Tanks
 - Analysis Equipment
 - Custom Engineered Systems

ARMSTRONG

- Performance Series:**
 - Up to 90% Thermal Efficiency
 - Up to 20:1 Turndown Ratio
 - Direct Vent Fuel Efficiency up to 95.5%
 - SmartModem™ Operating Control and CO₂ & O₂ Sensor Control
- Additional Options:**
 - High Efficiency Condensing Tube Bundle
 - Non-Condensing Burner
 - Pool Heaters
 - Water Pools Storage Tanks
 - Analysis Equipment
 - Custom Engineered Systems

Parts and Service: When you need service, reach us at Commercial Parts. It's essential to us to provide the best service professionals who not only understand the capabilities of your equipment but the complex nature of your business. V-Systems provides comprehensive and professional service & repair alignment for pumps, boilers, equipment, and controls.

Contact: www.v-systems.com
1-800-858-6222

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- Products:** Boiler Feed Units, Condensate Pump Systems, De-aerators, Vacuum Systems, and Heat Traps
- Performance:** 1/2" HP to 50 HP

CRANE PUMPS & SYSTEMS

- Products:** Submersible Non-Dug Pumps, Effluent Pumps, Grinder Pumps, Vertical and Electric-Driven Self-Priming Pumps, Horizontal Pumps, Transfer Pumps, and Pump Pans
- Performance:** 1/4" HP to 100 HP

SEPRIC THERMO

- Products:** Air & Oil Separators, Air Vents, Oil/Gas/Steam Separator, Deaerators, and Industrial Separators
- Size:** 1/2" to 24"

ENERVEX

- Products:** Fans & Ventilators, Exhaust Fans & Heat Exchangers, Draft Control, Chimney & Grease Duct, Controls & Accessories
- Size:** 1" to 24"

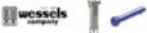
SECURITY

- Products:** Industrial storage containers for heating boilers, domestic water boilers, and low to high temperature hydronic heating applications
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BOILERS & WATER HEATERS	OTHER
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ELECTRICAL	Contact:
Andor Scientific Baker Vacuum (ABE) C&D Controls Carter Hess Motor Corporation Ohio Electric Control, Inc.	301 Parkway Ave. Middletown, OH 45128 www.v-system.com www.v-system.com 413-628-0000
TANKS, BASINS, & SEPARATORS	QR CODE
American (Hessco) Compartments Solutions B&I of America Highland Tank John Reed Company Niles Steel Tank Spartan TDFP Industries Thames Company	
VALVES	PLEASE CALL OUR OFFICE FOR ALTERNATE MANUFACTURERS AND OTHER OPTIONS
Apex Valve Bo-Top Hessco (Hessco) Hessco Valve Co. Hessco Controls Hessco Valve Ohio Valve Company Sun-Pac Company The Valve Company	
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About Us	Facilities & Equipment	Bottom Line
Warren Design & Build is a multi-disciplined service firm and engineering design firm that company. We specialize in emergency response, pipe and materials, as well as building system installation.	Located in Warren Ohio, WDB's 100,000 square foot complex facility houses a variety of engineering, machining, assembly, and finishing capabilities. Our large assembly area can handle everything from a single section to a full assembly line with all the available tooling and equipment. We have a full range of tooling and equipment for all types of projects. We have a full range of tooling and equipment for all types of projects.	With WDB's unique combination of service, design, engineering, and manufacturing capabilities, we can help you with all aspects of your project. We have a full range of tooling and equipment for all types of projects. We have a full range of tooling and equipment for all types of projects.
Our team is comprised of 40 professionals, 4 welders, 4 machinists, and 4 electrical engineers with a focus on P&E and computer systems programming capability.	Located in the Northeast Ohio Manufacturing Corridor, we also have the capability to manufacture many special processes that can be brought off-site or start-up in order to avoid design periods in special projects.	With years of Northeast Ohio manufacturing, engineering, and highly trained engineering and manufacturing staff, we can help you with all aspects of your project. We have a full range of tooling and equipment for all types of projects. We have a full range of tooling and equipment for all types of projects.



204 Chambers Street
Middletown, PA 17056 (412) 773-5160

About Us	Facilities & Equipment	Bottom Line
All things mechanical, electrical, machine and design is a multi-disciplined service firm and engineering design firm that company. We specialize in emergency response, pipe and materials, as well as building system installation.	Located in Middletown PA, we have a 100,000 square foot complex facility houses a variety of engineering, machining, assembly, and finishing capabilities. Our large assembly area can handle everything from a single section to a full assembly line with all the available tooling and equipment. We have a full range of tooling and equipment for all types of projects. We have a full range of tooling and equipment for all types of projects.	With WDB's unique combination of service, design, engineering, and manufacturing capabilities, we can help you with all aspects of your project. We have a full range of tooling and equipment for all types of projects. We have a full range of tooling and equipment for all types of projects.
Our team is comprised of 40 professionals, 4 welders, 4 machinists, and 4 electrical engineers with a focus on P&E and computer systems programming capability.	Located in the Northeast Ohio Manufacturing Corridor, we also have the capability to manufacture many special processes that can be brought off-site or start-up in order to avoid design periods in special projects.	With years of Northeast Ohio manufacturing, engineering, and highly trained engineering and manufacturing staff, we can help you with all aspects of your project. We have a full range of tooling and equipment for all types of projects. We have a full range of tooling and equipment for all types of projects.



- Emergency Repair:** We fix all electro-mechanical devices. We have field tech service and in-house staff that will repair all forms of equipment.
- Water Jet:** We have a water jet for cutting all types of material in almost any shape.
- Design Engineering:** Senior mechanical designers. Each is ready to provide support from concept to build, refinement and rollout of the solution to your manufacturing or operational bottleneck.
- Electrical Engineering:** Electrical engineers ready to design electrical schemes, write control programming, and provide data analysis.
- Robotics Cell Design and Assembly:** Project design & management of all needs and programming related to robot cells.
- Machining:** We offer a wide variety of precision machining capabilities, including CNC turning and milling, general machining, grinding, and boring mill work. Our machining capabilities may be used for both short run and sustained production work.
- Fabrication:** We can meet all your prototyping and production needs with complete sheet metal and welding services that include quick delivery on small quantity prototypes, close tolerance fabrications and high volume production runs.
- Assembly:** We can meet your entire contract manufacturing needs with our diverse assembly capabilities and wide range of electrical, mechanical, and electro-mechanical assembly services, ranging from special, one-of-a-kind automation machinery to ongoing, low to medium volume production assembly. Our assembly capability includes turn-key production of final-to-gate assembly work, including full management of key suppliers. WDM is capable of handling full production volumes.
- Painting:** Using both oil and water based materials, our state-of-the-art paint shop can produce finishes from Class 1A to MIL-Spec, including textured finishes color matched to your specifications. Powder coat finishes also available.

MECHANICAL AUTOMATION SYSTEMS <ul style="list-style-type: none"> Automated assembly systems Automated material handling Automated inspection systems Automated test systems 	COMPLEX PRECISION MACHINING <ul style="list-style-type: none"> High speed machining Micro-machining Grinding Boring Turning Milling Drilling Electro-discharge machining Water jet cutting 	ON-SITE AUTOMATION SERVICE <ul style="list-style-type: none"> Automated assembly systems Automated material handling Automated inspection systems Automated test systems
PRECISION BUILDING <ul style="list-style-type: none"> Sheet metal fabrication Welding Structural steel Machine shops 	COMPLETE PRECISION DESIGN <ul style="list-style-type: none"> 3D CAD 2D CAD Reverse engineering Prototyping Simulation 	AUTOMATION <ul style="list-style-type: none"> PLC programming Robotics programming Machine vision Industrial networking

ATTACHMENT N

Supporting Emission Calculations

4. OVERALL EMISSION CALCULATIONS ALL DEVICES

Plastic waste has significant variation in form and quantity. It seems the best way to describe it as a heterogeneous mixture of solids and semi-solids. Literature review showed a. References to the paper are given at the end of the section. This table a heterogeneous plastic waste composition based on the is required in Table 2. Municipal Solid Waste (MSW) data by Simpson, et al., is given for comparison purposes.

Table 2. Range of Heterogeneous Properties of "Plastic"

Plastic Waste Composition	Range Plastic Range (wt. %)
C	50.8
H	9
O	25.1
N	1.0
S	0.20
Cl	1.2
Ca	
Mg	21.64
MSW (Plastic)	

Composition of the plastic waste and the synthesis output are shown in Table 3. Pyrolysis emissions are also shown for plastic waste by Garbhanjan et al. by Technomic.

Table 3. Plastic Composition and Pyrolysis Output

Component	Plastic Composition (wt. %)	Element	Pyrolysis Component	Pyrolysis Synthesis (wt. %)	Pyrolysis Synthesis (wt. %)	Pyrolysis Synthesis (wt. %)
C	50.8	C	C	1.64	1.64	1.20
H	9	H	H	0.20	0.20	0.15
O	25.1	O	O	0.07	0.07	0.05
N	1.0	N	N	0.01	0.01	0.01
S	0.20	S	S	0.00	0.00	0.00
Cl	1.2	Cl	Cl	0.00	0.00	0.00
Ca		Ca	Ca	0.00	0.00	0.00
Mg	21.64	Mg	Mg	0.00	0.00	0.00
Si		Si	Si	0.00	0.00	0.00
Al		Al	Al	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Na		Na	Na	0.00	0.00	0.00
K		K	K	0.00	0.00	0.00
As		As	As	0.00	0.00	0.00
Se		Se	Se	0.00	0.00	0.00
Br		Br	Br	0.00	0.00	0.00
I		I	I	0.00	0.00	0.00
B		B	B	0.00	0.00	0.00
Y		Y	Y	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cd		Cd	Cd	0.00	0.00	0.00
Ce		Ce	Ce	0.00	0.00	0.00
Pr		Pr	Pr	0.00	0.00	0.00
Ta		Ta	Ta	0.00	0.00	0.00
Bi		Bi	Bi	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Mb		Mb	Mb	0.00	0.00	0.00
Ba		Ba	Ba	0.00	0.00	0.00
Li		Li	Li	0.00	0.00	0.00
Rb		Rb	Rb	0.00	0.00	0.00
Sr		Sr	Sr	0.00	0.00	0.00
Zr		Zr	Zr	0.00	0.00	0.00
Ag		Ag	Ag	0.00	0.00	0.00
Hf		Hf	Hf	0.00	0.00	0.00
V		V	V	0.00	0.00	0.00
Ti		Ti	Ti	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0.00	0.00	0.00
Cr		Cr	Cr	0.00	0.00	0.00
Fe		Fe	Fe	0.00	0.00	0.00
Mn		Mn	Mn	0.00	0.00	0.00
Co		Co	Co	0.00	0.00	0.00
Ni		Ni	Ni	0.00	0.00	0.00
Cu		Cu	Cu	0.00	0.00	0.00
Zn		Zn	Zn	0.00	0.00	0.00
Pb		Pb	Pb	0.00	0.00	0.00
Sn		Sn	Sn	0.00	0.00	0.00
W		W	W	0.00	0.00	0.00
Mo		Mo	Mo	0		

Pyrolysis organics and solids can vary about 30% or greater based on the input feed. An attempt was made to produce a syngas composition where some of the elements and compounds were near the middle and greater than the middle of the feed composition shown in Table 2. Interpretation was shown as individual compounds instead of just the term "gas".

Once a syngas was established within the variations described, the next step was to combine these available compounds through the Engine, Vector and Thorpe (Vector) Stoichiometric combustion equations as shown below:

$$2C2D + O2 \rightarrow 2CO2$$

$$2C2H6 + 7O2 \rightarrow 4CO2 + 6H2O$$

$$2C2H4 + 3O2 \rightarrow 2CO2 + 4H2O$$

$$2C2H2 + 5O2 \rightarrow 4CO2 + 2H2O$$

Reasons for all devices are shown in Figure 2 (see Attachment 2, and of the TMS) and are shown as an expanded Block Flow Diagram in the following section. The only equipment for the pyrolysis process system shall be 70 U.S. tons per day and will operate twenty-four (24) hours a day, seven (7) days a week, for 310 days per year. This equates to 90% availability for processing equipment. All calculations are based on 70 tons/day which equates to one hour of reactor-feed load. All major charaters have the design life of twenty (20) years before replacement.

Throughput calculations are as follows:
Calculated Throughput
 • Hourly Throughput (2,666.710 kg/hr) (2,294.46kg = 5,883.33 t/yr)
 • Annual Throughput
 = 2,666.710 kg/hr x 24 hr/day x 310 days/yr = 1,989,000 kg = 21,888.98 tonnes/yr = 21,780 Ton/year

Table 3. Hourly, Daily and Annual Biomass Outputs from model results (Figure 2)

Output/Component	Hourly	Daily	Annual	US Tons/yr
CO2	7,666.81	183,999.84	1,989,000.00	23,981.05
H2O	17,251.87	414,045.00	4,354,672.50	51,461.18
H2	20,861.55	500,677.20	5,263,062.50	63,241.38
CH4	283.10	6,794.40	71,231.25	855.21
CO	3.84 x 10 ²	9,216.00	96,768.00	1,161.84
H2O	6.14 x 10 ²	1,4736.00	153,726.00	1,844.72
H2	1.18 x 10 ³	28,320.00	296,760.00	3,563.52
CO	1.74 x 10 ³	41,760.00	435,720.00	5,228.64
H2O	4.29 x 10 ³	103,104.00	1,072,620.00	12,871.44
CH4	6.03 x 10 ³	144,720.00	1,509,540.00	18,114.48
CO	8.84 x 10 ³	212,160.00	2,227,620.00	26,731.44
H2O	3.30 x 10 ⁴	792,000.00	8,307,000.00	99,684.00
H2	4.93 x 10 ⁴	1,183,200.00	12,373,500.00	148,506.00
TOTAL	31,427.58	756,255.24	7,962,187.77	95,786.62

Table 4. Biomass Outputs

Component	Hourly	Daily	Annual	US Tons/yr
CO2	32.26	773.84	7,962.19	95.79
H2O	51.26	1,230.24	12,762.24	153.15
H2	146.24	3,509.76	36,457.44	437.49
CH4	49.74	1,193.76	12,373.50	148.51
CO	40.26	966.24	10,000.50	120.01
H2O	129.54	3,108.96	32,362.50	388.35
TOTAL	716.52	1,719.34	17,886.21	214.62

CALCULATION WORKSHEET DATE: 9/6/2023

Pyrolysis Syngas Emissions

Component	mol %	Molecular Weight	60°F, 14.7psia Density (lb/ft ³)	wt%	normalized	lb/hr	lb/day	lb/yr	kg/yr
Oxygen	0.18	32	0.0044	0.18	0.37	7.77	185.60	0	0
Carbon monoxide	10.00	28.01	0.0785	9.72	19.45	400.14	9,603.36	103,236.00	1,238,832.00
Hydrogen	12.00	2	0.000104	0.74	1.54	31.37	752.88	8,434.08	99,009.12
Hydrogen	2.00	28.01	0.0785	1.74	3.48	70.54	1,692.96	18,622.56	223,470.72
Overall	25.18	136.672	0.0854	46.65	100.00	2092	49,813.4	562,337.6	6,702,812.8

Source: Provided by Technochem. See Overall Emission Calculations

Gas Yield	30% mass
Feed Input (lb/hr)	16,600.00
Gas Produced	4,980.00

Note: All Gases will be routed to the engine and/or RTO prior to release into the atmosphere.
 Details: <https://www.technochem.com/our-services/>

6. Conversion Factors:
 35.31 ft³/m³
 0.4536 kg/lb
 1,000,000 Btu/MMBtu
 1,000,000 Btu/MMBtu
 60 sec/hr
 7440 hours/yr (based on 310 days per yr as given by Technochem)
 2,204.62 kg/ton
 2000 lb/ton

Calculations
 # Calculate the estimated amount of syngas produced from the pyrolysis in tpd.

$$PR_{\text{syngas}} = \text{mass}_{\text{pyrolysis}} / (CF_{\text{gas}} \times 2000) \times (PR_{\text{pyrolysis}})$$

$$= \frac{1,989,000 \text{ kg}}{100 \text{ kg}} \times \frac{1}{2000 \text{ lb}} \times 25 \frac{\text{ton}}{\text{day}} = 25 \text{ tpd syngas}$$
 # Calculate the maximum hourly emission rate for the components ER_{max}.
 Using CO as an example.

$$ER_{\text{maxCO}} = (mW_{\text{CO}}) / (CF_{\text{CO}} \times 2000) \times (PR_{\text{CO}}) \times (CF_{\text{gas}}) \times (CF_{\text{pyrolysis}})$$

$$= \frac{0.37 \text{ mol}}{100 \text{ mol}} \times \frac{1}{2000 \text{ lb}} \times 25 \frac{\text{ton}}{\text{day}} \times \frac{1}{1 \text{ ton}} \times 24 \text{ hrs} = 7.77 \text{ lb/hr CO}$$
 Calculations (continued)

3.2.1.2.1 Engine Green Generation Florida Pyrolysis Facility - Engine Emissions from Pyrolysis

3. Calculate the annual average emission rate for the components CR_{annual}
 Using CO as an example:
 $CR_{annual} = CR_{base} \times (CF_{fuel}) \times (CF_{type})$

$$= \frac{1}{1} \times \frac{7.77}{1} \times \frac{1.00}{1} \times \frac{1}{1} = 7.77 \text{ lbs/hr}$$

4. Calculate the heat value of the engine for each component in Btu/lb
 Using CO as an example:
 $HV_{CO} = (CR_{CO}) \times (HV_{CO}) \times (CF_{type})$

$$= \frac{1}{1} \times \frac{10,222}{1} \times \frac{1}{1} \times \frac{1}{1} = 10,222 \text{ Btu/lb}$$

5. Calculate the gross calorific heat value of the engine in Btu/hr
 $Heat_{gross} = CR_{annual} \times HV_{CO}$

$$= \frac{1}{1} \times \frac{7.77}{1} \times \frac{10,222}{1} = 79,445 \text{ Btu/hr}$$

**Engine Green Generation, LLC
 Florida Recycling Pyrolysis Facility
 Generator Emissions Calculations**

Parameter	Value
Number of Engines	4
Control Date	08/20/23
Gross Calorific Value	13,174.00 Btu/lb
Annual Amount of Gas Produced	1,000,000 lbs
Amount of Gas Produced per Hour	114.63 lbs/hr
Amount of Gas in Each Engine	28.66 lbs/hr
Fuel Consumption	1,000,000 lbs
Density	0.0044 lbs/lb
Engine Rating	250 HP
Rated Power	425 kW
Annual Operating Hours	7,000 hours
Fuel Heat Value	13,174 Btu/lb
NOx Factor	1.00
CO Factor	1.00
VOC Factor	0.10
PM10 Factor	0.05
PM2.5 Factor	0.05
SO2 Factor	0.10
CO2 Factor	1.00
CH4 Factor	0.10
NO2 Factor	0.10
CH4 Factor	0.10

Pollutant	Per Engine	Total
NOx	1.14	4.58
CO	2.87	11.48
VOC	0.11	0.43
PM	0.57	2.31
SO2	0.29	1.16
CH4	0.11	0.43
NO2	0.11	0.43
CO2	100.20	396.80

Notes:
 1) All engine emissions will be subjected to the Regenerative Thermal Oxidizer (RTO) prior to venting to atmosphere.
 2) Engines will run on pyrolysis gas only.
 3) NOx, CO and VOC factors are from 40 CFR 60, Subpart JJJJ, Table 1 and adjusted for engine using a ratio.
 4) PM and SO2 factors are from AP-42, Table 2.2, 8th Edition, July, 2006 and adjusted for engine.
 5) Emission Factor for CO₂ is from 40 CFR 60, Table C-1 - Default CO₂ Emission Factors.
 6) Emission Factor for CH₄ and CO₂ are from 40 CFR 60, Table C-2, Default CH₄ and N₂O Emission Factors.
 7) NO₂ Emission, lbs/hr = Emission Factor, lbs/MWh_{th} x Fuel Consumption, lbs/hr / 1000 (MWh_{th}/hr) / 1000 (MWh_{th}/hr) / 1000 (MWh_{th}/hr).
 8) CH₄ Emission, lbs/hr = Emission Factor, lbs/MWh_{th} x Fuel Consumption, lbs/hr / 1000 (MWh_{th}/hr) / 1000 (MWh_{th}/hr).
 9) SO₂ Emission, lbs/hr = Emission Factor, lbs/MWh_{th} x Fuel Consumption, lbs/hr / 1000 (MWh_{th}/hr) / 1000 (MWh_{th}/hr).
 10) Fuel Consumption based on estimates provided by Technorm. See Default Emissions Calculations.
 11) PM = PM₁₀ and PM_{2.5}.
 12) CO₂ calculated using the following: $CR_{CO_2} = (1 - (CF_{fuel} \times 25)) \times (CR_{fuel} \times 284)$

**Florida Recycling Pyrolysis Facility
 Generator Emissions Calculations**

Parameter	Value
Number of Engines	4
Control Date	08/20/23
Gross Calorific Value	13,174.00 Btu/lb
Annual Amount of Gas Produced	1,000,000 lbs
Amount of Gas Produced per Hour	114.63 lbs/hr
Amount of Gas in Each Engine	28.66 lbs/hr
Fuel Consumption	1,000,000 lbs
Density	0.0044 lbs/lb
Engine Rating	250 HP
Rated Power	425 kW
Annual Operating Hours	7,000 hours
Fuel Heat Value	13,174 Btu/lb
NOx Factor	1.00
CO Factor	1.00
VOC Factor	0.10
PM10 Factor	0.05
PM2.5 Factor	0.05
SO2 Factor	0.10
CO2 Factor	1.00
CH4 Factor	0.10
NO2 Factor	0.10
CH4 Factor	0.10

Pollutant	Per Engine	Total
NOx	1.14	4.58
CO	2.87	11.48
VOC	0.11	0.43
PM	0.57	2.31
SO2	0.29	1.16
CH4	0.11	0.43
NO2	0.11	0.43
CO2	100.20	396.80

A. Maximum Temperature °C		F. Maximum Pressure PSI, Max. Set Pressure for setting	
°C		mmHg	
°C		PSIG	
B. Output Data	Flow Out + Material Name and CAS No.	Phase	Quantity Mass
			kg/hr or gal/hr
		Temperature	°C
		Pressure	PSI
		Hourly or Batch Output Rate	kg/hr or gal/hr
		Rate	kg/hr or gal/hr
		Other	

9. Complete the following emission data for equipment connected to a heater exhaust system, giving emissions when **PROCESS** entering heater system (i.e. before control equipment).
 Check here if not applicable
 Emission Point ID (omit last part of heater system):
 Material Name and CAS No. Maximum Potential Emission Rate (kg/hr) Notes **

** 100 - Material balance; EE - Engineering Estimate; T1 - Test (measurement) (batch) test data; O - Other (Equipment)

10. Provide the following information pertaining to each compressor that may be attached to this reactor. Attach additional pages as necessary if more than one compressor is used for this reactor. Complete the Compressor Air Pollution Control Device Sheet if necessary.
 Check here if not applicable

10A. Cooling medium:
 10B. Minimum and Maximum Inlet rate of cooling medium (gph):
 10C. Inlet temperature of cooling medium (°F):
 10D. Outlet temperature of cooling medium (°F):
 10E. Pressure drop of gas to be compressed from inlet to outlet (psig):
 10F. Inlet temperature of gas stream (°F):
 10G. Outlet temperature of gas stream (°F):
 10H. Number of passes
 10I. Cooling surface area

11. Provide the following pertaining to auxiliary equipment that burns fuel (heaters, dryers, etc.):
 Check here if not applicable

11A. Type of fuel and maximum fuel burn rate, per hour:
 11B. Provide maximum percent sulfur (S), ash content of fuel, and the energy content using appropriate units:
 % S % Ash BTU/lb. or 1000 gal (include unit)
 11C. Theoretical combustion air requirement in SCFD per unit of fuel (include appropriate units @ 70°F and 14.7 PSIA):
 SCFD lb. SCFD, gal (include unit)
 11D. Percent excess air: %
 11E. Type, amount, and BTU rating of burners and all other firing equipment that are planned to be used:
 11F. Total maximum design heat input: $\times 10^6$ BTU/hr.

12. Process Monitoring, Recordkeeping, Reporting, and Testing
 Please provide monitoring, recordkeeping, and reporting in order to demonstrate compliance with the proposed operating parameters. Please provide details in order to demonstrate compliance with the proposed emission limit.

MONITORING	RECORDKEEPING
REPORTING	TESTING

MONITORING: PLEASE LIST AND DESCRIBE THE PROCESS PARAMETERS AND RANGES THAT ARE PROPOSED TO BE MONITORED IN ORDER TO DEMONSTRATE COMPLIANCE WITH THE OPERATIONS OF THE PROCESS EQUIPMENT OR POLLUTION CONTROL DEVICE.
RECORDKEEPING: PLEASE DESCRIBE THE PROPOSED RECORDKEEPING THAT WILL ACCORD WITH THE MONITORING.
REPORTING: PLEASE DESCRIBE THE PROPOSED FREQUENCY OF REPORTING OF THE RECORDKEEPING.
TESTING: PLEASE DESCRIBE ANY PROPOSED TESTING TESTS FOR THE PROCESS EQUIPMENT OR AIR POLLUTION CONTROL DEVICE.

13. Describe all operating ranges and maintenance procedures required by manufacturer to maintain capacity.

NOTE: An AIR POLLUTION CONTROL DEVICE SHEET must be completed for any air pollution device(s) (except emergency relief devices) used to control emissions from this reactor.

DISTILLATION COLUMN DATA SHEET

Identification Number (as assigned on Equipment List Form):

1. Name and size of equipment

2. Proposed water use (normal operating schedule) (complete appropriate item)

recycle	recycle	recycle/lean
no/flush	backwash/ backwash/lean	recycle/lean/flush

3. Number of stages (packed, including condenser)

4. Number of feed points and stage location

5. Specify details of any heating, cooling, or stage conditioning along with the stage locations

6. Specify reflux ratio (where A is defined as the ratio of the reflux to the overhead product, given hydrocarbon as DMSO, water, or liquid stream column, or a distillation product)

7. Specify the feed(s) of feed which is recycled (where 1 is the mole fraction of the feed that recycles the feed (give continuously as report)

7A. Type of condenser used: None partial full other

7B. For each condenser provide process operating details including all inlet and outlet temperatures, pressures, and compositions

8. Feed Characteristics

A. Vapor composition	B. Inlet liquid pressure of each component
C. Total feed stage pressure	D. Total feed stage temperature
E. Feed mass flow rate of each stream into the system	

9. Overhead Product

A. Vapor composition of components	B. Feed composition of components
C. Total mass flow rate of all streams leaving the system as overhead product	

10. Bottom Product

A. Vapor composition of all components	B. Total mass flow rate of all streams leaving the system as bottom product
--	---

11. General Information

A. Distillation column diameter	B. Distillation column height
C. Type of column	D. Plate spacing
E. Theoretical plate efficiency	F. Any other information necessary to describe the operation of the distillation column

12. Proposed Monitoring, Recordkeeping, Reporting, and Testing

Please process monitoring, recordkeeping, and reporting in order to demonstrate compliance with the proposed operating parameters. Please process monitoring in order to demonstrate compliance with the proposed emissions limits.

MONITORING	RECORDKEEPING
REPORTING	TESTING

MONITORING: PLEASE LIST AND DESCRIBE THE PROCESS PARAMETERS AND RANGES THAT YOU PROPOSED TO BE MONITORED IN ORDER TO DEMONSTRATE COMPLIANCE WITH THE PROVISIONS OF THE PROCESS EQUIPMENT PERFORMANCE AIR POLLUTION CONTROL DEVICE.

RECORDKEEPING: PLEASE DESCRIBE THE PROPOSED RECORDKEEPING THAT WILL ACCOMPLISH THE MONITORING, REPORTING, AND TESTING. PLEASE DESCRIBE THE PROPOSED FREQUENCY OF REPORTING OF THE RECORDKEEPING.

TESTING: PLEASE DESCRIBE THE PROPOSED FREQUENCY OF TESTING FOR THIS PROCESS EQUIPMENT OR AIR POLLUTION CONTROL DEVICE.

13. Describe all operating ranges and maintenance procedures required by manufacturer to maintain efficiency.

ATTACHMENT O

Monitoring/Recordkeeping/Reporting/Testing Plans

ATTACHMENT O - MONITORING, RECORDING, REPORTING, and TESTING PLANS

Plan Type	Emission Unit	Pollutant	Requirements	Frequency	Method of Measurement	Regulatory Reference
Recordkeeping	Regenerative Thermal Oxidizer (RTO)	PM ₁₀ , PM _{2.5} , VOC and inorganic particulate	Control of visible particulate emissions Temperature monitoring	Daily Continuous	Visual	40 CFR 63, Subpart HHHH 40 CFR 60.12
Recordkeeping	Furnaces (BOS)	Single	Operate and maintain the source in a manner consistent with safety and good air pollution control practices to minimize emissions. Monitor for temperature.	Continuous during operations Daily and weekly maintenance	Operate the control equipment in accordance with manufacturer's instructions (demonstrate)	N/A
Monitoring/Recordkeeping	Smelters (AS)	PM ₁₀ , PM _{2.5} , sulfate	Pressure Drop monitoring	Annual	The pressure drop across the system, not production rate Detailed Management Plan will monitor Detailed Activity by adding separate bins and sending them to the laboratory for activity testing.	N/A

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ATTACHMENT P

Public Notice

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AIR QUALITY PERMIT NOTICE
Notice of Application

Notice is given that Empire Green Generation, LLC has applied to the West Virginia Department of Environmental Protection, Division of Air Quality, for a modification of Permit #13-2650 for a Phoslock Recycling Plant located on 801 Kappeler Road, near Polarisville, in Boone County, West Virginia. The address and longitude coordinates are: 40 22820 N, -80 62543 W.

The applicant estimates the source to discharge the following Regulated Air Pollutants will be less than 24.0 tons per year (tpy) of VOCs, 14.0 tpy of PM₁₀, 8.0 tpy of CO, 24.0 tpy of NO_x, 2.0 tpy of HAPs and 28.0 tpy of SO₂.

Operation is planned to begin on or about the 15th day of July, 2022. Written comments will be received by the West Virginia Department of Environmental Protection, Division of Air Quality, 401 E. 9th Street, SE, Cheyenne, WY, 82002, or at least 30 calendar days from the date of publication of this notice. Written comments will also be received via email at DEP@AirQualityPermitting@wv.gov.

Any questions regarding this permit application should be directed to the DAQ at (304) 526-0495, ext. 41231, during normal business hours.

Dated this the (24th) day of (Month), (2022).
By: Empire Green Generation, LLC
Bryan Brown
Chief Technology Officer
1400 Main Street
Polarisville, WV 26027

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STATEMENT OF PUBLICATION

State of North Carolina, County of Orange, etc.

Notice is given that Orange Green Generation, LLC, has applied to the West Virginia Department of Environmental Protection, Division of Air Quality for a Modification of Permit #13-3855 for a Thermal Generating Plant located on 801 Rogers Road, near Charleston, in Boone County, West Virginia. The notice and hearing date information are: 03/20/2018, 10:00 AM EST.

PUBLICATION DATES:

On: 8/2/2018

ISSUED BY: COURTNEY BUCHHEIT/WRVTV

PUBLISHER: G. L. COLE

NOTICE NAME: Green Gen Public Notice Modification

Publication Fee: \$2.15

Rade Hill

VERIFICATION

State of North Carolina, County of Orange

Submitted in my presence and seen to be true on this 12/17/2018

Notary Public

Notary Seal Online using audio/visual communication

AIR QUALITY PERMIT NOTICE
Notice of Application

Notice is given that Orange Green Generation, LLC, has applied to the West Virginia Department of Environmental Protection, Division of Air Quality for a Modification of Permit #13-3855 for a Thermal Generating Plant located on 801 Rogers Road, near Charleston, in Boone County, West Virginia. The notice and hearing date information are: 03/20/2018, 10:00 AM EST.

The applicant certifies the proposed modification to the permit is not a major modification as defined in the permit. The modification is limited to the following: 1) The permit is being modified to allow for the use of a 2000 BTU/hr gas boiler to provide heat for the plant. 2) The permit is being modified to allow for the use of a 2000 BTU/hr gas boiler to provide heat for the plant.

Notice of application is given to all persons who are known to be affected by the proposed modification. Any questions regarding the permit application should be directed to the DAP at 2044 5th Street, Charleston, WV 25301. Orange Green Generation, LLC, 801 Rogers Road, Charleston, WV 25301. Chief Technology Officer, 1000 Main Street, Huntington, WV 25701. 800.746.6888

ATTACHMENT Q

Business Confidential Claims

November 16, 2018



Mr. Edwards Andrews, P.E.
West Virginia Department of Environmental Protection
Division of Air Quality
801 57th Street, SE
Charleston, WV 25304

RE: Empire Green Generation Confidential Business Information

Dear Mr. Andrews,

Please find enclosed Empire Green Generation's submital of Confidential Business Information for the modification application for permit number #13-3855. This claim of confidentiality is due to trade secrets and intellectual property. A redacted version of this submital has been sent to you via email.

Sincerely,

Bernard S. Brown

Bernard Brown
Chief Operating Officer

Enc.

1400 Main Street, Putnamville, WV 26027
204-274-6824

Precautionary Notice — Claims of Confidentiality

The person submitting this information may assert that some or all of the information submitted is entitled to confidential treatment as provided by West Virginia Legislative Rule 41-28.1, entitled "Confidential Information." Information covered by such a claim will be disclosed to the Division of Air Quality (DAQ) only to the extent, and by means of the procedures set forth in 45CSR31. Please contact the West Virginia Secretary of State's Office at 744-555-6000 or <http://www.sos.wv.gov/office/daq> to obtain a copy of 45CSR31 in order to ensure that all required procedures are followed.

Information concerning the "types and amounts of air pollutants discharged," as that term is defined in WVCSR 445-31-2.4, shall not be claimed as confidential.

Any claim of confidentiality shall be made in accordance with the requirements of 45CSR31 and must accompany the information at the time it is submitted to the DAQ. If a claim of confidentiality is made at the time of submission or is not made in accordance with the requirements of 45CSR31, the DAQ may make the information available to the public without further notice.

Included below are procedures, and an example form, to be followed in submitting information claimed as confidential. This information is intended to assist a person with claiming confidential information and is not meant to reflect a person's fiduciary obligation to review the provisions of 45CSR31 and to comply with such rule. The procedures are as follows:

1. Indicate clearly the items of information claimed confidential by marking each page with the term "Claimed Confidential," with the date of such claim of confidentiality. With the exception of documents of a size greater than 8 1/2" x 11", information claimed confidential must be submitted on colored paper.
2. Include a cover document (See below) which justifies the claim of confidentiality in accordance with the specific criteria under WVCSR 445-31-4.1. A sample cover document is attached for your information and use. The cover document will be available for public disclosure and must include the following information:
 - (a) The identity of the person making the submission of information claimed confidential;
 - (b) The reason for the submission of information;
 - (c) The name, an address in the State of West Virginia and telephone number of the designee who shall be contacted in accordance with 45CSR31;
 - (d) Identification of each segment of information within each page that is submitted as confidential and the justification for each segment claimed confidential, including the criteria under WVCSR 445-31-4.1.

Precautionary Notice — Claims of Confidentiality
WVDEP DAQ Revised March 21, 2018

- (e) The period of time for which confidential treatment is desired (e.g., until a certain date, until the occurrence of a specified event or permanently, and,
 - (f) Signature of a responsible official or an authorized representative of such person.
3. At the same time as the information claimed confidential is submitted to the DAQ on colored paper, a complete set of the information, including the cover document previously required under paragraph 2, must be submitted with the information **claimed to be confidential bladed or white out the words "Redacted Copy — Claim of Confidentiality" marked clearly on each such page, so that the information is suitable for public disclosure. In the case of drawings and blueprints, mark each such page with the words "Redacted Copy — Claim of Confidentiality," include the title or legend of the drawing, and black or white out the information claimed confidential. The redacted page may be 8 1/2" x 11" in size.**

4. In the case of a permit application or supplemental information to an application which contains confidential information, DAQ requires the "Redacted Copy — Claim of Confidentiality," pages and the cover document which justifies the claim of confidentiality to be submitted by e-mail as a PDF file to: DEP@confidentiality@dmr.wv.gov

See instructions at: <http://dmr.wv.gov/office/daq/permitting/Permitting-forms.aspx>, OR <http://daq.wv.gov/daq/permitting/PermittingForms.aspx>

5. "Claimed Confidential" pages may not be e-mailed and shall be submitted, as hardcopy, on colored paper and mailed to:

WVDEP — DAQ — Permitting
Attn: NSR or Title V Permitting Secretary *
601 3rd Street SE
Charleston, WV 25304

* For a 45CSR31 application, send to NSR Permitting Secretary. For a 45CSR30 application, send to Title V Permitting Secretary. If this is a combined NSR/Title V Permit Application, send one copy to the NSR Permitting Secretary and one copy to the Title V Permitting Secretary.

Precautionary Notice — Claims of Confidentiality
WVDEP DAQ Revised March 21, 2018

**Sample Cover Document
Confidential Information**

This sample form contains each of the required elements for the cover document required under 45CSR31. The person submitting this form may wish to attach an additional page(s) to provide adequate justification under the "Rationale" section of the form.

Company Name	Empire Clean Coatings	Responsible Official	Edward Brown
Company Address	1403 Main Street Fayetteville, WV	Name	
		Title	
		Designee in State of WV	
Person/Title Submitting Confidential Information		Address	
		Phone	
		Fax	

Reason for Submission of Confidential Information:

Identification of Confidential Information	Rationale for Confidential Claim	Confidential Treatment Time Period
	Provide justification that the criteria set forth in § 45CSR31-4.1.a - c have been met.	

Responsible Official Signature	<i>Edward A. Brown</i>
Responsible Official Title	Chief Operating Officer
Date Signed	1/10/2018

NOTE: Must be signed and dated in BLUE INK

Precautionary Notice — Claims of Confidentiality
WVDEP DAQ Revised March 21, 2018

ATTACHMENT R

Authority Form

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AUTHORITY OF LIMITED LIABILITY COMPANY (LLC)

TO: The West Virginia Department of Environmental Protection, Division of Air Quality
DATE: July 8, 2022
ATTN: Director
LLC's Federal Employer I.D. Number: 87-0147036

The undersigned hereby files with the West Virginia Department of Environmental Protection, Division of Air Quality, a permit application and hereby certifies that the said name is a trade name which we are using in the conduct of an unincorporated business.

Further, we have agreed or certified as follows:

- (1) The undersigned is a member and in that capacity may represent the interests of the LLC and may obligate and legally bind at current or future meetings and the LLC.
- (2) The LLC is authorized to do business in the State of West Virginia.
- (3) The name and business address of each member:

Member: Frank Russo
Address: 451 East on Ohio West, Suite 1400, East Leitchfield, FL 33001
Telephone No.: 304-330-2423

Member: _____
Address: _____
Telephone No.: _____

Member: _____
Address: _____
Telephone No.: _____

- (4) If any other persons become members of the undersigned or our relations as such be altered in any way or if the business should become incorporated, the undersigned will notify you promptly.


Address: 451 East on Ohio West, Suite 1400
East Leitchfield, FL 33001
Telephone No.: 304-330-2423

MEMBER OF LLC (Signature)
Frank J Russo
MEMBER OF LLC (Typed)

LIMITED LIABILITY COMPANY'S NAME

WV PERM 10010-0000
REV 07/10/02 (REVISED 7/01)

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ATTACHMENT S

Title V Permit Revision Information
(Not Applicable)

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Inc App Email 12/20/2024

Tuesday, February 6, 2024 1:30 PM



Andrews, Edward S <edward.s.andrews@wv.gov>

Incomplete App Email for Permit App R13-3555A

1 message

Andrews, Edward S <edward.s.andrews@wv.gov>

Wed, Dec 20, 2023 at 2:33 PM

To: Bernard Brown <bbrown@empirede.com>, Farley Wood <fwood@empirede.com>, "Pugh, Katie" <Katie.Pugh@tetrattech.com>

Cc: Beverly D Mckeone <beverly.d.mckeone@wv.gov>, Brian S Tephabock <Brian.S.Tephabock@wv.gov>, Eric Blend <eric.n.blend@wv.gov>

**RE: Application Status: Incomplete
Empire Green Generation
Permit Application No. R13-3555A
Plant ID No. 009-00141**

Mr. Brown:

Your application for a modification permit for a plastic recycling by pyrolysis facility was received by this Division on December 1, 2023, and assigned to the writer for review. Upon initial review of said application, it has been determined that the application as submitted is incomplete based on the following items:

1. Affidavit of Publication of Class I Legal Ad.
2. Discussion of the proposed physical change(s), if applicable, and/or change in the method of operation. Specifically, this discussion needs to either outline or go into detail regarding proposed/suggested changes to the permit, if applicable; type and source(s)/origin of the plastic feedstock going to be process by the facility; any processing/pretearting/sorting going to be conduct on the plastic prior to being introduce into the pyrolysis unit(s), if these preprocessing is going to occur of off-site, it still needs to be identified and discussed; a discussion how the facility will switch back and forth in processing medical waste and plastic feedstock; and a discussion why processing the proposed plastic feedstock is not viewed as waste disposal through incineration in the content of the Clean Air Act and meet the criteria as a fuel(s) within the requirements and procedures of 40 CFR 241 - SOLID WASTES USED AS FUELS OR INGREDIENTS IN COMBUSTION UNITS.
3. Plot plan (Attachment E) needs to be updated to identify emission units and emission points.
4. Attachment J needs to be complete for each emission point.
5. Attachments K and L need to be completed. The potential for leaking equipement (e.g., valves, pumps, compressors, connectors, pressure relief devices) needs to be quantified and documented in these two attachments.
6. Each of the redacted pages that contain confidential business information (CBI) needs to be remarks "redacted copy - claim of confidentiality" in accordance with 45CSR31-3.4.

In addressing issues 2 through 6 needs to be reflective within the redacted application as a single PDF file.

The emissions estimates appear to be identical to the emission estimates for processing medical waste with the same pyrolysis unit. Please review these estimates and revise as necessary and/or justify in detail why the emission will not change given the change in feedstocks.

Please address the above deficiencies in writing by no later than January 16, 2024. Application review will not commence until the application has been deemed to be technically complete. Failure to respond to this request in a timely manner may result in the denial of the application.

Should you have any questions, please contact Ed Andrews at (304) 926-0499 ext. 41244 or reply to this email.

--

Edward Andrews, P.E.
Engineer
WVDEP/Division of Air Quality
304-926-0499 Ext 41244
601 57th Street, SE
Charleston, WV 20304

Parcel Notice 12-5-2023

Wednesday, December 6, 2023 8:03 AM



Andrews, Edward S <edward.s.andrews@wv.gov>

You received a parcel! Please come pick it up

1 message

ilobby@ilobbycloud.com <ilobby@ilobbycloud.com>
To: edward.s.andrews@wv.gov

Tue, Dec 5, 2023 at 10:26 AM



Parcel Pending Pick Up

Shipping Label

Date Received

Tracking #

Site Name



Dec 05, 2023

0077429849

WV Department of
Environmental
Protection



Note: AIR

Mark as picked up

App Submittal 12/1/2023

Wednesday, December 6, 2023 8:00 AM



Andrews, Edward S <edward.s.andrews@wv.gov>

Empire Green Generation Permit Modification Application

1 message

Wood, Katie <Katie.Pugh@tetrattech.com>
To: Edward Andrews <edward.s.andrews@wv.gov>
Cc: Farley R Wood <fwood@empirede.com>

Fri, Dec 1, 2023 at 4:09 PM

Ed,


Please find attached Empire Green Generations (EGG) permit modification application attached. EGG is making a claim for confidentiality so the attached file is redacted and the pages that are confidential will arrive via FedEx next week. Please feel free to reach out to me with any questions or concerns.

Thank you,

Katie Wood* | Environmental Scientist
Direct **+1 (740) 298-9062** | Mobile **+1 (304) 559-9980** | katie.wood@tetrattech.com
Formerly Katie Pugh, please note name change
Tetra Tech | *Leading with Science®* | OGA
47443 National Rd Suite 3 | St. Clairesville, OH 47443 | tetrattech.com

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 **R13-3555 Modification_Application_EGG - Redacted_Final 12-1-23.pdf**
21661K